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1 **Assessing the impact of granular anaerobic membrane bioreactor**
2 **intensification on treatment performance, membrane fouling and**
3 **economic balance**

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13 **ABSTRACT**

14 Anaerobic membrane bioreactors (AnMBRs) have attracted much attention for mainstream
15 domestic wastewater treatment. However, membrane fouling, operating costs, energy
16 consumption and low filtration flux are important challenges slowing the scale-up of the
17 technology. In this study and for the first time, granular sludge, submerged membrane, no gas
18 sparging and low permeate flux were chosen to mitigate membrane fouling and to improve the
19 energy and economic balance of an AnMBR. A granule-based AnMBR (G-AnMBR) was
20 operated under four organic loading rates (between 0.5 and 1.6 kgCOD.m⁻³.d⁻¹) with hydraulic
21 retention times ranged from 13.9 to 4.9 hours, and instantaneous permeate flux levels ($J_{p20,inst}$)
22 ranged from 2.8 to 6.0 LMH to evaluate OLR impact on anaerobic digestion performance,
23 membrane fouling extent and economic balance. Results show that COD removal rates above

24 83% were achieved during the four experimental periods. Membrane fouling was directly
25 correlated to the flux and OLR and increased from 0.03 to 2.86 kPa.d⁻¹ as the $Jp_{20,inst}$ increased
26 from 2.8 to 6.0 LMH and the OLR increased from 0.5 to 1.6 kgCOD.m⁻³.d⁻¹, respectively. In
27 all the periods, macromolecules and colloidal proteins were the major foulants deposited on the
28 membrane. Most of the fouling was reversible and was easily removed by physical cleaning
29 (>97.7%). A preliminary economic assessment revealed that the permeate flux and OLR are
30 key economic drivers for the G-AnMBR economic balance and allowed to define the
31 satisfactory compromise between membrane purchase and chemical consumption for the long-
32 term control of membrane fouling.

33 **KEYWORDS**

34 filtration flux; membrane fouling; techno-economic evaluation; granular sludge; anaerobic
35 digestion

36 **1. Introduction**

37 Anaerobic membrane bioreactors (AnMBRs) have gained interest over conventional aerobic
38 biotechnologies by combining the advantages of anaerobic processes with membrane
39 technology. Anaerobic digestion reduces energy demand, produces fewer bio-solids, reduces
40 sludge disposal costs and generates bioenergy through the complete conversion of organic
41 materials into methane (Vinardell et al., 2020). Micro- or ultrafiltration membranes offer strong
42 retention of particles and microorganisms, leading to higher biomass concentration and longer
43 sludge retention time (SRT), which is beneficial for slow-growth anaerobic microorganisms
44 and the quality of the permeate. Therefore, the AnMBR is a cost-effective alternative that allows
45 conversion of wastewater organic material into renewable methane energy while achieving a
46 high-quality effluent, free of suspended solids and pathogens, which could be easily reused for
47 various applications (Aslam et al., 2022; Vinardell et al., 2020).

48 Nonetheless, the AnMBR still faces some issues that have hindered its large-scale development
49 for domestic wastewater treatment. The main operational and technical challenges in its
50 application to domestic wastewater have been linked to low psychrophilic temperatures, low-
51 strength wastewater, dissolved methane and membrane fouling, among others (Aslam et al.,
52 2022; Lei et al., 2018; Vinardell et al., 2020). The large amount of dilute domestic wastewater
53 results in lower methane conversion rates ($L\text{-CH}_4\text{.m}^{-3}$) and higher quantities of methane lost
54 within the effluent (Maaz et al., 2019). The loss of methane in the liquid phase diminishes the
55 potential for energy recovery and produces direct greenhouse gas emissions (Smith et al., 2012).
56 Membrane fouling has a high impact on the capital and operational costs of AnMBR due to its
57 effect on permeate flux, membrane lifespan and energy demand (Maaz et al., 2019; Smith et
58 al., 2012). Accordingly, all these factors decrease energy recovery efficiency and increase
59 energy demand (e.g., membrane fouling mitigation, dissolved methane recovery, etc.), which
60 reduce the economic and energy advantages of AnMBRs.

61 In recent years, several studies have applied lab- and pilot-scale AnMBRs to low-strength
62 wastewater treatment at psychrophilic temperatures. As described in Table 1, various
63 configurations and fouling control strategies have been tested to achieve high efficiencies of
64 organic matter removal and methane conversion while mitigating membrane fouling. The
65 chemical oxygen demand (COD) removal efficiencies obtained through AnMBR processes
66 have typically ranged between 85% and 95% at both lab- and pilot-scales under sub-optimal
67 conditions (i.e., ambient temperature and low-strength wastewater) (Table 1), suggesting that
68 efficient methane yield and conversion rates can be reached (Chen et al., 2017b; Nie et al.,
69 2017; Shin et al., 2021). Then, many investigations have focused on AnMBR configuration and
70 membrane fouling (Table 1). The different system configurations have mainly differed in terms
71 of (i) integration of the membrane (i.e., submerged, external submerged, side-stream), (ii) type
72 of anaerobic bioreactor (i.e., continuous stirred tank reactor [CSTR], upflow anaerobic sludge

73 blanket [UASB], etc.), (iii) membrane module type (i.e., flat sheet, hollow fiber, etc.), (iv)
74 fouling control strategies and (v) operating conditions. Innovative strategies have been
75 implemented to better control membrane fouling (Aslam et al., 2017), including rotating
76 membrane (Ruigómez et al., 2016a, 2016b), dynamic membrane (Hu et al., 2018; Quek et al.,
77 2017; Yang et al., 2020), granular activated carbon media (Evans et al., 2019), sponge media
78 (Chen et al., 2017b) and granular biomass (Chen et al., 2017a; Gouveia et al., 2015). However,
79 gas sparging remains the most frequently used approach for long-term mitigation of membrane
80 fouling despite its energy consumption, which amounts to up to 70% of the total AnMBR
81 energy input (Batstone and Viridis, 2014; Smith et al., 2014).

82 Among the innovative strategies that have been tested, the use of granular biomass is an
83 attractive cost-effective solution, as it reduces energy consumption for the control of membrane
84 fouling (Martin-Garcia et al., 2013) and does not involve additional material costs or technical
85 complexity. Granulation is a process resulting in dense granule-shaped biomass which
86 improves biological activity and strength (van Lier et al., 2008). Unlike the suspended sludge
87 configuration, a granule-based AnMBR (G-AnMBR) helps to lower the concentration of
88 suspended solids in contact with the membrane, improving the control of membrane fouling
89 and decreasing cake-layer formation (Vinardell et al., 2022). However, the G-AnMBR
90 configuration does not fully overcome membrane fouling, as complete membrane retention
91 leads to the accumulation of colloidal and fine particulate matter, either introduced from the
92 influent or released from the granular sludge bed due to too strong hydrodynamic conditions
93 (Anjum et al., 2021; Vinardell et al., 2022). To mitigate membrane fouling, gas sparging is also
94 used in this granule-based configuration. Kong et al. (2021b) reached the maximal TMP (235
95 mbar) in 8 days at a filtration flux of $17.8 \text{ L}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ (LMH) and at a specific gas demand (SGD)
96 of $0.75 \text{ m}^3\cdot\text{m}^{-2}\cdot\text{h}^{-1}$. Wang et al. (2018) applied a SGD of $0.2 \text{ m}^3\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ and achieved the
97 maximal TMP (550 mbar) in less than one day when the flux was increased from 5 to 10 LMH.

98 Gas sparging is cost-intensive and damages granule structures which increases granules fines
99 and dissolved and colloidal organic matter (DCOM) release in the bulk (Martin-Garcia et al.,
100 2013). For these reasons, avoiding the use of gas sparging may assist in maintaining the
101 integrity and the particle size distribution of granules, with a substantial benefit in energy
102 consumption (Sanchez et al., 2022). In addition, it appears that a high permeate flux cannot be
103 maintained, which reduces the competitiveness of G-AnMBR compared to conventional
104 processes (AeMBR). Hence, it is necessary to explore the behavior of the process following the
105 increase of permeate flux and, more specifically, to characterize membrane fouling and its
106 economic and energetic implications during process intensification.

107 This study aimed to evaluate the impact of G-AnMBR process intensification – by increasing
108 the organic loading rate and the permeate flux – on anaerobic digestion performance, membrane
109 fouling and process economics. The G-AnMBR system was operated without gas sparging to
110 limit the energy consumption during domestic wastewater treatment and ambient temperature.
111 To the best of our knowledge, the impact of process intensification on membrane fouling of a
112 G-AnMBR in submerged mode and operated without gas sparging, has not yet been evaluated.
113 For this reason, a G-AnMBR without gas sparging for fouling control was operated under four
114 permeate flux conditions of 2.8, 3.6, 4.1 and 6.0 LMH and OLR of 0.5, 0.8, 1.0, 1.6 kgCOD.m⁻³.d⁻¹.
115 Anaerobic digestion performance and membrane fouling behavior were assessed. Based
116 on the experimental results, an economic evaluation was conducted to elucidate how the
117 intensification of the process (through OLR and permeate flux increase) might influence the
118 operational and capital costs of a G-AnMBR operated under low filtration rates.

