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# Guano-related phosphate-rich minerals in European caves

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#### Abstract:

Guano is a typical deposit found in caves derived from the excretions of bats and in minor cases of birds. These organic deposits decompose and form a series of acid fluids and gases that can interact with the minerals, sediments, and rocks present in the cave. Over sixty phosphates are known and described from caves, but guano decay also often leads to the formation of nitrates and sulfates. In this study twenty-two European caves were investigated for their guano-related secondary minerals. Using various analytical techniques, seventeen phosphates, along with one sulfate (gypsum), were recognized as secondary products of guano decay. Among those minerals, some are very rare and result from the interaction of guano leachates with clays, fluvial deposits, or pyrite. Some of these minerals are even found only in the studied caves (spheniscidite, robertsite). The most common minerals belong to the apatite group. The common mineral association present in fresh decaying guano is brushiteardealite-gypsum, minerals that usually are not present in older deposits because of their higher solubility. Most minerals are in hydrated form because of the wet cave environment; however, some specific dry conditions may favor the presence of dehydrated minerals, such as berlinite, formed during guano combustion. Investigation on the acidity of guano piles shows pH values as low as 3.5 with an increase of acidity with age and depth. Finally, cave guano deposits should be better studied in the future because of their role in paleoenvironmental and paleoclimatic reconstructions and because it is important to better understand the origin of guano-related minerals, especially the phosphates and sulfates. Among all of the caves studied, Corona 'e sa Craba (Italy) and Domica-Baradla Cave (Slovakia-Hungary) are considered to be outstanding sites with respect to their phosphate mineralogy.

#### Keywords: secondary cave minerals, phosphates, minerogenesis, limestone caves, bat guano

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#### INTRODUCTION

Phosphate minerals are commonly found in caves containing guano accumulations from bat colonies (e.g., Martini, 1996; Hill, 1999; Onac & Vereş, 2003) or

in caves which contain a large quantity of bone deposits (Goldberg & Nathan, 1975; Karkanas et al., 2002). The phosphates result from the interaction between guano-derived leachates with cave bedrock, generally limestone, (calcite) speleothems or cave sediments

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(detrital clay deposits). A review of cave phosphate minerals was first published by Hill & Forti (1997), reporting a total of fifty-two mineral species. The minerogenetic mechanisms forming cave phosphates are microbial decay, dissolution, double replacement, and redox reactions (Onac & Forti, 2011b). Around 60 cave phosphate minerals are currently known (Onac & Forti, 2011a), with five of common occurrence: hydroxylapatite, brushite, ardealite (Schadler, 1932; Pogson et al., 2011), taranakite (Hill & Forti, 1997), and variscite (Onac et al., 2004). They usually display as crusts, nodules, lenses, and earthy or powdery masses (Onac, 2012). Some cave phosphate minerals are very rare, such as tinslevite (Marincea et al., 2002), foggite and churchite-(Y) (Onac et al., 2005a), hydroxylellestadite (Onac et al., 2006a), and berlinite (Onac & White, 2003; Marincea & Dumitras, 2005; Onac & Effenberger, 2007; Sauro et al., 2014). Others have been found only in a limited number of caves around the world, such as vashegyite (Forti et al., 2000), stercorite (Frost & Palmer, 2011), newberyite (Bridge, 1977; Frost et al., 2011), and crandallite (Kaye, 1959; Frost et al., 2012).

Through microbial activity the breakdown of guano produces strong acids (sulfuric, phosphoric), whereas degradation of bat urea yields nitric and other weak organic acids. All processes contribute to increase acidity toward the base of the guano accumulation, with pH as low as 2-4 (Martini, 1993; Hill & Forti, 1997; Martini, 2000; Forti, 2001). Nitrogen and sulfur can also be active above the guano accumulation on the ceiling and walls, through their gaseous phases (NH<sub>3</sub> and H<sub>2</sub>S, respectively). These gases rise through warm air convections produced by exothermic guano breakdown or by bat colonies, then oxidize, often through microbial mediation, into nitric and sulfuric acids, respectively, and together with CO<sub>2</sub> eventually strongly corrode the limestone host rock and calcite speleothems (Hill & Forti, 1997; Martini, 2000; Audra et al., 2016; Onac, 2019). Contrary to sulfur and nitrogen, phosphorus does not have a gaseous phase at ambient temperature.

With time and increase of guano accumulation depth, organic matter decays, and the organic compounds gradually transform into secondary mineral deposits. The oxidation of organic phosphorous into  $PO_4^{3-}$  ion seems to be at least partly driven by microorganisms (Forti, 2001; Chang et al., 2010). Very soluble nitrogen is washed away, as ammonium phosphate  $(NH_4)_2HPO_4$  transforms into calcium phosphate  $(CaHPO_4)_2$  allowing the ammonium ion  $(NH_4^+)$  to be carried away, according to the following reaction (Frost & Palmer, 2011):

$$(NH_4)_2HPO_4 + Ca^{2+} + 2H_2O \rightarrow CaHPO_4 \cdot 2H_2O + 2NH_4^+$$

After nitrogen, easily soluble sulfur, in the form of H<sub>2</sub>SO<sub>4</sub>, produced by sulfo-oxidant bacteria (Yoshimura et al., 1989) is consumed thereby producing gypsum, whereas phosphate (PO<sub>4</sub><sup>3-</sup>) and cations (Fe<sup>2+</sup>, Fe<sup>3+</sup>, Al<sup>3+</sup>, K<sup>+</sup>) gradually increase by relative enrichment, making the P/S and P/N ratios increase with depth (and age) of the guano pile (Hill & Forti, 1997; Shahack-Gross et al., 2004; Onac, 2012). Within the guano

accumulation, sulfuric acid reacts with the bedrock and with speleothems and produces sulfates, mainly gypsum (Hill & Forti, 1997), whereas the reaction with phosphoric acid produces phosphates (Onac & Veres, 2003). Both gypsum and phosphates are the only stable compounds in such acidic environment (Pogson et al., 2011). At the base of guano accumulations, after leaching of alkali ions (Onac, 2012), the interaction with limestone bedrock (or calcite speleothems) reduces acidity to a pH close to 6-7 and produces Carich stable phosphates such as hydroxylapatite and brushite (Hill & Forti, 1997). Fluorine (F-) can easily replace the hydroxyl ion (OH-) generating almost pure fluorapatite (Onac, 2012). The contact with clay materials allows cation exchange (such as Fe<sup>2+</sup>, Fe<sup>3+</sup>, Al<sup>3+</sup>, and K<sup>+</sup>) with production of other phosphate minerals, mainly taranakite and variscite, or less frequent mineral species (see Hill & Forti, 1997). Deeper in the sediment, still acidic conditions allow the vertical leaching of Mn compounds that form a black accumulation horizon marking the boundary of the phosphatization front with the unaltered sediment (Martini, 1993).

Environmental conditions play a key-role in the formation or inhibition of phosphate minerals (Onac & Veres, 2003; Shahack-Gross et al., 2003; Puscas et al., 2014). From a Romanian cave study, Onac & Vereş (2003) stated that pH, humidity, alkali content, and Ca/P ratio are the major parameters controlling the mineral assemblages. Variscite, taranakite, and brushite are stable under acidic conditions, whereas hydroxylapatite indicates less acidic conditions. In low-humidity conditions (and higher temperature), dehydration occurs, causing new minerals to form (i.e., berlinite). In arid regions with extremely dry conditions, very soluble species may precipitate around guano accumulation through evaporation, producing sylvite and nitrate minerals such as saltpeter (Martini, 1993). On the contrary, in damp places, the complete breakdown of guano at very low pH (2-4) leads to pure gypsum displayed as candyfloss (Martini, 1993). Therefore, phosphate minerals, combined with stable isotope analyses on guano, can be considered as good proxies of past environmental conditions, at both regional and local scales (Onac et al., 2014).

Gypsum often occurs together with secondary phosphate minerals and can be considered in most cases a by-product of guano decay. This mineral probably always forms because of the interaction of sulfuric acid with the carbonate host rock (often limestone) or with the Ca-rich dripping water, but in humid conditions it is generally dissolved away by seeping waters. Gypsum, and more generally sulfates, are thus mainly found in drier cave conditions in which they can persist (do not go into solution) and where evaporation processes dominate.

This paper reports the mineralogical findings of guano-related minerals (phosphates and sulfates) in a variety of European caves, from Hungary, Slovak Republic, France, Macedonia, and Italy, along with the processes and environmental conditions that control their occurrences.

#### **METHODS**

**Sampling**. Samples were taken in separate containers or plastic bags, scraping or scratching from the walls or rocks with a chisel, or collecting enough material with a clean spatula. In some cases, physical environmental parameters, temperature (T), relative humidity (RH) and  $CO_2$  concentration were measured using a pSENSE RH from SENSEAIR AB, with a precision of  $\pm 0.6^{\circ}$ C,  $\pm 3-5\%$  RH,  $\pm 30$  ppm  $CO_2$ .

**pH**. The pH of guano deposits was determined using a pH meter in a 1:5 proportional guano: distilled water suspension.

X-ray diffraction. Samples were analyzed on different diffractometers: a Philips PW 1050/25 (40 kV and 20 mA, CuKa radiation, Ni filter) at the University of Modena and Reggio Emilia, Italy; a Philips (40 kV and 20 mA, CoKa radiation, Graphite filter) at the CEREGE - CNRS, Aix-Marseille University, France; a Philips PW3710 (40 kV and 20 mA, CoKa radiation, Fe filter) interfaced with Philips High Score software package for data acquisition and processing at the DISTAV (Genoa University); a Bruker Analytical X-Ray System, D8 Endeavor (50 kV and 40 mA, CuKa radiation, Ni filter) at the University of South Florida; and a Panalytical Xpert Pro (60 kV and 200 mA, Cu Ka radiation) at CINaM - CNRS and Aix-Marseille University. All samples were scanned from 5° to 75° 2θ with a step increment of 0.02°, a scan speed of 0.5 s/step. Mineral identification and abundance were evaluated semi-quantitatively by determining peak intensities (peak height) of the X-ray diffraction results with the DIFFRACPhus EVA V.8.0 software and compared to the ICDD-PDF2 database for phase identification.

radiometric dating. Sixteen bat guano samples and a piece of wood found at the base of the Raganeous guano heap were radiocarbon dated by means of accelerator mass spectrometry (AMS) in the Poznań Laboratory (Poland). Additionally, thirteen other bat guano samples were dated by the ARTEMIS Radiocarbon Laboratory Facility (LMC14, Saclay France) and one sample was dated at the Centre de Datation de Radiocarbone (CDRC) in Lyon (France). Radiocarbon calibrated ages were obtained using the online program CALIB 7.10 (Stuiver and Reimer, 1993) with the IntCal13 atmospheric calibration curve (Reimer et al., 2013). All radiocarbon ages (20), including those from previous studies, are reported in calibrated years as cal yr. AD and BC. In addition, we used 6 results from previous studies (Krištůfek et al., 2008).

