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magali.duvail@cea.fr

Microemulsions: a new model for organic phases involved in ion separation methods

Magali Duvail, Jean-François Dufrêche and Thomas Zemb

Institut de Chimie Séparative de Marcoule (ICSM), UMR 5257, CEA – CNRS – UM2 – ENSCM Site de Marcoule, Bât. 426, BP 17171, F-30207 Bagnols-sur-Cèze Cedex, France

Introduction

Cea





Context

- Improvement of the reprocessing of spent nuclear fuel \checkmark
- Liquid-liquid extraction equilibria
- **Microemulsion** model to describe the **organic phase**
- Multi-scale approach



Method

Results

- ✓ The **physics** of a system composed of **soft interfaces** (low bending energy compared to the thermal one, negligible long-range electrostatic and steric interactions) is **dominated** by the Helfrich Hamiltonian of the surfactant interface (between oil and water) [1].
- **Rigidity** controlled by κ and $\bar{\kappa}$ ✓ Two constants:

Curvatures:



✓ In our model, the surfactant interface is described by appropriate two level-cuts of a Gaussian random field based on wavelets proposed by Arleth et al. [2].

 $f_{\rm free} = \frac{S}{V} \left[2\kappa \langle (H - H_0)^2 \rangle + \bar{\kappa} \langle K \rangle \right] - \frac{1}{2\pi^2} \int_{\mu} \mathrm{d}k \ k^2 \ln \nu(k)$

- The free energy density of the Gaussian random field with level-cuts can be approximated by:
- \rightarrow the bending elastic constant κ
- \rightarrow the Gaussian elastic constant $\overline{\kappa}$
- ✓ One unique constant:
 - $\kappa^* = 2\kappa + \bar{\kappa}$
 - \rightarrow Analogy with the packing parameter p₀

$$f_{\rm free} = \frac{1}{2} \boldsymbol{\kappa}^* (p - \boldsymbol{p_0})^2$$



microemulsion ($\kappa = 2 k_B T$ and $\overline{\kappa} = 0 k_B T$) and $p_0 = 1$ ($H_0 = 0$).

Spinodal instablities calculations (case of rigid surfactant film) \checkmark

Thermalized unfrustrated, frustrated and bicontinuous microemulsions

of surfactant $\Phi_s = 0.15$.

✓ **Frustration** appears only in the case of **rigid microemulsions**

✓ Unfrustrated bicontinuous microemulsions for $0.2 < \Phi_{apolar} < 0.8$



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References

[1] W. Helfrich, Z. Naturforsch. C 28, 693 (1973) [2] L. Arleth et al., J. Chem. Phys. 115, 3923 (2001) [3] J. Jouffroy *et al.*, *J. Phys.* **43**, 1241 (1982)

Conclusions and outlooks



Improvement of our model to take into account:

✓ the presence of charged species (ions and ionic surfactant), ✓ the **aggregation** of surfactant molecules.