119

Table 1 – Reactor configurations, operational parameters, treatment performance and membrane fouling strategies presented in recent AnMBR studies for domestic wastewater treatment at ambient temperature.

Reactor	Membrane setup	Fouling control strategy	Scale	Type of WW	COD _{in}	T	HRT	OLR	COD _{removal}	Methane yield	Flux	dTMP/dt	Reference
–	–	–	(liters)	–	(mg.L ⁻¹)	(°C)	(h)	(kg.m ⁻³ .d ⁻¹)	(%)	(L-CH ₄ /gCOD _{removed})	(LMH)	(kPa.d ⁻¹)	
UAGB	HF Sub.	Sponge cubes + intermittent cycle	(L) 3	Synthetic	330–370	20	12	–	93.7	0.156 (G)	5.3	0.5	(C. Chen et al., 2017b)
UAGB	HF Sub.	Granular sludge + Intermittent cycle	(L) 4	Synthetic	320–360	20	12	0.64–0.72 *	91.3	0.16 (G)	7	0.9	(C. Chen et al., 2017a)
UAGB	FS Sub.	Granular sludge + Intermittent cycle	(L) 6.2	Synthetic	300–350	25	13.9–4.9	0.5–1.6	83–93	70–77% ^a (G + D)	2.8–6	0.03–2.9	This study
UASB	FS sub.	Dynamic membrane	(L) 6.9	Raw	107–137	25–30	3–6	1.5–3.0	64–71	0.354 (–)	100	1.5–6.2	(Quek et al., 2017)
UAGB	HF Ext. sub.	Granular sludge + Intermittent cycle + gas sparging	(P) 72.5	Raw	221	16.3	8	<1.0 *	83	–	5–15	0.2–960 *	(Wang et al., 2018)
UAGB	HF Sub.	Granular sludge + Intermittent cycle + gas sparging	(P) 459	Raw	978	18	9.8–20.3	0.72–3.18	≈90	0.216–0.226 (G + D)	10–14	0.1–1.9	(Gouveia et al., 2015)
CSTR	FS Sub.	Intermittent cycle + gas sparging	(L) 6	Synthetic	700	25	8–48	0.35–2.1	≈95	0.277–0.328 (G + D)	1.1–6.5	–	(R. Chen et al., 2017a)
CSTR	FS Sub.	Intermittent cycle + gas sparging	(L) 6	Synthetic	700	25	8–48	0.35–2.1	>90	0.305–0.338 (G + D)	1.1–6.5	0.05–1.1 *	(R. Chen et al., 2017b)
CSTR	– Sub.	Gas sparging	(L) 6	Synthetic	492	25	24–12	3–6	97–94	89–84.7% ^a (G + D)	–	0.08–0.3	(Nie et al., 2017)
CSTR	FS Sub.	Intermittent cycle + gas sparging	(L) 40	Raw	428–477	14–26	12–48	0.23–0.9	69–89	0.28–0.35 ^{*b} (G + D)	1.6–6.6	<0.06 *	(Plevri et al., 2021)
CSTR	HF Ext. sub.	Gas sparging	(P) 630	Synthetic	304–388	23	8.5	0.85–1.1 *	88–92	0.076–0.115 (G + D)	17	0.02–0.61	(Dong et al., 2016a, 2016b)

AFBR	HF Sub.	Granular activated carbon + gas sparging	(P) 990	Raw	210	13–32	3.9	1.3–1.4	86–90	0.17 (G)	7.6–7.9	–	(Evans et al., 2019)
AFBR	HF Ext. sub.	Intermittent cycle + gas sparging	(P) 4500	Raw	720–893	18–25	5.3–10	1.3–2.4	87–90	43–62 % ^a (G + D)	6.5–12.3	–	(Shin et al., 2021)
–	HF Sub.	Intermittent cycle + gas sparging	(P) 20	Raw	422	25	4–12	1.52–0.72	84–89	0.14–0.21 (G + D)	7.2–14.2	0.8–2.1	(Ji et al., 2021a)
–	HF Sub.	Gas sparging	(P) 20	Raw	422	15–25	6	0.15–0.18	77–90	0.06–0.17 (G + D)	9.5	0.1–13.0	(Ji et al., 2021b)
–	HF Sub.	Gas sparging	(P) 20	Raw	300–600	15	6–24	0.4–1.6	77–91	0.06–0.23 (G + D)	2.4–9.4	1.4–9.1	(Ji et al., 2022)
–	HF Sub.	Intermittent cycle + gas sparging	(P) 5000	Raw	403–461	25–27	6–24	0.37–1.84	90–93	0.16–0.26 (G + D)	4.4–17.8	0.08*–2.9	(Kong et al., 2021a) (Kong et al., 2021b)
–	HF Ext. sub.	Intermittent cycle + Gas sparging	(P) 40000	Raw	1235	27–30	24–60	–	92	0.21 (G + D)	15–23.5	0.04–0.22	(Robles et al., 2020)
–	HF Sub.	Intermittent cycle + Gas sparging	(P) 5000	Raw	414	25	8	1.2	90	0.222 (G + D)	10.85	0.04–0.2	(Rong et al., 2021)
–	HF Sub.	Rotating membrane	(L) 3	Raw	1462	19	33	–	91	0.154 * (G)	10	3.8*–14	(Ruigómez et al., 2016a, 2016b)
–	– Sub.	Gas sparging	(L) 6	Synthetic	491	10–25	6–48	0.25–1.0	71–98	32–77 % ^a (G)	–	0.17–0.72	(Watanabe et al., 2017)
UASB	– Sub.	Dynamic membrane	(L) 3.6	Raw	251–284	20–25	1–8	0.8–6.8	60–77	0.05–0.12 (G + D)	22.5–180	0.4–2.1	(Yang et al., 2020)

120 WW: wastewater; HRT: hydraulic retention time; OLR: organic loading rate; UAGB: upflow anaerobic granular bioreactor; UASB: upflow anaerobic sludge blanket; CSTR: completely stirred
121 reactor; AFBR: anaerobic fluidized bed reactor; FS: flat sheet; HF: hollow fiber; Sub.: membrane submerged in the bioreactor; Ext. sub.: membrane submerged in an external tank; L: lab-scale;
122 P: pilot-scale; G: gaseous methane; D: dissolved methane.

123 * Values were calculated or approximated through data available in the publication.

124 ^a Methane conversion rate (%)

125 ^b Values were calculated based on the gas yield provided, considering a concentration of CH₄ of 70%.

126 **2. Materials and methods**

127 **2.1 G-AnMBR setup**

128 The G-AnMBR system consisted of a parallelepipedic tank ($266 \times 68 \times 523$ mm) with a
129 working volume of 6.2 L that was operated in continuous mode for 357 days. The membrane
130 module was immersed in the middle of the liquor (at a height of 205 mm from the bottom). The
131 ultrafiltration membrane was a polyethersulfone (PES) flat-sheet membrane with a pore size of
132 $0.04 \mu\text{m}$ (Microdyn Nadir[®], Germany). The module was composed of six membrane sides,
133 providing a total surface area of 0.34 m^2 . The permeate was suctioned through a peristaltic
134 pump (LeadFluid[®], China) following an operation cycle of 8 min 15 s of filtration, 30 s of
135 relaxation, 45 s of backwash and 30 s of relaxation. No method for the mitigation of membrane
136 fouling was used apart from the intermittent cycle. The operation of the reactor was
137 automatically managed by automation software developed by AC2I Automation (France). The
138 reactor was thermoregulated at 25°C . The influent was fed at the bottom of the reactor by a
139 peristaltic pump and diffused over the entire length by a pierced hollow tube ($\varnothing 1$ mm) to
140 minimize dead zones. Continuous recirculation from the top to the bottom of the reactor was
141 used to provide an upflow liquid velocity (ULV) of 2.6 m/h (Boulenger and Gallouin, 2009)
142 and no other mixing that this recirculation was operated. A schematic representation of the lab-
143 scale reactor is shown in Fig. A1.

144 A complex synthetic influent was used to mimic domestic wastewater treatment, as described
145 in previous studies (Layer et al., 2019; Sanchez et al., 2022). The feed solution was prepared
146 every week with and stored at 4°C for one week. During the overall experiment, the average
147 composition of the influent was approximately $290\text{--}350 \text{ mgCOD.L}^{-1}$ for total COD (tCOD),
148 $250\text{--}290 \text{ mgCOD.L}^{-1}$ for soluble COD (sCOD), $40\text{--}60 \text{ mgCOD.L}^{-1}$ for particulate COD
149 (pCOD) and $160\text{--}210 \text{ mgCOD.L}^{-1}$ for volatile fatty acids (VFA).

150 The G-AnMBR was inoculated with granular sludge from a mesophilic (35–38°C) UASB that
151 treats wastewater from a recycled paper factory (Saica Paper Champblain-Laveyron, France) at
152 a high organic loading rate (18 kgCOD.m⁻³.d⁻¹). The granular sludge was acclimatized stepwise
153 to the low-strength (0.5 kgCOD.m⁻³.d⁻¹) and psychrophilic temperature (25°C) for 20 months
154 before the beginning of this study. The total solids concentration in the G-AnMBR was 70–80
155 gTS.L⁻¹. No sludge was purged during the 357 days of the experiment except for sampling.