**Micro-Raman spectroscopy**. Since XRD spectra are sometimes less relevant for certain phosphate minerals, further information was obtained from micro-Raman spectroscopy. These analyses were carried out at the Unidad Asociada UVA-CSIC at the Centro de Astrobiología, University of Valladolid (Valladolid, Spain). Up to 11 Raman spectra were taken from different areas of each sample. The excitation source was a Laser Research Electro-Optics (REO) working at 632.8 nm. The KOSI HoloSpec f/1.8i spectrometer from Kaiser Optical covered a spectral

range of 150-3,800 cm<sup>-1</sup> and a spectral resolution of 5 cm<sup>-1</sup>, while the CCD (charge coupled device) employed was a DV420A-OE-130 model from Andor. The Raman head used was KOSI MKII, HFPH-FC-S-632.8 model from Kaiser Optical Systems coupled by optical fiber to a Nikon Eclipse E600 microscope, which in turn, was attached to a JVC TK-C1381EG video camera for visual analysis and precise control of the measured spots. An objective of 50× allowed microanalyses of 15 µm diameter spots. The laser power on the sample was maintained around 15 mW (corresponding irradiance of 76 kW/cm²) to ensure no thermal damage occurred to the samples. Typical integration time for spectral acquisition was 10 s and 10 accumulations were done. The sample was manually scanned, whereas the height of focus adjusted in order to optimize the intensity of the spectra signals.

In Isturitz Cave, an in-situ mineral characterization was carried out by the Departamento de Química Analítica, Universidad del País Vasco, using a portable Raman spectrometer innoRam (B & WTEKINC., Newark, USA), equipped with a 785 nm excitation laser (with 225 mW of nominal power), with a measurement area of about 100 µm. The Raman signals are collected by a refrigerated CCD detector using the Peltier effect. The spectrometer allows analyzing a fixed spectral range between 100 and 3,500 cm<sup>-1</sup>, with an average spectral resolution of 3 cm<sup>-1</sup>. Twenty measures by point were taken for representativity of the analysis. Spectra were produced using BWSpecTM v.4.02\_15 (B & WTEKINC., Newark, USA) and data processed using Omnic 7.2 (Nicolet).

Scanning electron microscopy (SEM). Scanning electron microprobe analyses were performed with a SEM VEGA3 TESCAN (DISTAV, Genoa University) operated at 20 kV and equipped with the EDAX-APOLLO\_X DPP3 energy-dispersive (EDS) X-ray spectrometer having an ultrathin polymer window and with resolutions for Manganese Ka = 126 eV and typically ranging detection for chemical elements of atomic number greater than 5 (Boron). Data acquisition and elaboration were performed with the TEAM Enhanced Version: V4.2.2 EDS software. Qualitative chemical analyses were also carried out on a Philips XL40 environmental scanning electron microscope (ESEM) equipped with an energy dispersive spectrometer (EDS-EDAX 9900) at the C.I.G.S. (Centro Interdipartimentale Grandi Strumenti) of the Modena and Reggio Emilia University.

Subsequently, elemental characterization by Energy Dispersive X-ray Fluorescence (EDXRF) was carried out using XMET5100 (Oxford Instruments), equipped with a Rhodium X-ray tube having a maximum voltage of 45 kV and a high-resolution silicon detector (SDD) with a spectral resolution of 20 eV and an energy resolution of 150 eV (measured with the K-alpha transition of manganese at -20°C). Each point was measured at least three times. Some samples of the Grotta dei Personaggi and Salnitro cave (Sicily) were analyzed with a SEM LEO 440 equipped with a EDS Oxford ISIS system with a Si (Li) PENTAFET detector (DiSTeM, Palermo University).

# CAVES AND SAMPLING SITES DESCRIPTION

We studied 22 caves systems in Europe where guano deposits and bat colonies are or were present: one on the border between the Slovak Republic and Hungary, seven in France, three in Macedonia, and the remaining eleven in Italy (eight of which on the island of Sicily).

# Domica-Baradla (Kečovo, Slovakia and Aggetlek, Jósvafő, Hungary)

The Domica-Baradla cave system contains 21 km of passages developing across the Slovakian and Hungarian border (Fig. 1). It has been included in the UNESCO World Heritage List since 1995. Domica Cave has its main entrance at 339 m asl, whereas the mountains range from 180 to 947 m asl (Gallay et al., 2015). The cave is formed in the Middle Triassic (Ladinian) lagoon limestones of Wetterstein type (Gaál & Vlček, 2011). The main cave displays large passages of up to 25 m diameter. It is fed by multiple sinking rivers that bring sediments (mainly clay) from outside. Cosmogenic (Al-Be) nuclide dating (3.47 ± 0.78 Ma) of the oldest gravels in the upper level shows that the cave originated in or before middle Pliocene (Bella et al., 2019). Relative humidity ranges from 95 to 98%, and temperature from 10.2 to 11.4°C, placing the cave among the warmest in Slovakia according to its latitude and altitude (Bella & Lalkovič, 2001). The partial pressure of CO<sub>2</sub> of the cave air is 1700-1800 ppm (July 2015). Vegetation above the cave comprises shrub and deciduous forest. The cave is equipped for tourist visits with 1.6 km of foot trails and it attracts 30,000 visitors per year.

Past records of bat species mention the presence of Rhinolophus hipposideros, R. euryale, Plecotus auritus, Myotis myotis, M. emarginatus, M. natterei, and Eptesicus discolor (Kettner, 1948). Currently, a colony of up to 1,500-2,500 individuals of Mediterranean horseshoe bats (R. euryale) use Domica Cave during their pre-hibernation stage (Kováč et al., 2014). Due to the large extension and dimension of passages, bats are settling up to more than 1 km away from the closest entrance. Many guano deposits are present as small accumulations from individual bats, or as large guano heaps. Some cone-shaped heaps are actively accumulating such as in Prales Chamber or in Palmový háj, which basal radiocarbon date is at about AD 990 (Krištůfek et al., 2008). Other guano accumulations are not active anymore such as in Libanon-hegy in Baradla Cave. Previous phosphate studies mention the occurrence of gypsum, brushite, and "apatite" (Kettner, 1948). The soluble brushite, associated to rims of calcite, deposited as wall crusts from leachates through above-lying guano deposits only when dry conditions were present. Gypsum, associated with brushite was identified at the base of guano piles, in guano pots (small depressions on the cave floor created by guano and its acids), and at the top of large stalagmites that collected bat droppings. In Palmový háj, Kereskényi (2014) identified brushite and traces of ardealite at the guano surface, and taranakite,

gypsum, and hydroxylapatite at depth. In Baradla Cave, the same author identified hydroxylapatite and taranakite.

The following three sites were sampled: in Prales Chamber (Fig. 1A), close to the show cave entrance of Domica, guano corrosion pots on top of some stalagmites originating from guano drops of individual bats were found to contain some secondary mineral deposits (Fig. 2A). Below the fresh guano, we sampled a whitish-yellowish paste (Sample Pr. B) covering weathered limestone (Pr. A). In Teknösbéka (Turtle Gallery; Fig. 1B), close to the show cave entrance of Baradla, ceiling cupolas are covered with patches of dark brown crusts (BA 3 in Fig. 2B). Libanon-Hegy is a chamber located at about 1.5 km from Baradla Cave entrance (Fig. 1C), where we found some recent guano deposits. The BA 4 guano accumulation is old and no longer active; it displays a dark brown compact and sticky material, about 50 cm thick (Fig. 2C). Here we sampled a white pasty wet layer (BA 4a), another white layer located above clay (BA 4b), a sticky black layer at the contact of the limestone base (BA 4c), and a wet pasty pure white deposit also at the contact with the limestone base (BA 4d) at 5 cm above the base of the guano deposit.

### Julio Cave (Saint-Étienne-d'Albagnan, Hérault, France)

Julio Cave is located in the Montagne Noire, a southern segment of the French Massif Central. The cave opens along the Jaur Valley, at 205 m asl. It is a several hundred meters long maze, developed along a steep dip (about 60°) of the Lower Devonian limestone (Fig. 3). The lower levels contain large cobbles introduced by the Jaur River. Decantation silts and fine sands are present up to the top of the cave, 73 m above the entrance. They contain abundant allogenic minerals (quartz, mica, feldspars) from past flooding events. The cave was mined for guano in the early 20<sup>th</sup> century. A recent study regarding important bat colonies mentions Miniopterus schreibersii (3,000 individuals), Rhinolophus euryale (800 individuals), Myotis capaccinii, and R. ferrumequinum (Nemoz, 2008). The cave is included in a Natura 2000 protected area. In its upper part, old and fresh guano cover the silty-sandy sediments that are entirely weathered to a grey or white toothpaste-like material. We sampled the lower grey layer (Vezelle 3) and the upper white one (Vezelle 2).

### Isturitz-Oxocelhaya Caves (Isturitz and Saint-Martin-d'Arberoue, Pyrénées-Atlantiques, France)

The Isturitz-Oxocelhaya caves are located on the NW piedmont of the Pyrenees Range. The *Gatzelu* Hill (209 m asl) is built of Lower Cretaceous (Aptian) limestones surrounded by marls and appearing as a rocky spur blocking the *Arberoue* Valley. The homonymous river crossed the *Gatzelu* Hill forming through-caves and leaving fluvial sediments within them. Four tiers are arranged over an altitudinal range of 75 m, the lowest one still active (Fig. 4). The caves were once intensively used as shelters by animals, mainly bears and bats, and by humans since the Middle Paleolithic,

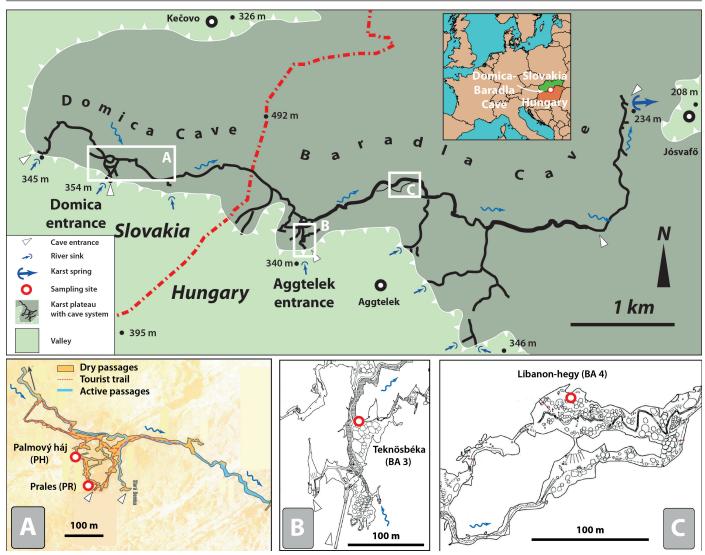


Fig. 1. Plan view of Domica-Baradla cave system in Slovakia and Hungary, with location (A-C) of studied places (redrawn after Ország et al., 1989). Detailed survey of Domica Cave after Slovak Cave Association (<a href="http://www.ssj.sk/en/jaskyna/7-domica-cave">http://www.ssj.sk/en/jaskyna/7-domica-cave</a>) and of Baradla Cave after Ország et al. (1989).