156 **2.2 Experimental design and operating conditions**

157 Four experimental periods were used to assess the impact of design parameters on G-AnMBR
158 performance, membrane fouling behavior and economic assessment. The influent flow was
159 progressively increased, with a direct impact on hydraulic retention time (HRT), organic
160 loading rate (OLR) and permeate flux (J_{20}). The G-AnMBR was then operated according to the
161 following conditions: instantaneous filtration fluxes ($J_{20,inst}$) of 2.8, 3.6, 4.1 and 6.0 LMH, HRT
162 from 14 to 5 h, and OLR from 0.5 to 1.6 kgCOD.m⁻³.d⁻¹ (Table 2), which are within the
163 common range of AnMBR studies for domestic WW treatment (Vinardell et al., 2020). The end
164 of each period was reached when a steady efficiency of COD removal was achieved and the
165 fouling rate was constant and considered to be sufficiently long to be representative of the
166 period. The maximal TMP recommended for the membrane supplier was 400–500 mbar.

167 Table 2 – G-AnMBR operating conditions during the different operating periods.

Parameters	Period 1	Period 2	Period 3	Period 4
Days of operation (d)	1–134	135–203	204–289	290–357
$J_{P_{20,inst}}$ (LMH)	2.8 ± 0.1	3.6 ± 0.1	4.1 ± 0.3	6.0 ± 1.4
$J_{P_{20,net}}$ (LMH)	1.3 ± 0.1	1.8 ± 0.1	2.5 ± 0.2	3.6 ± 1.2
HRT (h)	13.9 ± 0.5	9.9 ± 0.3	7.0 ± 0.7	4.9 ± 1.4
OLR ($\text{kgCOD}\cdot\text{m}^{-3}\cdot\text{d}^{-1}$)	0.5 ± 0.1	0.8 ± 0.2	1.0 ± 0.4	1.6 ± 0.7
Temperature ($^{\circ}\text{C}$)	25.0 ± 0.3	25.0 ± 0.8	24.9 ± 1.2	25.3 ± 2.0
pH (-)	7.2 ± 0.6	7.0 ± 0.2	7.3 ± 0.2	7.4 ± 0.2
Redox (mV)	-488 ± 22	-488 ± 17	-494 ± 14	-478 ± 23
ULV ($\text{m}\cdot\text{h}^{-1}$)	2.6 ± 0.03	2.6 ± 0.03	2.6 ± 0.03	2.6 ± 0.03

168

169 2.3 Membrane fouling indicators

170 The extent of membrane fouling was evaluated through the monitoring of transmembrane
 171 pressure (TMP). A pressure gauge was installed on the permeate line, and data were recorded
 172 every 15 seconds. Data processing allowed calculation of the average maximal TMP per day.

173 Filtration resistance analysis was used to understand fouling mechanisms. Fouling resistance
 174 was measured using Darcy's law (Eq. 1) and the resistance-in-series model (Eq. 2).

$$R_t = \frac{TMP}{\mu_{20} \cdot J_{20}} \quad \text{Eq. 1}$$

$$R_t = R_m + R_f = R_m + (R_{reversible} + R_{irreversible} + R_{residual}) \quad \text{Eq. 2}$$

175 where R_t is the resistance (m^{-1}), TMP is the transmembrane pressure (Pa), J_{20} is the normalized
 176 flux at 20°C ($\text{L}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$), μ_{20} is the viscosity of water at 20°C (Pa.s), R_m is the membrane
 177 resistance (m^{-1}), R_f is the fouling resistance (m^{-1}) and $R_{reversible}$, $R_{irreversible}$ and $R_{residual}$ are the
 178 resistances caused by reversible fouling, irreversible fouling and residual fouling, respectively
 179 (m^{-1}).

180 All resistances were measured by filtering deionized water through the membrane. R_m was
181 measured with a clean membrane, R_t was measured at the end of experimental periods with a
182 fouled membrane and R_f was deduced. Subsequently, the fouled membrane was physically
183 cleaned by water rinsing with 5 L of deionized water allowing for calculation of $R_{reversible}$.
184 Finally, the membrane was chemically cleaned by soaking (2 h) in a 0.2% NaOCl solution,
185 providing the resistance removed by chemical cleaning $R_{irreversible}$ and the residual resistance
186 $R_{residual}$.

187 **2.4 Analytical procedures**

188 Influent, supernatant and effluent were sampled twice per week to measure the tCOD and sCOD
189 concentration using commercial kits (Hach, Germany, LCK 500, 314, 514). The supernatant
190 was sampled on the liquid recirculation line. The sCOD concentration was measured after pre-
191 filtration at 0.45 μ m. sCDO removal was divided into ‘biological removal’ and ‘membrane
192 removal’ (section 3.1.1). The sCOD removed between influent and supernatant was associated
193 to ‘biological removal’ whereas the sCOD removal measured between supernatant and
194 permeate was related to ‘membrane removal’. It should be noticed that the ‘membrane removal’
195 takes into account the membrane barrier rejection and the potential biological removal by the
196 biofilm developed on membrane surface. VFAs were analyzed regularly by ion-exclusion
197 chromatography (ICS-900, Dionex, USA; column BP-OA_2000, Benson Polymeric Inc., USA)
198 coupled with UV detector (210 nm). H₂SO₄ (0.05N) was used as eluent at 0.4 mL.min⁻¹. Six
199 VFAs were quantified, namely acetic acid, propionic acid, butyric acid, iso-butyric acid, iso-
200 valeric acid and valeric acid. All samples were passed through a 0.22 μ m filter prior to HPLC-
201 UV analysis. Mixed liquor suspended solids were measured according to standard methods
202 (APHA et al., 1998).

203 A three-dimensional fluorescence excitation emission matrix (3DEEM) was used to
204 characterize the composition of the DCOM of the foulant. After physical cleaning, the collected

205 foulant was mixed thoroughly and pre-filtered through a 1.2 μm filter to conserve the colloidal
206 compounds. A fluorescence spectrophotometer (FL 6500, Perkin-Elmer, USA) was used with
207 excitation and emission scan ranges of 200–500 nm and 280–600 nm, respectively. Data
208 analysis was performed according the protocol of Jacquin et al. (2017). Three regions of
209 fluorophores were distinguished, namely (a) region I + II, associated with protein-like
210 molecules, (b) region IV, corresponding to soluble microbial product (SMP)-like molecules,
211 and (c) region III + V, related to humic substances.

212 Quantification of proteins (PN) and polysaccharides (PS) was conducted following the Lowry
213 and Dubois methods, respectively (Dubois et al., 1951; Lowry et al., 1951). Bovine serum
214 albumin (BSA) and glucose were used as standards. All samples were pre-filtered through a
215 0.45 μm filter before PN and PS quantification.

216 The extracellular polymeric substances (EPS) of the fouling layer were quantified because they
217 play a key role in membrane fouling. The EPS basically can be divided into three fractions
218 depending on their structure: soluble EPS, also known as soluble microbial products (SMPs),
219 loosely-bound EPS (LB-EPS) and tightly-bound EPS (TB-EPS). The heating extraction method
220 described by Li and Yang (2007) was used. The concentration of each EPS fraction was
221 quantified as the sum of PN and PS contents.

222 **2.5 COD mass balance**

223 The COD mass balance ($\text{gCOD}\cdot\text{d}^{-1}$) was determined according to Eq. 3.

$$tCOD_{in} = tCOD_{out} + COD_{CH_4}^G + COD_{CH_4}^L + COD_{SO_4} + \Delta COD_{biomass} \quad \text{Eq. 3}$$

224 where $tCOD_{in}$ and $tCOD_{out}$ are the tCOD concentrations experimentally measured in the
225 influent and effluent ($\text{mg}\cdot\text{L}^{-1}$), respectively; $COD_{CH_4}^G$ and $COD_{CH_4}^L$ are the equivalent COD
226 concentrations of the produced gaseous and dissolved methane, respectively; COD_{SO_4}
227 corresponds to the COD used for the reduction of sulfate by sulfate-reducing bacteria (0.67

228 gCOD.gSO_4^{-1}); and $\Delta\text{COD}_{\text{biomass}}$ is the COD used for biomass synthesis ($0.1 \text{ gVSS.gCOD}^{-1}$).
229 The dissolved methane ($\text{COD}_{\text{CH}_4}^{\text{L}}$) was quantified experimentally following the headspace
230 method described by Sanchez et al. (2022). Total methane produced was calculated using the
231 theoretical value of $0.38 \text{ L-CH}_4/\text{gCOD}_{\text{removed}}$ at 25°C , allowing for the gaseous methane
232 ($\text{COD}_{\text{CH}_4}^{\text{G}}$) calculation.

233 **2.6 Preliminary economic evaluation**

234 The economic assessment was conducted for four scenarios, corresponding to the four operating
235 periods. The design parameters, process performance and membrane fouling rates of the four
236 experimental periods were used for the calculations. The economic analysis was performed
237 considering a wastewater flow rate of $20,000 \text{ m}^3.\text{d}^{-1}$ (i.e. 100,000 population equivalent).
238 Detailed information of the design criteria for each scenario can be found in Table A1 of the
239 supplementary data.

240 The capital and operating costs influenced by permeate flux, membrane fouling rate and process
241 performance were included in the economic evaluation. For this reason, the costs not influenced
242 by the evaluated parameters have not been included in the economic evaluation because they
243 are expected to be similar for the four scenarios. Capital expenditures (CAPEX) comprised the
244 purchase of the membranes, the bioreactor construction and the combined heat and power
245 (CHP) unit for biogas valorization. Operational expenditures (OPEX) comprised the chemical
246 reagents needed for membrane chemical cleaning, energy consumption for CHP unit and the
247 equipment replacement cost (i.e., membrane and CHP unit). Revenues corresponded to the
248 energy recovered from the biogas. All design and cost parameters used to calculate costs and
249 revenues are provided in Table A3 of the supplementary data.

250 The chemical cleaning protocol was adapted from Brepols et al. (2008) and used a 0.05%
251 NaOCl solution (2 h) and a $2,000 \text{ mg.L}^{-1}$ citric acid solution (2 h). A chemical cleaning was

252 considered required when the maximal applicable TMP of 400 mbar was reached. It was
253 assumed that the initial membrane permeability was recovered after each chemical cleaning.
254 Chemical cleanings have an impact on membrane lifetime because the chemicals damage the
255 membrane materials and modify their properties and performance (Chheang et al., 2022). The
256 replacement of the membrane was presumed to be conducted when a maximum cumulative
257 chlorine contact of $500,000 \text{ mg}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$ was reached (Robles et al., 2014; Vinardell et al., 2022).
258 The present value of the net cost (PV_{NC}) was calculated for the four scenarios as the difference
259 between the PV of the gross cost (CAPEX + OPEX) and the PV of the electricity revenue
260 considering three plant lifetimes (i.e., 20, 30 and 40 years) and a discount rate of 5%. Detailed
261 information regarding the equations used for the economic evaluation can be found in Table
262 A3 of the supplementary data.

263 **3. Results and discussion**

264 **3.1 Overall treatment performance**

265 **3.1.1 Biological and membrane rejection performances**

266 Table 3 summarizes the average tCOD, VFA and MLSS values for the four periods. The tCOD
267 removal efficiency ranged between 82.5% and 92.6%, with average tCOD concentrations in the
268 effluent varying from 22.5 to $52.9 \text{ mgCOD}\cdot\text{L}^{-1}$, which was in compliance with European Union
269 discharge standards (Directive 91/271/EEC). The MLSS in the permeate were below $5 \text{ mg}\cdot\text{L}^{-1}$
270 in every experimental period, which was considerably below the $35 \text{ mgSS}\cdot\text{L}^{-1}$ regulation
271 discharge (Directive 91/271/EEC). The concentration of MLSS in the supernatant slightly
272 increased with the decrease of HRT (Pearson's correlation coefficient $r = -0.852$). This result
273 was expected because (i) ultrafiltration membranes retain particulate matter, which can
274 accumulate over time and (ii) lower HRT induces higher OLR, increasing levels of particulate
275 and colloidal materials and thus promoting biomass growth (R. Chen et al., 2017b; Huang et

276 al., 2011). Solid concentration build-up near the membrane has been previously correlated with
 277 TMP increase (Gouveia et al., 2015).

278 Table 3 – Influent and effluent compositions and removal efficiencies (mean values \pm SD; $n \geq 10$).

Parameters	Units	Period 1	Period 2	Period 3	Period 4
OLR	kgCOD.m ⁻³ .d ⁻¹	0.5 \pm 0.1	0.8 \pm 0.2	1.0 \pm 0.4	1.6 \pm 0.7
tCOD _{in}	mgCOD/L	311.4 \pm 54.7	346.4 \pm 63.8	303.0 \pm 91.0	298.8 \pm 74.5
tCOD _{eff} = sCOD _{eff}	mgCOD/L	22.5 \pm 7.7	24.8 \pm 7.4	32.7 \pm 17.6	52.9 \pm 28.4
tCOD _{removal}	%	92.4 \pm 3.1	92.6 \pm 2.2	89.1 \pm 4.7	82.5 \pm 7.7
VFA _{inf}	mgCOD/L	209.2 \pm 43.2	159.2 \pm 38.0	170.7 \pm 10.0	190.3 \pm 11.1
VFA _{eff}	mgCOD/L	0.0 \pm 0.0	3.6 \pm 4.6	2.3 \pm 1.7	1.6 \pm 2.2
VFA _{removal}	%	100 \pm 0	97 \pm 4	96 \pm 5	99 \pm 1
MLSS _{supernatant}	mg/L	88 \pm 50	179 \pm 108	209 \pm 106	194 \pm 99
MLSS _{eff}	mg/L	3.6 \pm 3.9	1.3 \pm 2.1	1.8 \pm 0.7	3.8 \pm 2.8

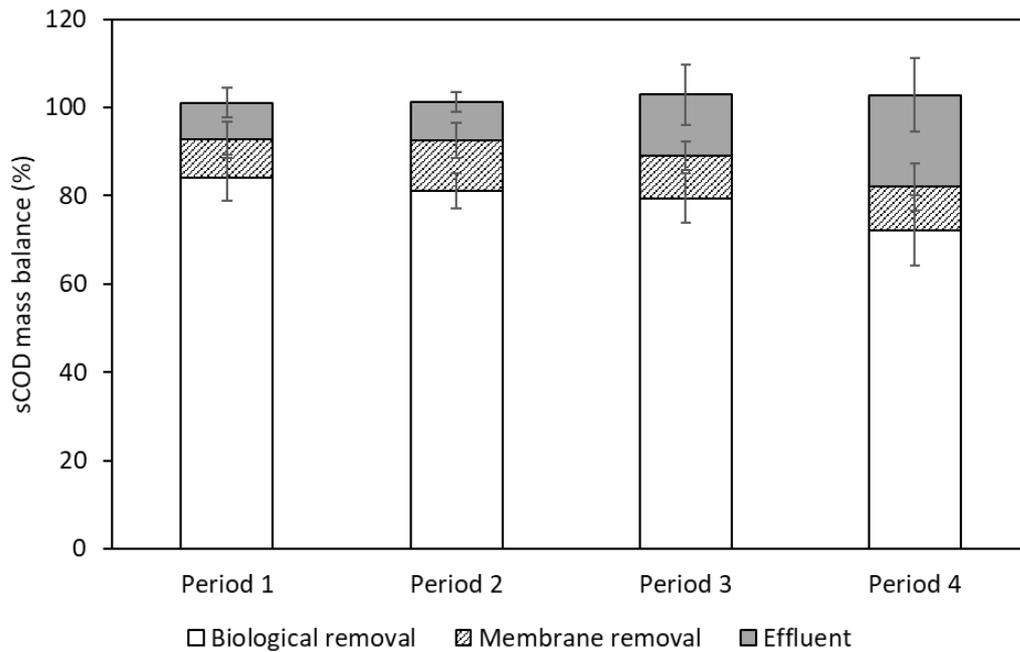
279

280 Fig. 1a presents the distribution of sCOD removal during the different periods. Under all
 281 conditions, 10 \pm 1.5% of the sCOD influent was removed by the physical membrane barrier
 282 and/or by the biomass attached to the membrane surface. This highlights that the effectiveness
 283 of membrane separation was not affected by the filtration flux. The retention of particulate
 284 matter and some DCOM by the membrane was observed to improve effluent quality further and
 285 enhance the removal rate of organic material (Gouveia et al., 2015; Sanchez et al., 2022; Smith
 286 et al., 2013). No statistical difference was observed among the tCOD removal efficiencies
 287 achieved in all experimental periods ($p > 0.05$) with the exception of Period 4.0 ($p < 0.05$). In
 288 this latter case, when the OLR increased from 1.0 to 1.6 kgCOD.m⁻³.d⁻¹, biological sCOD
 289 removal dropped to 57%. The G-AnMBR performances rose progressively for 45 days until
 290 reaching an average biological sCOD removal of 80.6 \pm 5.5 and a tCOD removal rate of 92.7
 291 \pm 1.5%. Thus, the anaerobic microbial community eventually adapted to harsh operational
 292 conditions, probably due to membrane retention that uncoupled HRT and SRT and thus allowed
 293 for acclimation of anaerobic bacteria and archaea (Stuckey, 2012). These results are consistent