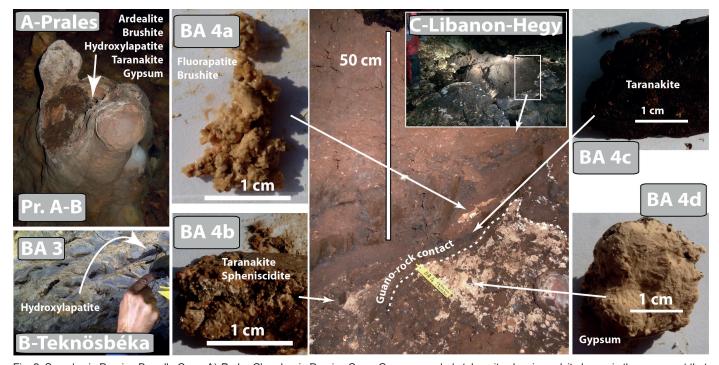


Fig. 2. Samples in Domica-Baradla Cave. A) *Prales* Chamber in Domica Cave. Guano-corroded stalagmite showing calcite layers in the guano pot that perforates its upper part. Below the guano, a whitish-yellowish paste (Pr. B) covering weathered limestone (Pr. A) was sampled (photo by V. Krištůfek); B) *Teknösbéka* (Turtle Gallery) in Baradla Cave. Dark brown apatite crusts (BA 3) on the ceiling of a cupola (Photo by A. Chroňáková); C) The guano accumulation in *Libanon-Hegy*, Baradla Cave (Photo by V. Krištůfek) with the location and mineralogy of BA 4 samples (Photos by P. Audra).

as shown by major art testimonies such as paintings and engravings (Garate et al., 2013). Also, guano and phosphates were intensively mined during the second half of the 19<sup>th</sup> century. After being classified as a Historical Monument in 1953 and opened as a show cave, all its entrances were closed. Consequently, bat colonies (mainly *Myotis myotis*) disappeared. Currently, only a couple of bats are still present (*Rhinolophus hipposideros*, *R. ferrumequinum*).

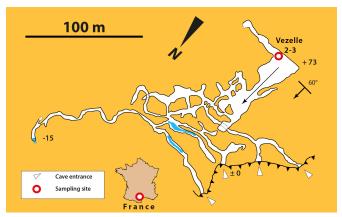


Fig. 3. Plan view of the Julio Cave with sampling-site locations (Survey after J. Fauré, SC Béziers, 1981, redrawn).

Four sites were investigated in-situ using portable Raman spectrometer and XRF (Figs. 4-5). An artificial tunnel allows the connection of the Isturitz upper level with the Oxocelhaya middle level. Before the tunnel enters the host rock it cuts and displays the entire sediment filling of Isturitz Cave along the staircase ("Staircase" sample, Fig. 5A). The lower part of the sediment filling resting on the rocky floor consists of fine colored sands (yellow, orange) separated by a dark layer. The floor of the *Rhinolophes* Chamber, located in Isturitz Cave, is covered by boulders. Boulders are partly covered by white crusts ("*Rhinolophes*" sample, Fig. 5B). Below these crusts, the limestone rock is

weathered and has a powdery appearance. The "Great Pilar" is a huge calcite column, deeply corroded by condensation niches linked to aerial corrosion above guano (Audra et al., 2016). We analyzed dark crusts and corroded white surfaces ("Great Pilar" sample, Fig. 5C). At the *Lithophone* site, a guano corrosion pot (depression formed by acid guano corrosion) is covered by fresh guano and white deposits ("*Lithophone*" sample, Fig. 5D).

# Raganeous Cave (Saint-Benoît, Alpes-de-Haute-Provence, France)

Raganeous Cave opens in a cliff along the Coulomp Valley at 698 m asl in the Southern French Alps. Together with neighboring Radar and Théoriciens caves, they constitute a system of maximal altitude difference of 54 m and of more than 1 km of mainly large galleries (Fig. 6). The cave is developed in a 30-50 m-thick, steeply dipping (30-70°), fossiliferous (Nummulites sp.) limestone bed, resting on Cretaceous marls and partly covered by Upper Eocene (Priabonian) marls. The cave was an important underground drain, which developed as a phreatic lift in a per ascensum way, following the Pliocene aggradation in Coulomp Valley; the current entrances acted as springs (Mocochain et al., 2011). Underground flow carried fine sandy-clayey sediments into the cave, having originated from the weathering of Priabonian marls and Annot sandstones that were brought in through sinkholes. The caves were "discovered" by climbers in 2012. However, artifacts of Neolithic and Middle Age show that the cave was known and used for over millennia by local inhabitants. Currently, a colony of >1000 bats uses the cave for reproduction, roosting, transit, and hibernation. Seven species are present (Rhinolophus ferrumequinum, R. hipposideros, R. euryale, Miniopterus schreibersii, Myotis blythii, M. myotis, and M. emarginatus) (obs. M.-Cl. Lankester).

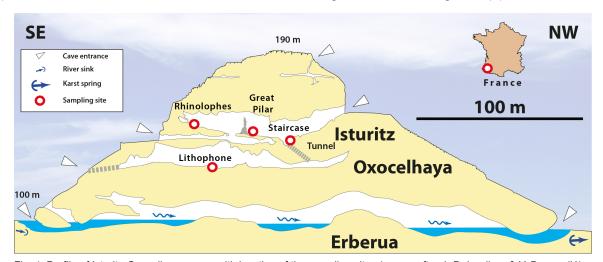


Fig. 4. Profile of Isturitz-Oxocelhaya caves with location of the sampling sites (survey after J.-D. Larribau & Y. Bramoullé). Horizontal passages correspond to successive-lowering river levels, which were utilized by prehistoric people and bats after each passage drained.

Guano deposits are important in different places. Some are still active and occur as soft accumulations of dry pellets, whereas other deposits are older and appear as dark sticky layers. Two sites were sampled (Fig. 6A). A 1.8 m-high cone is made of dry and poorly compacted material, which is composed of loose guano

pellets showing alternating brown-white layers (Raga, Fig. 6B). White and brown layers have been sampled for mineralogy both on top and at the base of the guano cone. Additionally, 10 samples of guano were collected for radiocarbon dating, along with an old wooden torch found close to the bottom of the guano cone. The cave

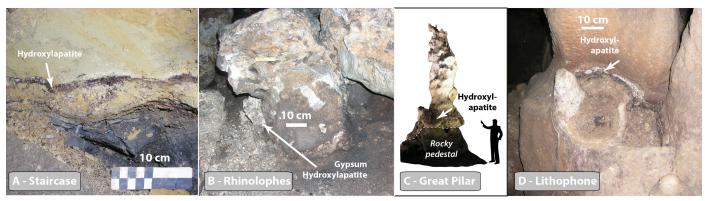


Fig. 5. Studied sites in the Isturitz-Oxocelhaya caves. A) Staircase profile corresponding to the base of the Isturitz Cave filling. Above the rocky floor, layers of yellow and orange stratified fine sands are separated by a dark layer; B) *Rhinolophes* Chamber. Blocks covered with dark crusts and thick detaching white crusts; C) The Great Pilar. Its surface is either bare, light-colored calcite or it is covered with dark deposits; D) *Lithophone* site. Guano corrosion pot showing an aureole of soft white material at the contact with the calcite (Photos by J.-Y. Bigot).

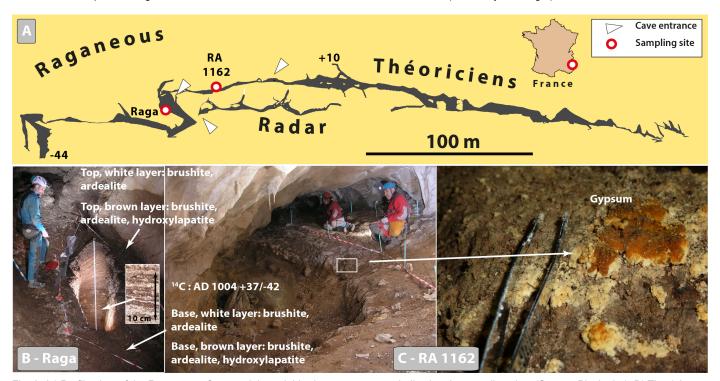


Fig. 6. A.) Profile view of the Raganeous Cave and the neighboring cave systems indicating the sampling sites (Survey: Ph. Audra). B) The 1.8 m guano profile (Raga). It displays a laminated deposit made of alternating brown and white layers (photo by J.-Y. Bigot); C) yellow-orange crust (RA 1162) covering old guano and clay-sand sediments (photos by J.-Y. Bigot and Ph. Audra).

was probably entirely mined for fertilizers at the time when the torch was abandoned; the present cone built up since the mining days. In the main gallery, clay-sand sediments are covered by old guano, which in turn is veneered by a yellow-orange crust (RA 1162) (Fig. 6C).

# Guano Cave (Saint-Cézaire, Alpes-Maritimes, France)

The Guano Cave opens at the foot of the cliff in the Siagne Gorge, 350 m asl, in Alpes-Maritimes, France. It is developed in dolomitic and calcareous Bajocian-Bathonian carbonates. The 280 m-long horizontal looping passage harbors numerous features of extreme corrosion, such as corroded blocks, ceiling cupolas, potholes, and weathered walls. Extensive guano deposits were mined by the owner of the cave. A large bat colony (300 individuals) uses the cave for reproduction. Seven species are present (Rhinolophus ferrumequinum, R. hipposideros, R. euryale, Miniopterus schreibersii, Myotis blythii, M. myotis, and M. capaccinii). The cave is protected for this specific

biotope. About 100 m from the entrance, a narrow passage isolates the inner part, which is characterized by stable climatic conditions (RH 100%, 15°C). Old and fresh guano are present in a lateral chamber (Fig. 7), covering a thick red clay deposit (rich in quartz and muscovite grains), originating from erosion of "terra rossa" at the surface, which then was brought in during an early phreatic phase. We sampled a crust between guano and clay (Gua 1b). The dolomitic walls are highly weathered and covered by a 5 mm-thick white paste (Gua 2).

#### Saint-Marcel Cave (Bidon, Ardèche, France)

Saint-Marcel Cave opens on the left bank of the scenic ~200 m-deep *Ardèche* Canyon that cuts through a Cretaceous limestone plateau covered by Mediterranean scrub ("garrigue"). The cave is 57 km long, including 18 km of underwater passages, has a total relief of 238 m (-101/+137), and consists mainly of large horizontal passages disposed in levels (Brunet et al., 2008; Fig. 8A-B). The cave has drained

the plateau and sinkholes along the Ardèche Gorge since Miocene times (Mocochain et al., 2006, 2011). Temperature in the cave is about 14°C, RH is generally close to saturation, and CO2 varies seasonally from 0.1 to 3-4% in some places. The cave was used since Middle Paleolithic; today part of the cave is opened for tourism. A colony of about 1,000 individuals of Rhinolophus euryale was present in the 1950s. In the Galerie des Chauves-souris ("Bats Passage"), guano on the surface yielded a 14C age of 1916 BC (Dodelin, pers. comm.). The guano accumulations (maximum thickness of 50 cm) extend along this passage, covering a surface of ~15 m2. It occurs as black sticky material, poorly layered, resting on older clay deposits. We sampled pale yellow crystals (SM 2b) on the guano surface (Fig. 8C).

#### Grosse Marguerite Cave (Aiguèze, Gard, France)

Grosse Marguerite Cave is located about 4 km upstream of Saint-Marcel Cave at an altitude of ~200 m asl, on the right bank of the Lower *Ardèche* Gorge. It opens on a ledge below the top of the cliff, about 115 m above the *Ardèche* River. The cave corresponds to an ancient passage of an underground flow, now

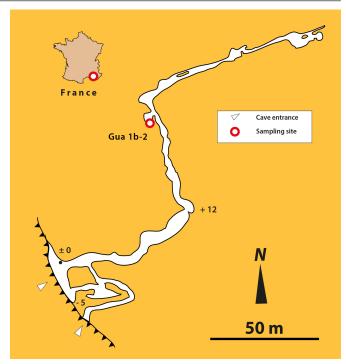


Fig. 7. Plan view of the Guano Cave, located in Provence, France (Survey after Créac'h 1967).

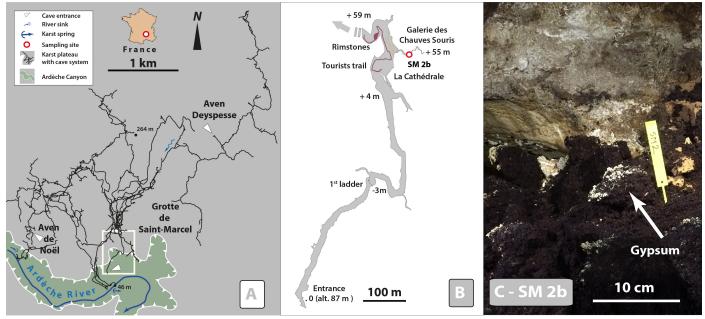


Fig. 8. A-B) Location of the Bats Passage and of SM 2b sampling site in Saint-Marcel Cave (survey after M. Faverjon, SCSM); C) SM 2 sampling site in Saint-Marcel Cave. The guano accumulation is covered by spots of pale-yellow crystals (sample SM 2b) (Photo by A. Chroňáková).

truncated by the cliff, and is clogged by sediments in its inner part. It is developed along ~200 m of rising passages, which split inward in two branches; passage size is about 10 m wide and 10 to 20 m high, with deep cupolas and chimneys where bat colonies gather (Fig. 9A). On the sampling day (15/09/2016), cave air was 15.1°C, with 780 ppm CO<sub>2</sub>, and 95% RH. The studied site is located in the main cave branch, at the foot of the last slope 20 m from the cave end. The guano deposit is about 6 m wide, and lateral extension shows it might be as thick as 3 m. It is covered by dust sediments brought in by the repeated passage of visitors. Below this reworked sediment, the sticky guano is intact and the stratigraphy is reliable.

This guano accumulation was cored up to 60 cm depth and six <sup>14</sup>C dates were obtained (Fig. 9B). It

is composed of 3 main units, which boundaries are located at 13 and 21 cm depth. The upper and lower units are made of dark sticky guano, with 3 to 5 mm-thick lighter brown beds in the lower unit. The central unit is made of a pure white soft crystalline powder (sample GM 2).

### Mescla Cave (Malaussène, France)

Mescla Cave opens in a gorge in the Southern Alps at 184 m asl, 30 km upstream of the mouth of *Var* River in Nice. The cave contains 3.5 km of passage developed along strike, in the limb of an anticline showing strong dip (45°), in Portlandian (upper Jurassic) cherty limestone. The cave displays a looping profile typical of phreatic and epiphreatic flow (Fig. 10, Audra et al., 2002). Upper passages are dry,

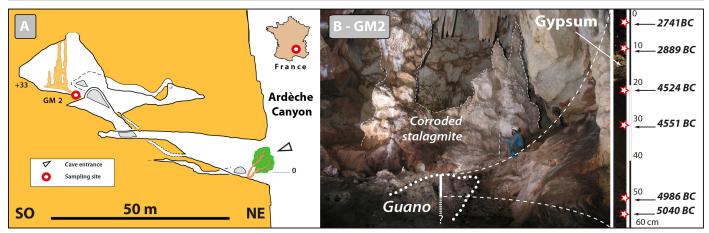


Fig. 9. A) Projected vertical profile of Grosse Marguerite Cave, which opens in the cliff of the *Ardèche* Canyon, France (after J.-Y. Bigot), with indication of sampling site; B) The GM 2 guano core accumulated between 5040 and 2741 BC, however its base was not reached. The guano is covered by dry clay and fire ash reworked along the pathway. A calcite stalagmite was strongly corroded by condensation waters forming above the guano (Photo by J.-Y. Bigot).

whereas the lower level corresponds to the active part of the river outflow. The upstream sump has been explored down to -267 m below the entrance, ranking it among the deepest explored phreatic caves in the world. The deep component of the water is thermal (23°C), with sulfated-chloride water (Reynaud, 2000). Calcite speleothems are present in the first sump, down to -12 m, demonstrating a recent rise of the water table due to sediment aggradation of the nearby

*Var* River. Currently bats and guano deposits are not frequent in the cave. However, the divers identified accumulations of bat bones in the second sump, at about 5 m depth, further confirming the recent rise of the water table. In the first sump, between 3 and 7 m depth, we collected white clay (Mescla 1) preserved in fissures protected from high velocity flow, and a fragment of black crust covering most of the walls (Mescla 2).

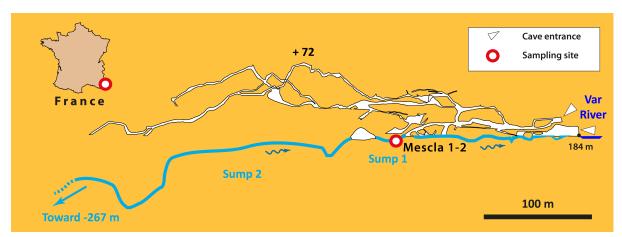


Fig. 10. Projected vertical profile of Mescla Cave showing the sampling sites (Survey after Audra et al., 2002).

## Aramiska Peštera (Dragožel, Macedonia)

Aramiska Peštera opens at 660 m asl. It is a ponor cave located in the upstream parts of *Kamenica* Valley, part of the *Cma Reka* drainage basin, in the southern part of Macedonia (Fig. 11). It is developed in Upper Cretaceous limestone from sinking allogenic waters coming from the Pliocene-Early Pleistocene pyroclastic cover (Temovski, 2016). Back-flooding clay sediments are present throughout most of the cave, especially in the upper (older) parts. In the Room of Bones, remnants of human bones and skulls are covered with flowstone. Here we sampled pyroclastic-derived clay and sand material (AR1, AR2).

#### Karši Podot Cave (Vrpsko, Macedonia)

Karši Podot is a 200 m-long horizontal hypogene cave located in *Crna Reka* Valley. It is formed mainly in Precambrian dolomitic marble, and partly in Pleistocene alluvial and travertine deposits (Fig. 12A).

Cave passages in the dolomitic marble formed by ghost-rock weathering with thermal waters leaving dolomitic sand residue (Temovski, 2013). The explorable passages were later opened due to backflooding from *Crna Reka*, removing the sediment, and redepositing it along with non-carbonate silt and clay coming from the allogenic waters or from the terrace deposits in which the northern parts of the cave are developed. A sediment profile in the Main Room has hydrothermally weathered dolomitic sand with fluvial clay and silt that was covered by a thick (30 cm to 1 m) guano deposit (Fig. 12B). We sampled a brown layer (KP06) from this profile (Fig. 12C).

### Lekovita Voda Cave (Pravednik, Macedonia)

Lekovita Voda Cave is a 90-m long old passage, remnant of a larger system, developed in Upper Cretaceous (Turonian) limestone, and located high in the *Doljani* Valley, a tributary to *Crna Reka*. The cave mostly consists of a single horizontal gallery



Fig. 11. Extended profile of Aramiska Peštera with the sampling location (Temovski, 2016).

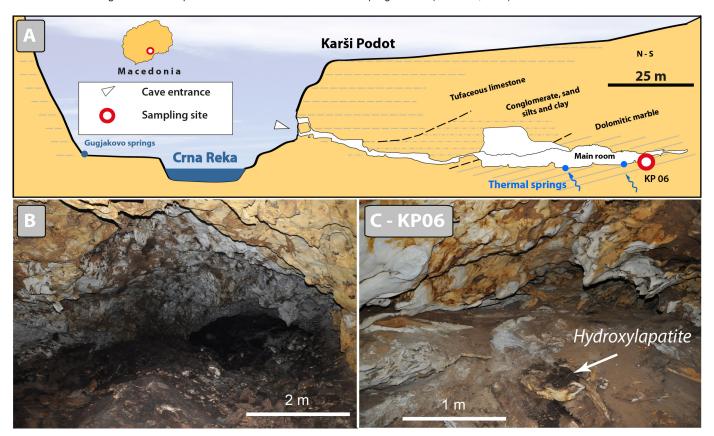


Fig. 12. A) Extended profile of Karši Podot Cave with sampling location (Temovski, 2016); B-C) Main Room in Karši Podot Cave, with detrital and weathered sand covered with thick guano deposits, where sample KP06 was collected (Photo by M. Temovski).

with a paragenetic ceiling morphology (Fig. 13A). It is mostly filled with gravel, sand, and clay sediments, which are partly exhumed and covered with thick flowstone deposits (Fig. 13B). It is presumed that the cave developed in Late Pliocene - Early Pleistocene, in relationship with the local geomorphic events (Temovski, 2016). The cave passage ends with a large sediment fill topped by a thick flowstone deposit, nearby which in a small niche we sampled a detrital yellow clay (LEK1).

### The Monte Inici cave system - Grotta dell'Eremita and Abisso dei Cocci (Castellammare del Golfo, Trapani, NW Sicily, Italy)

Mount Inici is an isolated massif reaching 1064 m asl, close to the NW Sicilian coast. Two caves belong to the *Monte Inici* system: *Grotta dell'Eremita* and *Abisso dei Cocci*, both opening at about 500 m asl on the southern scarp of the mountain (Fig. 14). These

are the two deepest caves in Sicily, with depths of 306 and 361 m (+61/-300), and developments of 2.9 and 2.1 km, respectively (Messana, 1994). They occur in Lower Jurassic limestones and dolomitic limestones (Inici Fm.), and Upper-Middle Jurassic reddish-gray limestones (Buccheri Fm.). The caves are typical 3D mazes with large, gently dipping galleries, and chambers connected by vertical shafts. They are of hypogene origin (Vattano et al., 2013, 2017), related to the nearby thermo-mineral spring with temperatures of up to 50°C (Grassa et al., 2006). Due to the heating of the rock above the thermal aquifer, the caves are warm (17.4 close to the entrances, up to 21.1°C in the lower parts) and dry (70-90% RH); CO<sub>2</sub> may reach 0.4% in the lowest confined areas.

Both caves are rich in phosphate minerals derived from large fossil bat guano deposits and occur especially in the deeper large rooms. Samples were taken in various parts, in small exploration trenches



Fig. 13. A) Profile of Lekovita Voda Cave with sample LEK 1 location (Temovski, 2016); B) View of the sediment choke at the southern end of Lekovita Voda Cave, with clay filled floor covered by thin black soot coating due to frequent visitors (Photo by M. Temovski).