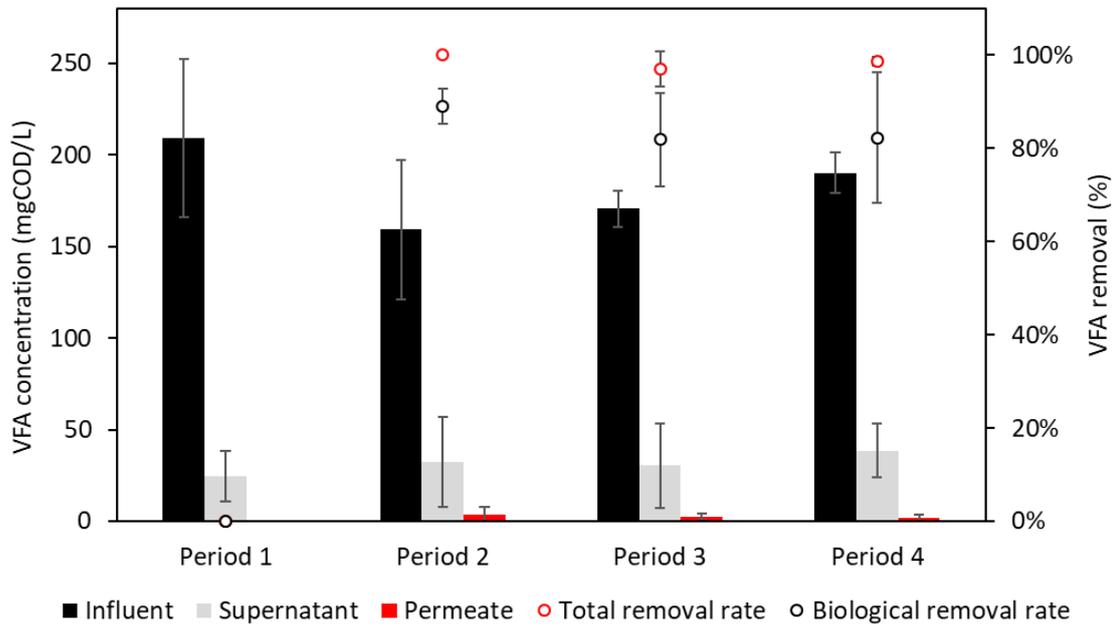
294 with those obtained by Ji et al. (2021), who operated a submerged AnMBR for domestic
295 wastewater treatment at 25°C under different operating conditions. For an OLR between 1.5
296 and 0.7 kgCOD.m⁻³.d⁻¹ (HRT 6–12 h), a steady high COD removal efficiency above 89% was
297 observed. However, the biological performance substantially decreased at an OLR of 2.1
298 kgCOD.m⁻³.d⁻¹ (HRT 4h), although the organic material retained by the membrane
299 compensated for the decrease in bioactivity (Ji et al., 2021a).

300 Fig. 1b shows the VFA concentrations measured in the influent, supernatant and effluent of the
301 G-AnMBR. The VFA concentration is expressed as the sum of all VFAs converted into COD
302 equivalents. Among the VFAs analyzed, acetic acid and propionic acid were the predominant
303 compounds (10–120 mg/L); butyric acid, iso-butyric acid and iso-valeric acid were near the
304 limit of quantification (<2 mg.L⁻¹); and valeric acid was below the level of detection (<0.5
305 mg.L⁻¹). VFAs are key intermediate products of anaerobic digestion and their concentrations
306 within the mixed liquor give an indication about the process stability and the proper functioning
307 of methane-producing archaea (van Lier et al., 2008). The lower the amount of VFA, the more
308 efficient the anaerobic reaction chain. Low amounts of VFAs were measured, and no VFA
309 accumulation was observed in the G-AnMBR supernatant during the experimental periods. This
310 analysis confirms that (i) a steady state was reached and (ii) the anaerobic bacteria and archaea
311 were consistent with all the conditions evaluated. Furthermore, Fig. 1b reveals very low VFA
312 concentration in the permeate (1.9 ± 1.3 mgCOD.L⁻¹), revealing that these compounds were
313 biologically degraded and additionally removed in the membrane separation step (Fig. 1b).
314 Theoretically, VFAs could pass through the membrane due to their low molecular weight but
315 this phenomenon has been previously observed, demonstrating the positive impact of fouling
316 layer biological activity (Chen et al., 2017a; Martinez-Sosa et al., 2011). Previous studies have
317 hypothesized that the biomass attached to membrane surface is considerably active, even more
318 active than suspended biomass, because of lower mass-transfer limitations (Smith et al., 2013).

(a)



(b)



319

320 Fig. 1 – (a) Distribution of sCOD biological removal, membrane removal and effluent content; (b) VFA
321 concentration and removal efficiency under all operating conditions (HRT 14, 10, 7, 5 h; OLR 0.5, 0.8, 1.0, 1.6
322 kgCOD.m⁻³.d⁻¹ for periods 1, 2, 3 and 4, respectively).

323 3.1.2 COD mass balance

324 COD mass balance distribution for the four operational conditions is provided in Table 4. The

325 COD contained in the effluent and the methane production were the two major factors affected

326 by the changes in OLR. As expected, the quantity of COD_{in} converted into methane increased

327 considerably, from 2.7 to 7.4 gCOD.d⁻¹ (2.7-fold), and the influent flux increased from 3.55 to
 328 10.70 gCOD.d⁻¹ (3-fold). However, the fraction of the tCOD converted into methane decreased
 329 from 77% to 70% as the OLR increased from 0.5 to 1.6 kg.m⁻³.d⁻¹, respectively. At the same
 330 time, the proportion of COD_{in} retrieved as dissolved methane (lost methane) decreased from
 331 27% to 10% when the OLR increased (Table 4). These results indicate that a trade-off must be
 332 found in the applied operating conditions to enable both a high rate of conversion of organic
 333 matter to methane and a lower concentration of dissolved methane.

334 Dissolved methane measured during the periods was lower as OLR increased ($r = -0.95$).
 335 Specifically, dissolved methane concentration was 21.2 ± 1.3 mg.L⁻¹ for Period 1, 20.5 ± 2.4
 336 mg.L⁻¹ for Period 2, 14.7 ± 2.1 mg.L⁻¹ for Period 3 and 9.3 ± 3.0 mg.L⁻¹ for Period 4. Similarly
 337 to the present study, Yeo et al. (2015) observed that the fraction of dissolved methane (relative
 338 to the total methane produced) decreased from 35 to 14% as the OLR increased from 0.4 to 1.1
 339 kgCOD.m⁻³.d⁻¹. The increase in gas production at higher OLR levels led to an increase in local
 340 turbulence and mass transfer, allowing more methane to escape the liquid phase and
 341 diminishing the level of supersaturation (Yeo et al., 2015).

342 Table 4 – COD mass balance during each experimental period.

Period	1		2		3		4	
	gCOD.d ⁻¹	%						
Influent COD	3.55	100	5.50	100	7.10	100	10.70	100
Effluent COD	0.27	8	0.41	7	0.83	12	1.74	16
Sludge growth	0.47	13	0.72	13	0.89	13	1.27	12
Sulfate reduction	0.10	3	0.16	3	0.20	3	0.28	3
Dissolved CH ₄	0.95	27	1.34	24	1.37	19	1.04	10
Gaseous CH ₄	1.77	50	2.86	52	3.82	54	6.37	60
Total CH ₄	2.72	77	4.20	76	5.19	73	7.41	70

343

3.2 Membrane fouling behavior

3.3.1 Membrane filtration performance

Fig. 2 shows the extent of membrane fouling through the TMP evolution along the operational periods. Table 5 provides the membrane fouling rates obtained from the TMP profiles. Substantial differences in TMP profiles were observed between each period. During Period 1 ($J_{p20,net} = 2.8$ LMH), TMP slowly increased to 46 mbar with a progressive rise of $0.03 \text{ kPa}\cdot\text{d}^{-1}$ (Table 5). The very low membrane fouling rate suggests that the membrane was operating at sub-critical filtration flux, as no severe fouling was observed. When the instantaneous filtration flux was increased to 3.6 LMH (Period 2), the TMP reached 305 mbar ($\Delta TMP = 236$ mbar) after 65 days of filtration. The TMP profile was characterized by a slow upward trend ($0.16 \text{ kPa}\cdot\text{d}^{-1}$), followed by a much faster fouling rate ($0.87 \text{ kPa}\cdot\text{d}^{-1}$), resulting in an average fouling rate estimated at approximately $0.40 \text{ kPa}\cdot\text{d}^{-1}$ (Table 5). In Period 3 ($J_{p20,inst} = 4.1$ LMH), TMP rapidly and steadily increased, with an average daily rise of $0.65 \text{ kPa}\cdot\text{d}^{-1}$, until a maximal TMP of 430 mbar was reached. During Period 4, the initial filtration flux targeted ($J_{p20,inst} = 7.5$ LMH) led to very strong and fast membrane fouling; the average instantaneous flux in this period was of 6.0 ± 1.4 LMH. The TMP increased from 40 mbar to 530 mbar within 9 days (day 290 to 299), and constant instantaneous filtration flux was unable to be sustained, thus decreasing slowly to 5.7 LMH. On day 303, the membrane was removed from the reactor for physical and chemical cleaning. The filtration for Period 4 was restarted, and the same sharp upward trend in TMP was observed after the membrane cleaning. The membrane fouling rate was approximately $2.86 \text{ kPa}\cdot\text{d}^{-1}$. On day 316, the maximal TMP was reached; then, the instantaneous permeate flux dropped progressively to 4.1 LMH until the end of the experiment. The concept of critical flux states that under fixed operating conditions (e.g., MLSS, hydrodynamics conditions, membrane properties) there is a threshold flux above which a sustainable flux cannot be further maintained (Bacchin et al., 2006). Fig. 2 clearly shows that

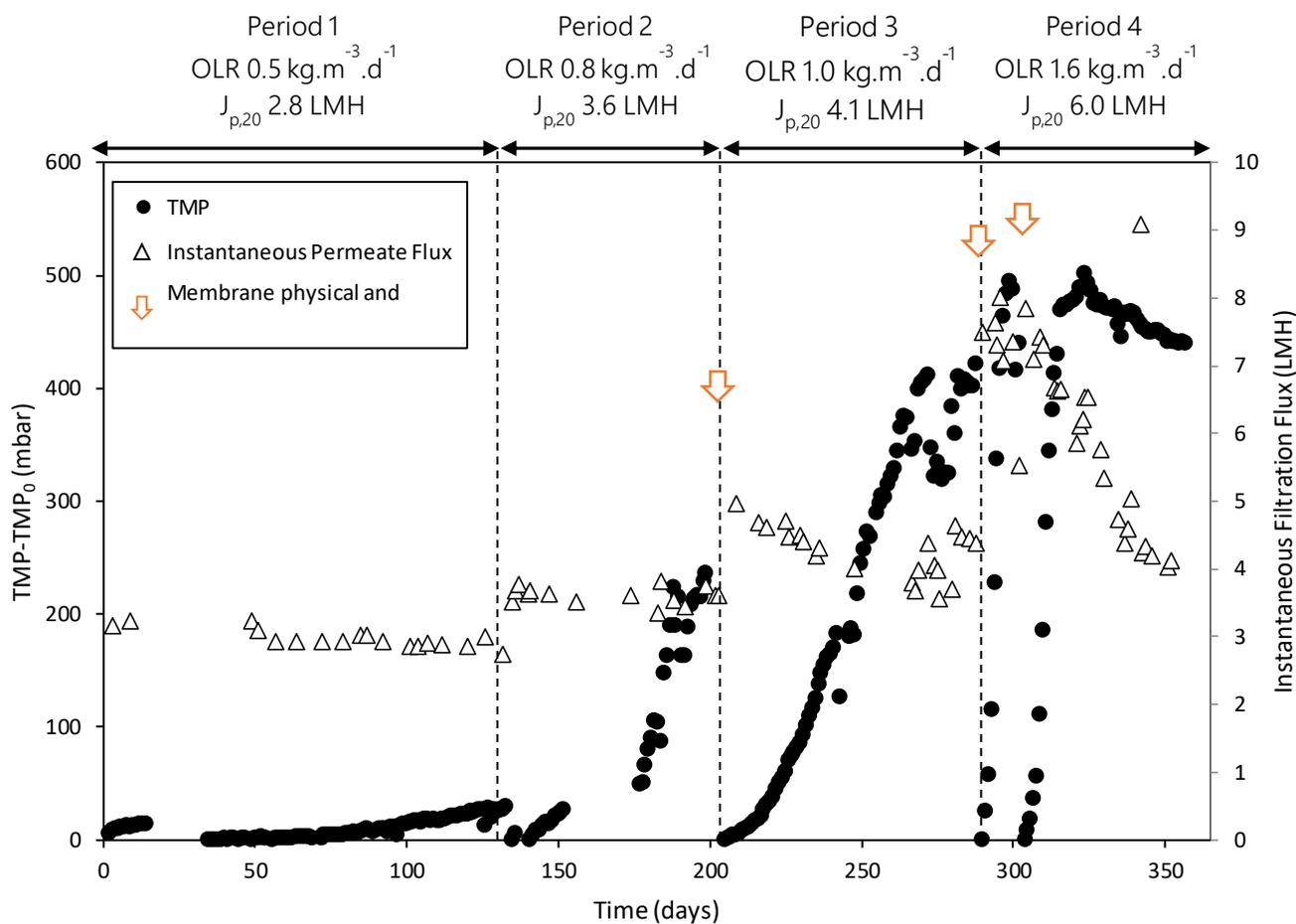
369 apart from Period 1 ($J_{p20,inst} = 2.8$ LMH), the critical flux was exceeded because apparent
 370 fouling was observed during the experimental periods. Conversely, a critical instantaneous flux
 371 of 5 LMH was obtained in a mesophilic (30°C) AnMBR at a TSS concentration of 50 g.L⁻¹
 372 when gas sparging ($U_g = 35$ m/h) was applied (Jeison and van Lier, 2006). The lower critical
 373 flux obtained in this study can be attributed to the very high concentration of solids and the
 374 poor shear rate due to the absence of gas sparging.

375 These results confirm that the TMP substantially increased with the permeate flux, which
 376 confirms the high impact of permeate flux on membrane fouling. The increase in filtration flux
 377 led to higher convective forces in such a way that foulants were pushed toward the membrane
 378 surface more harshly and quickly; thus, the lift and diffusion forces had less effect in carrying
 379 foulants away (Chen et al., 2017a).

380 Table 5 – Fouling rates at different filtration conditions.

Period	Filtration conditions	Fouling rates	
	$J_{p20,inst}$ (LMH)	$dTMP/dt$ (kPa.d ⁻¹)	dR/dt (m ⁻¹ .d ⁻¹)
1	2.8 ± 0.1	0.03	0.9 × 10 ¹¹
2	3.6 ± 0.1	0.40	9.3 × 10 ¹¹
3	4.1 ± 0.3	0.65	10.3 × 10 ¹¹
4	6.0 ± 1.4	2.86	34.0 × 10 ¹¹

381



382

383 Fig. 2 – Transmembrane pressure and instantaneous filtration flux during the operational periods.

384 **3.3.2 Membrane fouling resistances**

385 The extent of reversibility of the fouling layer was evaluated based on the distribution of the
 386 fouling resistances (Table 6). Because the fouling rate in Period 1 was very low (Table 5) and
 387 the TMP on day 134 was approximately 50 mbar, no membrane cleaning was performed
 388 between Period 1 and Period 2. Therefore, no membrane fouling resistance distribution is
 389 available for Period 1. Fouling resistance (R_f) corresponded to $15.0 \pm 1.1 \times 10^{12}$, $16.8 \pm 0.6 \times$
 390 10^{12} and $25.7 \pm 4.7 \times 10^{12} \text{ m}^{-1}$ for periods 2, 3 and 4, respectively. These results demonstrate
 391 that R_f was strongly correlated to permeate flux and OLR ($r > 0.99$, $p < 0.05$). $R_{reversible}$
 392 accounted for 100%, 97.9% and 97.7% of the membrane fouling resistance, indicating that
 393 fouling was mostly reversible under all the operating conditions tested. External deposition,
 394 especially in the cake layer, was the main mechanism involved in G-AnMBR fouling, in

395 agreement with previous studies (Anjum et al., 2021; Kaya et al., 2017; Lin et al., 2013). Only
 396 residual fouling of $0.1 \pm 0.0 \times 10^{12}$ and $0.2 \pm 0.1 \times 10^{12} \text{ m}^{-1}$ were measured for periods 3 and
 397 4. High filtration fluxes have been observed to promote the extent of intermediate pore blocking
 398 in the early filtration step, which may lead to more severe or resistant fouling (Lin et al., 2013;
 399 Yao et al., 2022). However, the contribution of residual fouling to the total resistance was minor
 400 in comparison to the clean membrane ($R_m = 0.8 \pm 0.1 \times 10^{12} \text{ m}^{-1}$).

401 These results are promising for mitigation of G-AnMBR fouling, suggesting that a well-
 402 designed backwash (i.e., physical cleaning) could lead to longer filtration performance. Further
 403 research is needed to optimize the backwash parameters, such as intensity, frequency and
 404 duration, for the purpose of developing sustainable G-AnMBR domestic wastewater treatment.

405 Table 6 – Fouling resistance distribution for experimental periods 2, 3 and 4.

Resistances	Units	Period 2	Period 3	Period 4
R_{fouling}	$\times 10^{12} \text{ m}^{-1}$	15.0 ± 1.1 (100%)	16.8 ± 0.6 (100%)	25.7 ± 4.7 (100%)
$R_{\text{reversible}}$	$\times 10^{12} \text{ m}^{-1}$	15.0 ± 1.1 (100%)	16.4 ± 0.7 (97.9%)	25.1 ± 4.6 (97.7%)
$R_{\text{irreversible}}$	$\times 10^{12} \text{ m}^{-1}$	–	0.3 ± 0.1 (1.9%)	0.4 ± 0.1 (1.6%)
R_{residual}	$\times 10^{12} \text{ m}^{-1}$	–	0.1 ± 0.0 (0.2%)	0.2 ± 0.1 (0.8%)

406

407 3.3.3 Membrane foulants characterization

408 The composition of the reversible fouling was characterized through 3DEEM analysis (Fig. 3a),
 409 while the 3DEEM content of mixed liquor and permeate have been analyzed and published in
 410 previously published study (Sanchez et al., 2022). Based on previous works, three regions of
 411 fluorophores were distinguished, namely (a) region I + II, associated with protein-like
 412 molecules, (b) region IV, corresponding to soluble microbial product (SMP)-like molecules,
 413 and (c) region III + V, related to humic substances (Chen et al., 2003; Jacquin et al., 2017) (see
 414 Fig. A2). Since polysaccharides are not analyzed through fluorescence excitation emission,