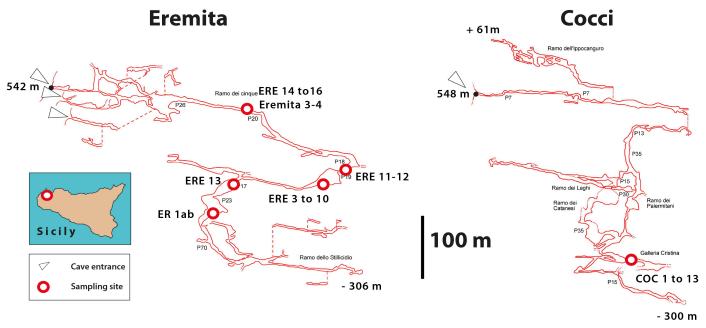


Fig. 14. Vertical profile of *Grotta dell'Eremita* and *Abisso dei Cocci* (after GS CAI Palermo and CSE Catania), with indication of sampling sites and their location within Sicily (inset map).

dug in old bat guano deposits until the underlying limestone was reached. Some broken stalactites found on the floor were also taken.

In *Eremita* Cave we collected 17 samples (Fig. 14-15): a grey-white laminated stalactite (ER1ab), soft laminated guano layers of various colors (white, yellowish, greenish) (ERE 3 and ERE 6 to 15), and a beige earthy part of a black crust (ERE 4-5) at the contact between rock and guano. In *Ramo dei Cinque* Passage, below a cone of old guano, we sampled a light beige soft powdery material in a white layer (ERE15), a white and pinkish earthy material from a yellowish layer (ERE16 and Eremita3), and a bright yellow layer (Eremita4).

In *Abisso dei Cocci* many samples were taken below the Cristina Passage (Fig. 14 and 16): white prismatic translucent crystals (COC1) found underneath boulders, a yellowish to white powder (COC4), a brownish crust covering the many boulders along this passage (COC5), a lemon-yellow colored vitreous layer (COC7), a part of a white layered stalactite showing no reaction with HCl (COC9), and a dark crust (COC10). In a shallow trench dug into the floor sediments a sequence of layers has also been sampled: a lower white layer in contact with the unaltered limestone below (COC11a), a layer with greenish spots inside a white mass (COC11b1) and with pinkish crystals (COC11b2). White microcrystalline powder was recovered in the center of the gallery (COC13).

# Acqua Fitusa Cave (San Giovanni Gemini, Agrigento, Sicily, Italy)

Acqua Fitusa is an inactive water-table sulfuric acid cave (De Waele et al., 2016; Vattano et al., 2017). It is located in Central Sicily (Fig. 17A) along the north-eastern fault scarp of a N-S oriented westwardvergent anticline forming the Mt. La Montagnola. The cave formed in the Upper Cretaceous rudist breccia member of the Crisanti Formation, composed of conglomerates and reworked calcarenites. Chloridesulfate alkaline-earth waters with temperature of 25.2°C (Grassa et al., 2006) still emerge 300 m to the north and at a lower altitude with respect to the cave. The cave consists of three stories of subhorizontal conduits, arranged in a maze pattern following sets of joints, forming large rooms on their intersections; its total length is 700 m and has a vertical range of 25 m (Fig. 17A). Due to the sulfuric acid speleogenesis (SAS), replacement gypsum crusts are present in fissures and pockets, and totally coat narrow blind passages. The cave environment is relatively dry, with a temperature of 12.8°C and RH of 76.5% (Dec. 2011). It hosts a community of Rhinolophus euryale, R. ferrumequinum, R. hipposideros, Myotis myotis, M. capaccinii, and M. schreibersii (Fulco et al., 2015). Samples were taken in the central part of the entrance chamber, where big blocks of collapsed limestone are covered with guano deposits and a dark brown crust has developed (FIT3) (Fig. 17B).



Fig. 15. Phosphates and sulfates sampled in Eremita Cave (see text for further details) (Photos by J. De Waele and Ph. Audra).

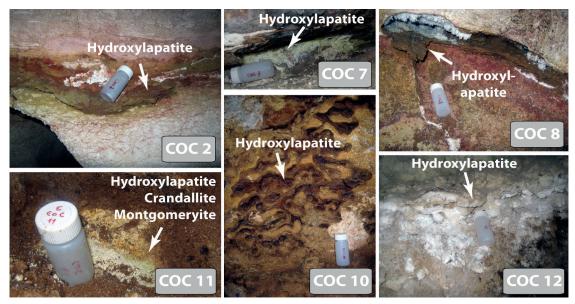


Fig. 16. Phosphates sampled in Cocci Abyss (Photo by J. De Waele).

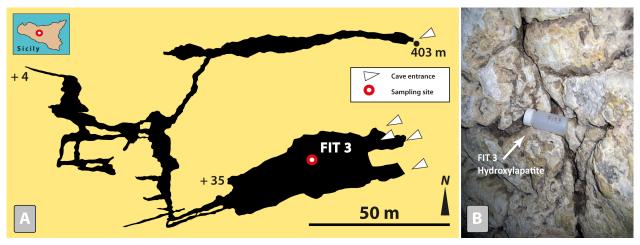


Fig. 17. A) Plan view of Fitusa Cave, with indication of the sampling site (Survey ANS Le Taddarite, Vattano, 2012). In inset, location of the cave in Sicily; B) Dark crust sampled in Fitusa Cave (Photo by J. De Waele).

#### Palombara Cave (Melilli, Syracuse, Sicily, Italy)

Palombara Cave opens at 150 m asl at the foot of the Monti Climition a plateau that inclines toward Syracuse Bay. The plateau is made of Miocene (Burdigalian-Serravalian) calcarenites, namely the Monti Climiti Formation (Lentini & Carbone, 2014). It is part of the Hyblean Plateau, which belongs to the African Plate. Volcanoes were active in this area during the Plio-Pleistocene until 1.4 Ma. The area was continuously uplifted during the Pleistocene as evidenced by several marine terraces and paleocliffs. Palombara Cave is about 800 m long, with a depth of ~80 m (Ruggieri, 2000). It consists of large subhorizontal passages cut by narrow mazes and shafts (Fig. 18A). The entrance is a collapse shaft that opens on the plateau. The cave displays phreatic morphologies, feeders, bubble trails, pointing toward a hypogene origin in relationship with deep CO<sub>2</sub> degassing, possibly brought about by the Hyblean magmatic bodies, and with the uplifting stages recorded by the Pleistocene marine terraces (Vattano et al., 2017). The cave contains fine-clay

sediments, originating from clastic material brought in from the surface. Two samples display reverse paleomagnetic orientations, showing that the cave itself should be older than 780 ka, which would be consistent with the lowest mean uplift rates of 0.2 mm/a proposed by some authors (Dutton et al., 2009, and references therein). After its opening to the surface, the cave was accessible to Neolithic tribes and to bats. The latter have left extensive guano deposits, among which the pile in the Guano Chamber reaches ~8-10 m in height. Five species are currently present (Myotis myotis, Miniopterus schreibersii, Rhinolophus ferrumequinum, R. euryale, and R. mehelyi). The cave is protected by a Regional Law (Di Maggio et al., 2012). Two sites were sampled for phosphate mineralogy. After the main climb and before the Guano Chamber, an old guano deposit (P5) is covered by a colorful crust (Fig. 18B). A little further into the cave, an old guano deposit naturally cut by erosion is covered on a vertical face by a pale violet crust (PA23) (Fig. 18C).

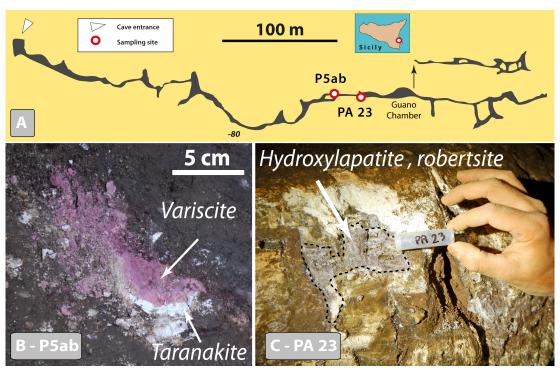


Fig. 18. A) Vertical profile of Palombara Cave (after Ruggieri, 2000), with indication of sampling sites; B) Sample P5 displaying stratigraphic succession of old guano covered by white and purple layers (Photo by M. Vattano); C) Sample PA23; A pale violet crust covers an old guano profile (Photo by D. Cailhol).

### Monello Cave (Syracuse, Sicily, Italy)

Monello Cave opens at 100 m asl in the same area as Palombara Cave, at the foot of the Monti Climiti paleocliff. Also, this cave is preserved as nature reserve (Di Maggio et al., 2012). It consists of ~540 m of galleries developed along faults to a depth of 40 m and well decorated large chambers (Fig. 19A). To the south, a series of shafts lead to the lower point of the cave. As Palombara, numerous feeders, bubble trails, and ceiling channels point toward a hypogenic origin. Some clastic material was introduced by sinkholes and deposited

as decantation clay in the deep parts. They bear normal paleomagnetic orientations, thus younger than 780 ka. Archeological material from Upper Neolithic has been found. Great accumulations of guano are present especially in the first large chamber; however, due to the closure of the cave, only few individuals of *Rhinolophus ferrumequinum* are now present. The lower gallery hosts pieces of collapsed calcarenite rocks, some being covered by a thin layer of brown decantation clay (Fig. 19B). Its surface is hardened by a thin (<1 mm) brown crust (MO 4) (Fig. 19C-D).

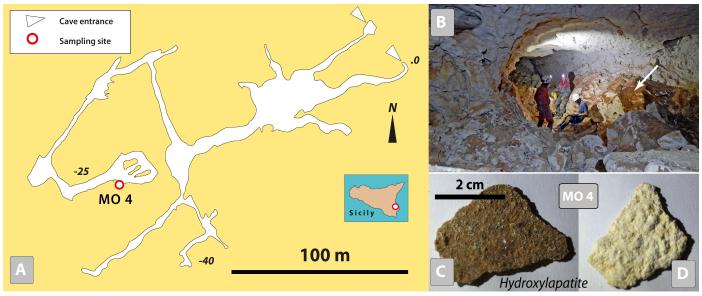


Fig. 19. A) Plan view of Monello Cave (after Ruggieri, 2000), with indication of sampling site; B) Floor of the lower gallery in Monello cave covered by collapsed calcarenite blocks, which in turn are covered by a thin crust of brown decantation clay and phosphate (Photo by D. Cailhol); C) close-up view of the upper part of the crust in B; D. lower side of the crust, at the contact with the calcarenite (Photos by Ph. Audra).

#### Grotta Scrivilleri (Priolo Gargallo, Syracuse, Sicily)

Grotta Scrivilleri, opening at 157 m asl, is an over 2 km-long maze of hypogene origin with most of the passages guided by prominent fractures (Fig. 20A). The cave is developed in the Miocene calcarenites of the Syracuse area, such as the Palombara and Monello caves. Since the cave was just recently opened by digging, no bats are present inside. At the bottom of the *Prima Frattura* Chamber, about

300 m from the entrance, a mixed deposit shows alternation of laminated clay deposited by flooding waters and white soft sandy residue originating from wall weathering by condensation (Fig. 20B). A thin dark-brown layer of presumed old guano (SC4) is present below the upper sandy layer, showing that some connection to the surface through which bats could have entered the cave existed sometime in the past.