415 additional foulant characterization were performed through PN and PS quantification (Fig. 3b).
416 Among the three fluorescent regions investigated, the major volume of fluorescence belonged
417 to region I + II (72–85%), relating to colloidal and macromolecular protein-like compounds
418 (Jacquin et al., 2017). In addition, between 11% and 20% of the total fluorescence volume was
419 part of region III + V, corresponding to humic substances. Measurement of PN and PS contents,
420 presented in Fig. 3b, corroborates the predominance of PN materials within the fouling layer,
421 with proportions of PN approximately 55%, 77% and 84% for periods 2, 3 and 4, respectively.
422 It can therefore be assumed that PN in the form of colloidal materials and macromolecules
423 prevailed in the G-AnMBR foulant, in accordance with previous observations (R. Chen et al.,
424 2017a; Yao et al., 2020; Zhou et al., 2016). The PN to PS ratio (PN/PS) was found to be strongly
425 correlated with the decrease in HRT ($r = -0.996$, $p < 0.05$), demonstrating the rise in PN
426 concentration in the cake layer during the experimental periods. Several factors, including the
427 following, may be responsible for this phenomenon: (i) the increase of the permeate flux leads
428 to higher OLRs, which promote the growth of biomass and biopolymers (mainly composed of
429 PN and PS) (Huang et al., 2011); (ii) the decrease of HRT reduces degradation of polymers and
430 macromolecules because PN-based organic matter has been found to be more slowly and
431 weakly biodegradable than that based on PS, thus leading to PN accumulation (R. Chen et al.,
432 2017a; Yang et al., 2015); and (iii) PN has been reported to have higher hydrophobic
433 characteristics than PS. Accordingly, PN has more tendency to adhere to membrane surface
434 (Kaya et al., 2019).

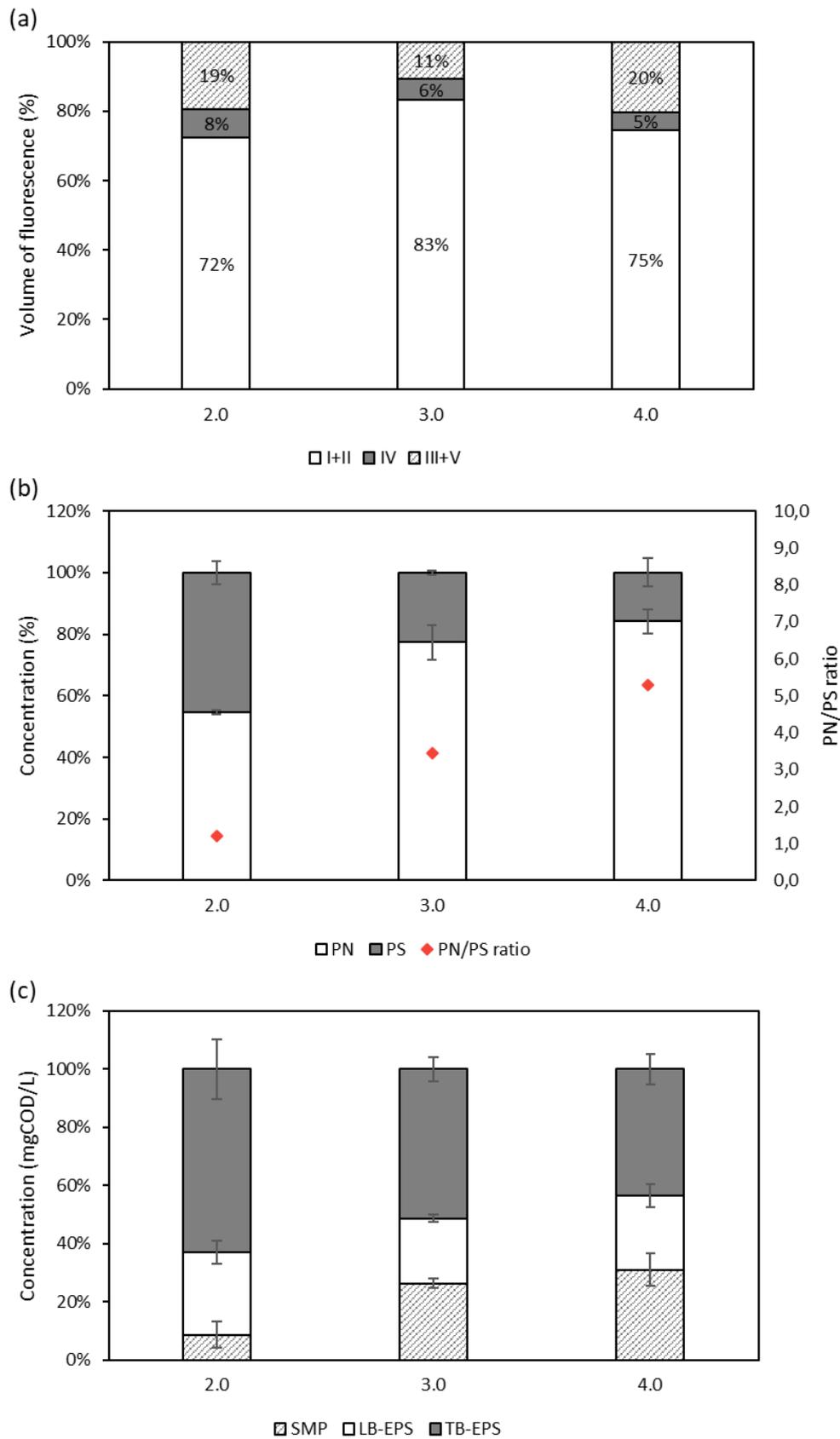
435 Because EPS have been identified as a key factor in membrane fouling (Anjum et al., 2021; R.
436 Chen et al., 2017a; Ding et al., 2015), the fouling layer was further fractionated into SMP, LB-
437 EPS and TB-EPS to identify the contribution of EPS fractions to total fouling. Fig. 3c shows
438 the repartition of the three EPS fractions extracted from the cake layer in periods 2, 3 and 4. Of
439 the three EPS fractions, the TB-EPS were in higher proportion within the fouling layer for all

440 the periods, followed by the SMP content and LB-EPS content. Nonetheless, the TB-EPS
441 content progressively decreased from 63% to 43% as the permeate flux increased from 1.8 to
442 3.6 LMH, respectively. Conversely, the fraction of SMPs extracted from the fouling layer
443 increased from 9% to 31% with the increase of permeate flux. The increase in SMPs was
444 probably caused by the OLR increase and HRT decrease, which stimulated production of SMPs
445 and reduced their degradation (Huang et al., 2011).

446 On the one hand, it might be concluded that bound-EPS (TB- and LB-EPS) represented an
447 important fraction of the foulants deposit because bound-EPS were mostly within the fouling
448 layer, in agreement with previous studies (R. Chen et al., 2017a; Gao et al., 2011). On the other
449 hand, the SMP fraction increased with the increase of filtration flux and fouling resistance,
450 which could suggest that SMPs play an important role in membrane fouling. Ding et al. (2015)
451 studied the extent of fouling caused by the three EPS fractions extracted from the cake sludge
452 of a mesophilic AnMBR. It was found that at the same TOC concentration, SMPs caused the
453 highest filtration resistance (50.2%) followed by LB-EPS (19.8%) and TB-EPS (30.0%). It was
454 suggested that the SMP fraction exhibited a lower energy barrier than the bound-EPS fractions,
455 so that SMPs needed to overcome weaker repulsive interaction energy to adhere to the
456 membrane surface (Ding et al., 2015). Accordingly, it is important to consider both the relative
457 contribution and specific fouling resistance when evaluating the impact of EPS fractions (i.e.,
458 SMP, LB-EPS, TB-EPS) on membrane fouling. Although no clear conclusion can be provided
459 regarding the major EPS contributor, the abundance of bound-EPS indicated the presence of
460 biomass attachment on the membrane surface, while the SMP revealed an implication of
461 dissolved compounds in global membrane fouling. Therefore, it is possible to conclude that the
462 fouling was mainly due to the combination of organic and biological fouling.

463 To summarize, fouling layer analysis indicated that PN materials were the main foulant in the
464 G-AnMBR system that was studied. PN content was highly influenced by the operating

465 conditions. EPS clearly played a relevant role in membrane fouling. The role of EPS fractions
466 in membrane fouling is still not clear; however, it is hypothesized that they can (i) participate
467 in biomass–membrane bonding (bound-EPS), (ii) form a strong interaction with the membrane
468 surface (SMPs) and (iii) create a wide EPS network, consolidating the cake layer.