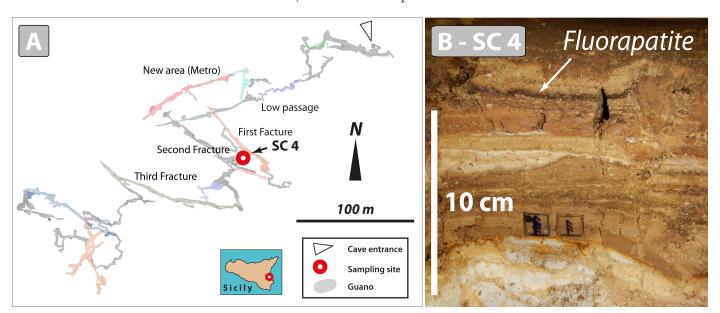


Fig. 20. A) Grotta Scrivilleri plan view, with sampling location (from Centro Speleologico Etneo, Gruppo Speleologico Siracusano, Speleo Club Ibleo); B) Sediment profile from where SC4 sample (a thin dark brown phosphate layer) was collected (Photo by D. Cailhol).

# Personaggi Cave (Montevago, Agrigento, Sicily, Italy)

The *Grotta dei Personaggi* is known since the early 1900s and is famous for the archeological findings that cover a period from the Neolithic to historical times. The cave opens at 350 m asl along a fault scarp in the NW sector of Mt. *Magaggiaro* and is developed in well-bedded white platform limestones (Inici Formation, Lower Jurassic) and in nodular or massive reddish to brown pelagic limestones (Buccheri Fm., Upper-Lower Jurassic) of the *Saccense* Domain (Di Stefano et al., 2013). Thermal springs, characterized by chloride-sulfate alkaline-earth waters with an average temperature of 39.2°C (Grassa et al., 2006) are located about 3 km NW of the cave, which has a total length of ~1.7 km and a relief of 47 m

(+15 m/-32 m). It consists of close-to-horizontal passages that follow bedding and fault planes forming a maze pattern with deep feeders along fractures pointing toward a hypogene origin (Fig. 21; Vattano et al., 2017). The cave morphologies are mainly linked to condensation-corrosion processes by convective airflow, such as upwardly developing cupolas, stacked spheres, weathered walls and boxworks. Alluvial deposits are absent. The cave hosts a large bat colony of Rhinolophus euryale responsible for an extensive amount of guano. A previous study of mineral chemistry shows the presence of phosphates, iron, manganese, and silica spherules, which are interpreted as being concentrated through microbial communities processing the impurities from host rock and clay (Vattano et al., 2015).

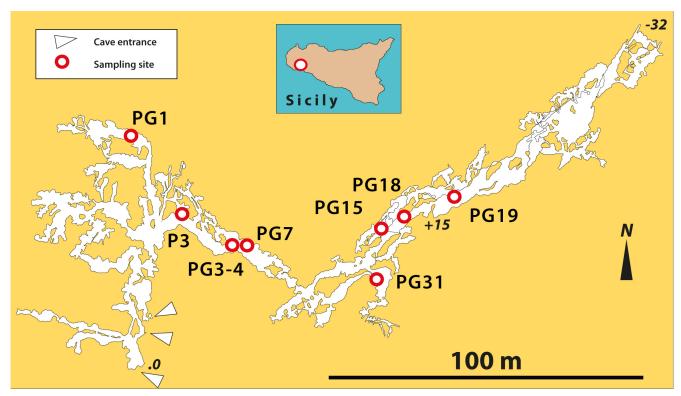


Fig. 21. Plan view of Personaggi Cave (Survey by ANS Le Taddarite, Vattano, 2014), with sample location.

Several soft samples were collected (Fig. 22): the lower black part of a crust in a cupola (PG1), the white part (PG3) and the blackish layer (PG4) of a deposit on a block, a blackish paste between white and reddish parts on the contact between guano and wall (PG7), a whitish paste at the bottom of gallery (PG15), a white-black-yellow crust attached to guano (PG18), a white-black-yellow crust in guano pot (PG19), a black crust on wall (PG31), and a multicolor green-white crust (P3).

# Salnitro Cave (Sambuca di Sicilia, Agrigento, Sicily, Italy)

Salnitro Cave develops in Eocene carbonate breccia resting on the limestones of the Inici Formation. The Mt. *Arancio*, where the cave opens at 252 m asl, is an anticline limb cut by the antecedent canyon of *Tardara*. Salnitro Cave has two entrances and is composed of three, 20 m large and high chambers connected by short passages. Its total development is ~260 m and it has a depth of 18 m (Fig. 23A). The original

shape of the cave voids is hidden by many rock falls, while the floor is covered with debris and large piles of guano (Fig. 23B). The roof is sculptured by cupolas probably related to condensation-corrosion processes above the guano heaps. The cave has been known for a long time, and in the past the guano deposits were mined as fertilizers. Salnitro Cave is home to *Myotis myotis*, *M. capaccinii*, *Miniopteris schreibersii* and only occasionally to *Rhinolophus euryale* and *Pipistrellus kuhlii*. Several samples of dark brown crusts were collected from boulders and cave walls in several areas of the main room (SAL1-4, Fig. 23E). Close to a dripping location (Fig. 23C), a powdery white crust (SN3, Fig. 23D) covered the dark guano deposits at the edge of wet guano, where evaporation occurred.

#### Carburangeli Cave (Carini, Palermo, Sicily)

Carburangeli Cave is located close to the city of Palermo. It is a 400-m long cave (Fig. 24), mainly horizontal, opening at 22 m asl on an elevated marine terrace attributed to MIS 5, at the foot of a paleocliff.

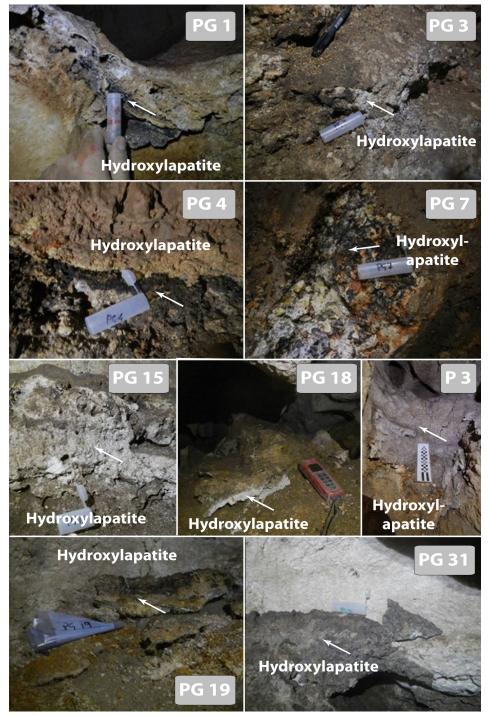


Fig. 22. Phosphates sampled in Personaggi Cave (Photos by M. Vattano).

The entrance part is developed in Pleistocene calcarenites and conglomerates, whereas the inner part extends in Triassic and Lower Jurassic limestones and dolomitic limestones (Madonia et al., 2003). The cave was probably a flank margin cave, transformed into resurgence at a later stage. Carburangeli Cave is famous for its archeological findings of Paleolithic and Neolithic age and is also of paleontological interest (Sicilian dwarf elephant). The cave is protected as natural reserve (Di Maggio et al., 2012), and it hosted an important colony of *Myotis myotis*. Three samples of phosphates were collected in the room where the bats are concentrated (CA1 to 3).

### Pertosa-Auletta Cave (Pertosa, Salerno, Italy)

Pertosa-Auletta is a show cave. It opens on the flanks of the *Tanaro* Valley, along the eastern slopes

of the Alburni Mts., and develops in well-bedded Jurassic micritic limestones (Santo, 1988). The cave is a resurgence with an important underground river in one of the galleries, and some fossil branches (Fig. 25A). The development is little over 3 km. It has two entrances (263 and 268 m asl, respectively), a positive altitude difference of 45 m, and temperatures ranging between 12.2 and 13.9°C. In several areas of the fossil branches old guano deposits are present (Fig. 25B). These were exploited in the 1940s, but parts of these sediments can still be seen along the show cave trail. Dark brown crusts are visible along the lower part of the walls and filling ceiling cupolas (samples P1 to 3). Along the underground river bank, a thick guano deposit is cut and shows an alternation of brown and more yellowish layers (P4, Fig. 25C).

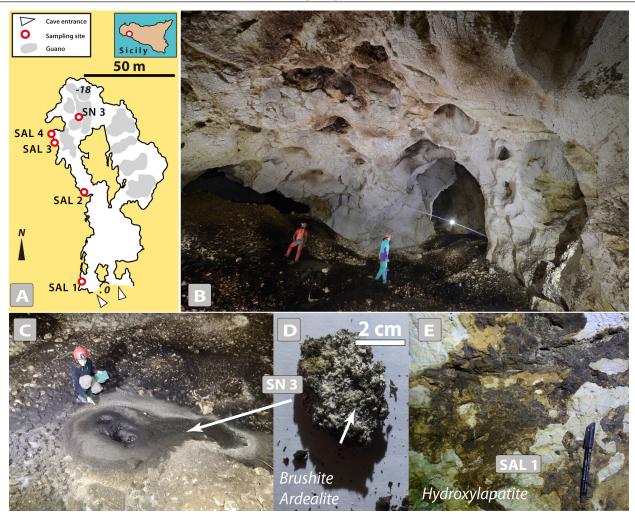


Fig. 23. A) Plan view of Salnitro Cave with sample locations (Survey ANS Le Taddarite, Vattano, 2010, 2015); B) Last chamber where extensive guano deposits cover the blocky floor (Photo by M. Vattano); C) The guano cone is wet in the center because of dripping, whereas its edge, D, is dry, allowing the crystallization of a ring of soft crystals; E) dark brown crust on the wall (Photos by M. Vattano and Ph. Audra).

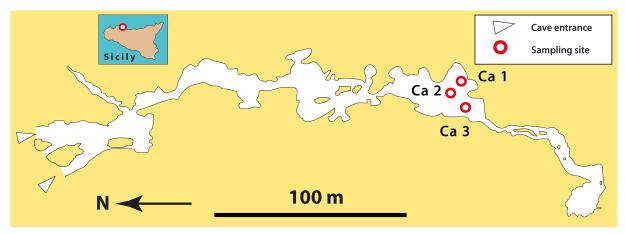


Fig. 24. Plan of Carburangeli Cave (Survey by A. Conigliaro, R. Di Pietro & D. Gueci, 1996-1997), with sampling locations.

### Toirano cave system (Toirano, Savona, Italy)

The Toirano cave system is located in the *Varatella* Valley and develops in dolostones and dolomitic limestones of the *San Pietro ai Monti* Formation (Triassic). From the lower (186 m asl) to the upper level (247 m asl), the caves are organized as follows: *Básura*, *Santa Lucia Inferiore*, *Santuario*, and *Colombo*. Morphological indices suggest this cave system originated from rising, possibly hypogene flow, as a small thermal sulfidic spring surfaces in the village of Toirano. Cosmogenic-nuclide (Al-Be) burial dating of the fluvial gravels found in the upper *Colombo* tier

yielded an age of 2 Ma, showing that the valley was a Messinian canyon eventually filled with thick Pliocene gravels, which were later eroded. Phosphates have been found and sampled only in the highest level of *Colombo*, which is a horizontal 310 m-long cave (Fig. 26A). Here, important bat guano deposits and associated intense condensation-corrosion features show that bats have used this cave for a long time. The cave is of archeological interest, as are most of the caves in the area, and guano deposits have been trenched by scientific excavations. We sampled a hard-yellow crust covering an intact old guano heap (C1, C3) (Fig. 26B).

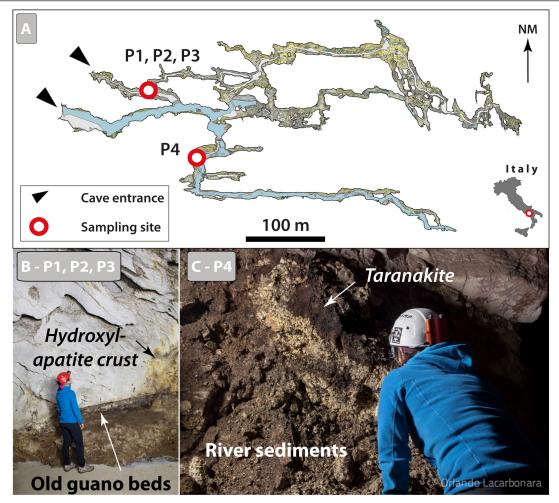


Fig. 25. A) Pertosa–Auletta Cave survey and sampling sites (survey F. Larocca, CRS "Enzo dei Medici"); B) sampling area of P1, P2, P3, with the old guano deposits on the lower part of the cave walls (dark areas); C) sample P4 is located on river sediments (Photos by O. Lacarbonara).

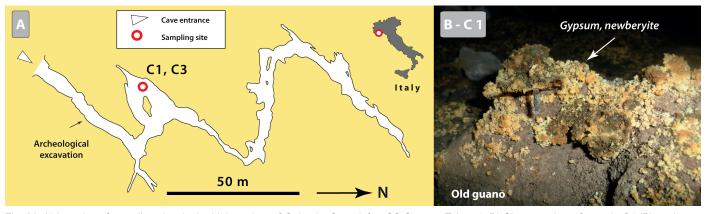


Fig. 26. A) Location of sampling sites in the highest tiers of *Colombo* Cave (after GS Cycnus, Toirano); B) Close-up view of sample C1 (Photo by J. De Waele).

### Corona 'e sa Craba Cave (Carbonia, SW Sardinia)

Corona 'e sa Craba is a 250-long cave entirely carved in a quartzite vein at the contact with Cambrian dolostones (Sauro et al., 2014). The small entrance to the cave has been excavated by local mineral collectors and opens at an altitude of 260 m asl (Fig. 27A). In the Corona 'e sa Craba area there are several mines for Pb, Zn, and Ba ores, which were active until the early 1980s. The cave consists of a series of chambers and lacks the typical carbonate speleothems. After the first room, the cave continues with a wide and high gallery that hosts a large guano deposit generated by an important bat colony inhabiting the cave during spring-early summer. The rock walls are covered

with powdery material and crusts of colors ranging from reddish to violet, brown, greyish, white, pale pink, blue, and black. These secondary minerals are mainly composed of oxides-hydroxides, sulfates, and phosphates. The sulfates are derived from the acid corrosion by sulfuric acid in a late stage of cave development (Sauro et al., 2014). Phosphates, on the other hand, derive from the interaction of the host rock and its minerals with fresh or old guano. Besides the common hydroxylapatite and taranakite, the rare minerals robertsite, spheniscidite (Fig. 27B), vashegyite, strengite, and possibly leucophosphite and berlinite were reported by Sauro et al. (2014).

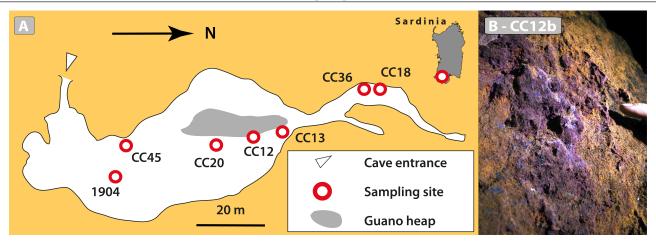


Fig. 27. A) Plan of the Corona 'e Sa Craba Cave (survey after GRSEAM caving Club, Carbonia); B) Purple-violet crusts (CC12b) with small blue fragments (Photo by L. Sanna)

# RESULTS: GUANO-DERIVED CAVE MINERALS

The mineralogical results for the sampling sites on the figures of the previous section are reported in <u>Supplementary Table</u>. In the below reported Table 1 these results are summarized, reporting a total of 17 phosphates found in the investigated caves; these are shortly described below. Together with these phosphates, an almost ubiquitous secondary mineral produced by guano decomposition is gypsum, a sulfate. These minerals were identified by XRD analysis, with the following exceptions: Raman spectroscopy for Isturitz Cave (all samples), Raganeous Cave (RBB), and Palombara Cave (PA23); in Salnitro they were confirmed using EDAX.

Table 1. List of the 18 minerals (17 phosphates and 1 sulfate) found in the investigated caves. Chemical formulae are according to the IMA-CNMNC list of mineral names (<a href="http://nrmima.nrm.se//imalist.htm">http://nrmima.nrm.se//imalist.htm</a>). For more details on the sampled that delivered these minerals in the different caves see <a href="Supplementary Table">Supplementary Table</a>.

Mineral name	Chemical formula	Cave					
Ardealite	Ca <sub>2</sub> (HPO <sub>4</sub> )(SO <sub>4</sub> )·4H <sub>2</sub> O	DB, RC, SN, TC					
Berlinite	AlPO <sub>4</sub>	СС					
Brushite	Ca(HPO <sub>4</sub> )·2H <sub>2</sub> O	DB, RC, SN, TC					
Crandallite	CaAl <sub>3</sub> (PO <sub>4</sub> )(PO <sub>3</sub> OH)(OH) <sub>6</sub>	GC, MI, CA					
Fluorapatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> F	DB, MC, AP, KP, LV, SC, PE, PR					
Hydroxylapatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> (OH)	DB, IO, RC, GC, MC, MI, AF, PA, MO, PE, SN, CA, PR, CC					
Leucophosphite	KFe <sup>3+</sup> <sub>2</sub> (PO <sub>4</sub> ) <sub>2</sub> (OH)·2H <sub>2</sub> O	CC					
Montgomeryite	Ca <sub>4</sub> MgAl <sub>4</sub> (PO <sub>4</sub> ) <sub>6</sub> (OH) <sub>4</sub> ·12H <sub>2</sub> O	MI					
Newberyite	Mg(HPO <sub>4</sub> )·3H <sub>2</sub> O	TC					
Robertsite	$Ca_2Mn^{3+}_3(PO_4)_3O_2\cdot 3H_2O$	PA, CC					
Spheniscidite	(NH <sub>4</sub> ,K)(Fe <sup>3+</sup> ,Al) <sub>2</sub> (PO <sub>4</sub> ) <sub>2</sub> (OH)·2H <sub>2</sub> O	DB, JC, CC					
Strengite	Fe <sup>3+</sup> (PO <sub>4</sub> )·2H <sub>2</sub> O	JC, CC					
Taranakite	(K,NH <sub>4</sub> )Al <sub>3</sub> (PO <sub>4</sub> ) <sub>3</sub> (OH)·9H <sub>2</sub> O	DB, MI, PA, PR, CC					
Variscite	AlPO <sub>4</sub> ·2H <sub>2</sub> O	MI, PA					
Vashegyite	Al <sub>11</sub> (PO <sub>4</sub> ) <sub>9</sub> (OH) <sub>6</sub> ·38H <sub>2</sub> O	СС					
Vivianite	Fe <sup>2+</sup> <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> ·8H <sub>2</sub> O	GC					
Whitlockite	Ca <sub>9</sub> Mg(PO <sub>4</sub> ) <sub>6</sub> (HPO <sub>4</sub> )	GC					
Gypsum	Ca <sub>2</sub> SO <sub>4</sub> ·2H <sub>2</sub> O	DB, IO, RC, SM, GM, MI, SN, TC					

DB - Baradla-Domica, JC - Julio, IO - Isturitz-Oxocelhaya, RC - Raganeous, GC - Guano, SM - Saint Marcel, GM - Grosse Marguerite, MC - Mescla, AP - Aramiska, KP - Karsi Podot, LV - Lekovita Voda, MI - Monte Inici, AF - Acqua Fitusa, PA - Palombara, MO - Monello, SC - Scrivilleri, PE - Personaggi, SN - Salnitro, CA - Carburangeli, PR - Pertosa-Auletta, TC - Toirano, CC - Corona 'e sa Craba. Italics indicate an uncertain presence of the mineral.

#### Phosphate minerals

**Ardealite** displays as white to yellowish, soft powder (Hill & Forti, 1997). In Domica Cave, it is associated with gypsum, brushite, hydroxylapatite, and taranakite (Fig. 2A), as also identified by Kereskényi (2014) in this cave system. In Salnitro Cave it occurs together with brushite and gypsum (Fig. 23D), whereas at Toirano (Colombo Cave) it has been found with newberyite, brushite, and gypsum. In Raganeous Cave, all brown and white alternating layers are

made of hydroxylapatite, brushite, and ardealite, but gypsum is also present (Fig. 6B). Further micro-Raman spectroscopy confirmed ardealite to be the major phase (Fig. 28). Since gypsum and brushite are isostructural, ardealite originating from an additional sulfate-ion exchange is frequently associated with these two minerals (Onac, 2012). This sulfate-calcium phosphate corresponds to the early stage of guano breakdown from the reaction of sulfuric and phosphoric acids with limestone (Hill & Forti, 1997; Puşcaş et al., 2014). It

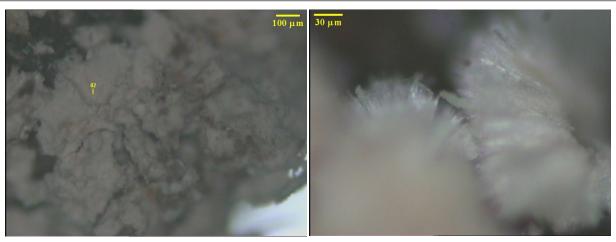


Fig. 28. Micro-photography of ardealite from Raganeous Cave (sample RBB).

is stable in slightly dry conditions and in an acidic environment, at pH of 5.7 to 6.6 (Puşcaş et al., 2014), or even less than 5.5, providing sulfur is available (Marincea et al., 2004a). Ardealite may form during successive stages involving hydroxylapatite ⇒ brushite ⇒ ardealite ⇒ gypsum, this last mineral depositing on the surface (Onac & Vereş, 2003; Marincea et al., 2004a; Onac et al., 2005a; Pogson et al., 2011).

The presence of **Berlinite** is suspected in Corona 'e Sa Craba Cave based on XRD results. Berlinite is a very rare anhydrous phosphate that requires rather high temperatures to form (over 550°C; Muraoka & Kihara, 1997; Onac & White, 2003; Onac & Effenberger, 2007). In this cave, its presence is very likely, but further investigations are needed to understand its genesis.