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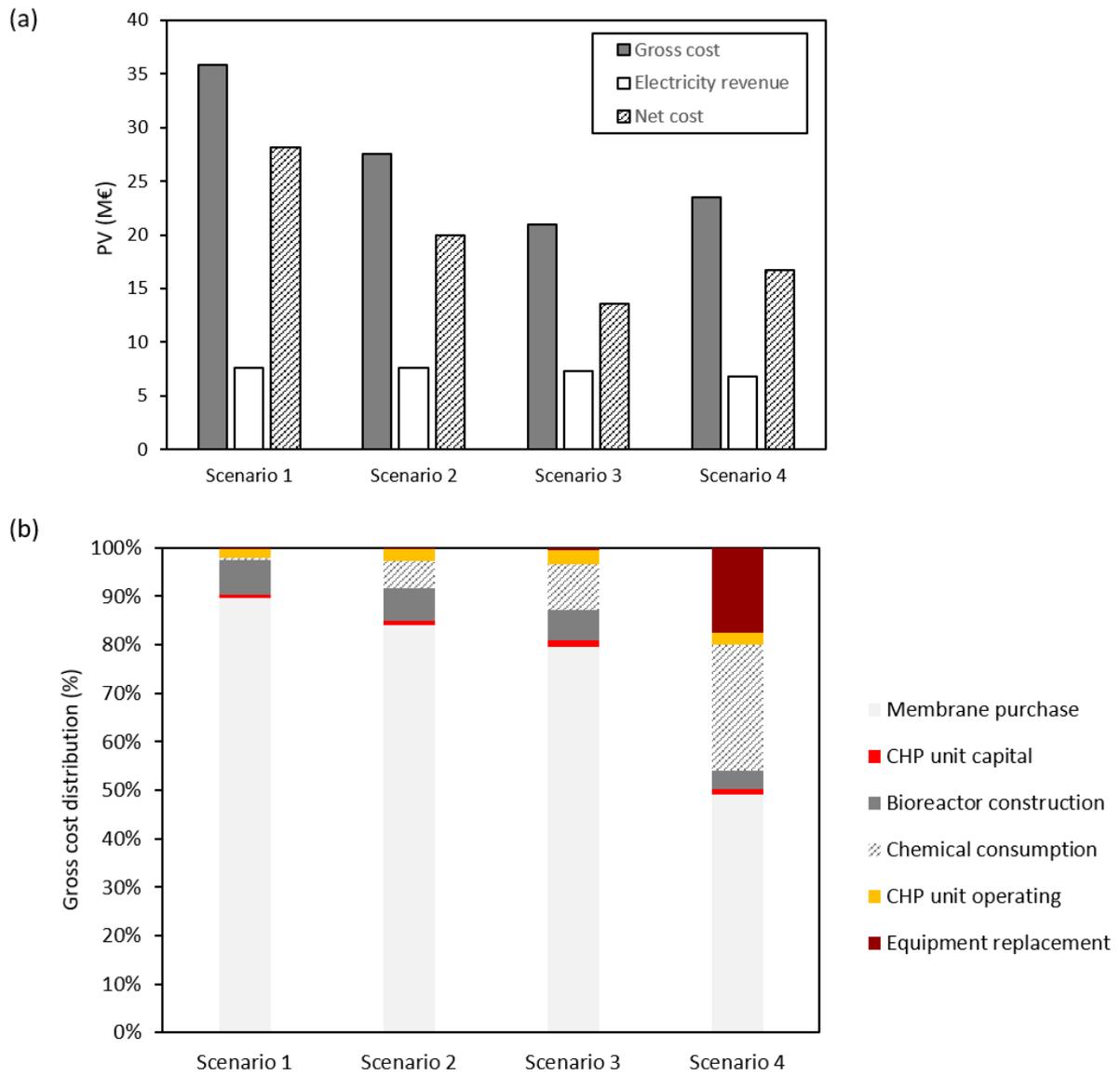
470 Fig. 3 – Composition of membrane foulants obtained through (a) 3DEEM (I + II: protein-like substances, III + V:
 471 humic-like substances, IV: SMP-like molecules); (b) PN and PS concentration; (c) EPS concentration.

3.3 Preliminary economic evaluation

472
473 Fig. 4a shows the gross cost, electricity revenue and net cost for the four scenarios evaluated
474 for a plant lifetime of 30 years. Fig. A3 and A4 of the supplementary information illustrate the
475 same results for a plant lifetime of 20 and 40 years, respectively. The results show that scenarios
476 3 and 4 featured the lowest net cost for the G-AnMBR system under study. The lower net cost
477 of these scenarios compared with scenarios 1 and 2 can be mainly attributed to the lower
478 membrane area required. This effect indicates that operating the membrane system at $J_{20,inst}$
479 below 3.6 LMH is not economically competitive for a G-AnMBR system operated without gas
480 sparging. Of note, the net cost of Scenario 3 ($J_{p20,inst}$ 4.1 LMH) was slightly lower than Scenario
481 4 (6.0 LMH) (Fig. 4a). In Scenario 4, high amounts of chemicals were needed for membrane
482 cleaning (due to the high fouling rate) with a direct impact on membrane replacement
483 frequency. In this regard, the extra costs regarding membrane cleaning and replacement in
484 Scenario 4 did not offset the lower membrane purchase costs when compared with Scenario 3.
485 Accordingly, it is important to achieve a compromise solution between operating the
486 membranes at relatively higher fluxes (CAPEX) while consuming a moderate amount of
487 chemicals (OPEX). In this regard, further pilot- and demonstration-scale studies are necessary
488 to determine, at larger scale, the optimum flux and chemical cleaning conditions to reduce the
489 costs and improve the economic competitiveness of the G-AnMBR system. Besides cost
490 considerations, the operation of the G-AnMBR operated without gas sparging has the potential
491 to make the WWTP energy positive due to the biogas (60-70% CH₄) produced in the system
492 (0.4-0.5 kWh·m⁻³).

493 Fig. 4b shows the gross cost distribution for the different scenarios. Detailed information
494 regarding the distribution of the gross costs can be found in Table A4 of the supplementary
495 information. Membrane purchase accounted for more than 49% of the gross cost for all the
496 scenarios, although its contribution progressively decreased from 89% to 49% as the permeate

497 flux increased from 2.8 to 6.0 LMH, respectively. The chemical cleaning also featured a
498 relatively important impact on the gross cost in scenarios 2, 3 and 4. The contribution of
499 chemical cleaning sharply increased from 0.5 to 26.0% as the permeate flux increased from 2.8
500 to 6.0 LMH, respectively, because the extent of membrane fouling was substantially higher at
501 higher fluxes. This was particularly important in Scenario 4, in which the high fouling rate and
502 chemical cleaning frequency made replacement of the membrane system necessary during the
503 plant's lifetime, with a direct impact on the OPEX (see Table A4 of the supplementary
504 information). This highlights that the membrane chemical cleaning strategy is a key economic
505 driver, especially in those scenarios in which intensive chemical cleaning is needed to control
506 long-term membrane fouling. Figure 4b also shows that the bioreactor construction contribution
507 decreased from 7.1 to 3.8% as the OLR increased from 0.5 to 1.6 kgCOD m⁻³ d⁻¹, respectively.
508 This illustrates that increasing OLR has also an impact on the economic balance of the system.
509 Overall, these economic results further corroborate that achieving a trade-off between
510 increasing the flux and OLR (system intensification) without requiring intensive physical and
511 chemical cleaning strategies to control membrane fouling is important for the economics of a
512 G-AnMBR system operated in submerged mode and without gas sparging.



513

514 Fig. 4– (a) Present value (PV) of the net cost, gross cost and electricity revenue; (b) gross cost distribution for
 515 the four scenarios evaluated for a 30-year plant lifetime.

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4. Conclusions

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The present work studied the impact of design parameters (i.e., OLR, permeate flux, HRT) on a G-AnMBR operated with a submerged UF membrane and without gas sparging as fouling control for mainstream domestic wastewater at ambient temperature. Four design OLRs were tested (between 0.5 and 1.6 kgCOD.m⁻³.d⁻¹) resulting in four HRT and permeate flux conditions.

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- The G-AnMBR achieved a rapid steady-state and high COD removal efficiencies (>90%) for HRTs ranging between 14 and 7 h.

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- The proportion of tCOD converted into methane was 77% at the lowest OLR and 70% at the highest OLR, confirming the decrease in biological activity with increasing OLR. However, more methane was recovered in the gaseous phase at the highest OLR (86% of the total methane produced) in comparison to the lowest OLR (65% of the total methane produced). Hence, at higher OLRs, the amount of bio-methane energy lost in the effluent and the greenhouse gas emissions were lowered.

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- Membrane fouling increased with the permeate flux and OLR, highlighting the important role of those parameters in cake layer formation and build-up. Under all conditions, proteinaceous colloidal and macromolecules were the main components of the G-AnMBR cake layer. Membrane permeability was almost totally recovered after physical cleaning with water rinsing, demonstrating that fouling was mainly reversible.

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- Finally, the economic evaluation showed that filtration flux, and the resulting membrane surface area, is a crucial parameter for G-AnMBR economics, with Scenario 3 (4.1 LMH) being the most economically favorable option.

540 Overall, these results have shown that achieving a compromise solution considering
541 permeate flux, OLR and chemical cleaning for membrane fouling control is important to
542 reduce the net cost of G-AnMBR systems operated without gas sparging.

543 On this basis, the G-AnMBR could be a sustainable and efficient process for domestic
544 wastewater treatment for local applications in which important financial, technical and
545 energy resources cannot be deployed.

546 **Supplementary information**

547 E-supplementary data for this work can be found in the e-version of this paper online.

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