**Brushite** occurs as ivory-yellow, soft, powdery nodules (Hill & Forti, 1997). In Salnitro Cave it occurs together with ardealite and gypsum (Fig. 23D); in Colombo Cave (Toirano) with newberyite, ardealite, and gypsum (Fig. 26B); in Raganeous Cave, with hydroxylapatite, ardealite, and some gypsum (Fig. 6B); in Baradla Cave with fluorapatite, taranakite, and gypsum. In Domica Cave, it is associated with ardealite and hydroxylapatite (Fig. 2A, 2C), where brushite occurs at the base of the guano pile, together with dominant gypsum and apatite. However, it mainly occurs as crusts at some distance below guano leachates, associated with calcite rims deposited later because of Ca oversaturation (Kettner, 1948). Brushite is often

associated with gypsum, as a crust below gypsum in guano pots. Brushite, generally associated with ardealite and gypsum, is mainly present on the surface of guano piles and at the base close to the limestone contact, or as crust below guano leachates (Kettner, 1948). Brushite is a common cave mineral, stable in acidic (pH < 6) and damp conditions (Hill & Forti, 1997), with nucleation limits comprised between pH 5.5 and 8 (Onac et al., 2005b). Brushite and hydroxylapatite alternatively precipitate from the same solution, with a nucleation boundary at pH 6.2-6.8, with brushite being the stable species in more acidic conditions (Puşcaş et al., 2014). It forms after reaction of phosphoric acid with limestone rock, where ammonium phosphate derives from guano breakdown, according to the following reaction (after Frost & Palmer, 2011):

$$2NH_3 + H_3PO_4 \rightarrow (NH_4)_2(PO_3OH)$$
 
$$(NH_4)_2(PO_3OH) + Ca^{2+} + 2H_2O \rightarrow Ca(PO_3OH) \cdot 2H_2O + 2NH_4^+$$

**Crandallite** was identified only in three caves. In Cocci, it is located between old guano and weathered limestone, associated with montgomeryite and hydroxylapatite (Fig. 16, Fig. 29A). It occurs as light greenish spherules in a white-yellowish-greenish material. In Guano Cave, it is located between guano and clay, and is also associated with hydroxylapatite. It was also found in Carburangeli Cave. Crandallite forms after neutralization of phosphoric acid solutions at the contact with limestone that provides Ca, whereas Al comes from clay (Hill & Forti, 1997).

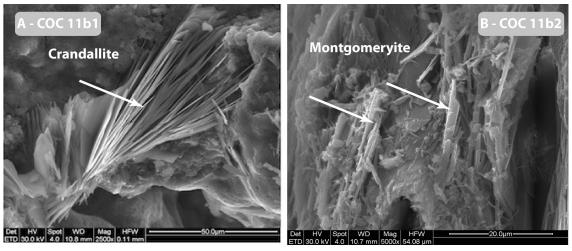


Fig. 29. SEM pictures of A) Crandallite and B) Montgomeryite from Cocci Cave (Photo by E. Galli).

**Fluorapatite** is found in Baradla Cave (Fig. 2C) in a yellow layer associated with brushite and quartz, in a crust now underwater in Mescla Cave (France), and in the caves of Macedonia region (Aramiska Peštera, Karši Podot (Fig. 12), and Lekovita Voda). A carbonate-rich fluorapatite is found in Scrivilleri (Fig. 20B) and Cocci caves, Italy, as a dark brown layer associated with detrital material (quartz, calcite, goethite, kaolinite).

Fluorapatite derives from hydroxylapatite after the replacement of the hydroxyl ion by fluorine, which may be provided by the host rock (Hill & Forti, 1997), but in our caves, it likely comes from paleontological remains (bones or teeth).

**Hydroxylapatite** exhibits various colors (brown, white, yellow, cream, etc.) and textures (crust, moonmilk paste, etc.). It is associated with brushite and ardealite in Domica Cave, with calcite in the "urine crust" covering ceiling cupolas of Baradla Cave and with gypsum in Cocci Cave (Fig. 16), as a stalactite in Eremita Cave (Fig. 15), and as crust associated to minor robertsite in Palombara Cave (Fig. 18C, Fig. 30). Hydroxylapatite is the most common phosphate mineral due to the neutralization of the acidic solution at the contact with limestone. It is stable at pH 6.2 - 6.6 (Hill & Forti, 1997) and indicates a low Mg environment since this element inhibits the precipitation of hydroxylapatite (Chang et al., 2010).

Hydroxylapatite is often difficult to distinguish through XRD analysis from fluorapatite. In Isturitz Cave, hydroxylapatite was identified on all four samples using Raman spectroscopy. The sample from Great Pillar Cave provided spectra with the characteristic bands of calcite (at 1,085, 711, 285, and 154 cm<sup>-1</sup>) and of hydroxylapatite at 960 cm<sup>-1</sup>, slightly distorted and wider than normal, indicating the presence of other phosphates (Fig. 31). EDXRF shows an increase of phosphorus concentration from the calcite host rock (3 mg/g) toward the phosphatized zones (40 to 120 mg/g). The highest concentration was found in the white weathered ring covering the inner part of the guano pot located in *Lithophone* site (Prieto et al., 2015).

**Leucophosphite** was identified only from Corona 'e sa Craba Cave. The mineral derives from interaction of bat guano solutions with iron-rich materials, such as

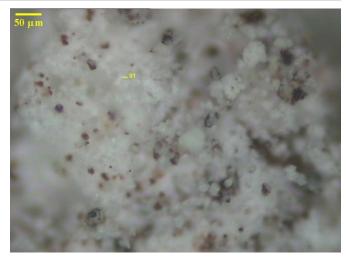


Fig. 30. Micro-photography of hydroxylapatite in Palombara Cave (PA23).

volcanic clays. It occurs together with its close relative spheniscidite, from which it is difficult to differentiate. The presence of leucophosphite was confirmed by SEM-EDS analysis showing the presence of potassium (Fig. 32).

**Montgomeryite** is present only in Cocci Cave, between old guano and weathered limestone, associated with crandallite and hydroxylapatite (Figs. 16, 29B). It appears as light pinkish aggregates in a white-yellowish-greenish material derived from reaction between guano and the host rock. Similar to crandallite, montgomeryite forms after neutralization of acidic solution at the contact with limestone, which supplies Ca and Mg, whereas Al comes from clay. Due to Ca and Mg leaching, it is metastable and converts into crandallite and eventually to variscite (Hill & Forti, 1997).

Newberyite has been discovered in Grotta del Colombo, part of the Toirano cave complex (Liguria). It was found together with gypsum as a hard-yellowish crust on old guano heaps in a large entrance hall in open communication with the outside atmosphere (Fig. 26B). This mineral is more typical of caves of a warm and dry climate (Hill & Forti, 1997). However, in Toirano, the genesis of this mineral likely comes from the interaction of phosphates with Mg provided by the disaggregation of dolomitic host rock or by seepage from this rock, whereas air exchanges with the atmosphere provide seasonal dry conditions.

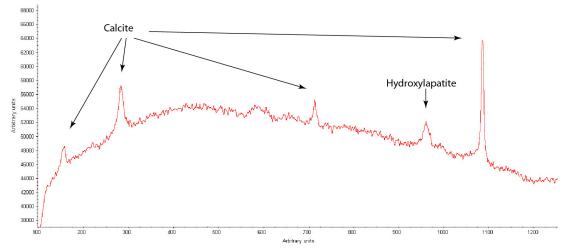


Fig. 31. Raman spectra from Isturitz Cave, France (Prieto et al., 2015).

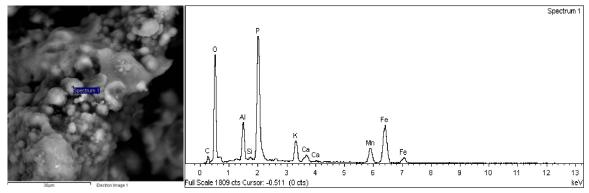


Fig. 32. SEM-EDS spectra of spheniscidite in Corona 'e sa Craba cave. Potassium (K) indicates the presence of a minor leucophosphite phase (Baldoni et al., 2013).

Robertsite is a calcium-manganese phosphate deriving from the reaction with host rock that provides the metallic ions. Robertsite is generally associated to the weathering of pegmatitic deposits. Its occurrence in caves is very rare, only reported in Puerto Princesa Underground River, Philippines (De Vivo et al., 2013), and in Corona 'e sa Craba Cave, Italy (Sauro et al., 2014). In Palombara Cave, robertsite is a minor compound associated with hydroxylapatite (Fig. 18C). Mn is likely related to rising fluids, since Palombara is a hypogene thermal cave. The presence of Pararobertsite was suspected from XRD. Further micro-Raman spectroscopy showed that manganese is indeed present in oxides (jacobsite or todorokite), but the Mn phosphate pararobertsite could not be confirmed because of the difficulty to distinguish its spectra from robertsite and hydroxylapatite.

Spheniscidite is found in Baradla Cave, in a white laminated layer near the bottom of a guano accumulation, close to the contact with limestone and clay substratum (Fig. 2C). It is associated with taranakite, quartz, and illite. In Corona 'e sa Craba Cave, small crystals of spheniscidite (10 µm) are forming purple-violet crusts that cover the walls close to fresh guano deposits (Fig. 27B). In Julio Cave, it crops out along with strengite as thick grey deposits derived from the weathering of fluvial silts. The chemical formula of spheniscidite is very close to that of leucophosphite (the latter containing Al and NH<sub>4</sub>). The breakdown of bat urea first produces gaseous ammonia (NH3), which in turn produces in the presence of water, the very soluble ammonium ion (NH<sub>4</sub>+). Spheniscidite forms from the reaction of ammonium-rich fresh guano leachates with clay sediments containing Fe, namely illite-group minerals. These three occurrences of spheniscidite are new additions to the first mention reported from a Sardinian cave (Sauro et al., 2014). This mineral is relatively soluble; therefore, it only occurs under conditions allowing its preservation as crusts in a dry environment (Corona 'e sa Craba Cave) or in places where drainage is low (Baradla, Julio).

**Strengite** has been found in Corona 'e Sa Craba and Julio caves. It derives from the interaction between acidic guano leachates and detrital material (clay, fluvial silts) or host rock providing Fe. A partial substitution of Fe by Al produces Al-rich strengite, a rare occurrence, which is not recognized by the IMA as a bona fide mineral (Onac et al., 2005a).

**Taranakite** appears as a soft flour-like white powder, and as nodules in clay (Hill & Forti, 1997). It occurs with calcium phosphates (brushite and ardealite), with gypsum in Domica (Fig. 2A, C) and Eremita caves (Fig. 15), with spheniscidite in Baradla Cave, and is always associated with clay material: illite in Baradla Cave; kaolinite in Palombara Cave (Fig. 18B); river sediments in Pertosa-Auletta cave system (Fig. 25C); kaolinite and montmorillonite in Eremita Cave. Taranakite forms in damp (Hill & Forti, 1997) and slightly acidic conditions at pH 5.7 – 6.1 (Onac, 1996; Puşcaş et al., 2014) from the reaction of guanorelated phosphate solutions with clay substratum containing illite or kaolinite that provide K<sup>+</sup> and Al<sup>3+</sup>.

In Palombara Cave (Sicily), **Variscite** occurs as an intense purple layer sandwiched below old guano and above white taranakite; this last mineral rests on clay substratum that contains quartz and kaolinite (Fig. 18B). In Eremita Cave, variscite is associated with gypsum and occurs as a white layer above a yellow accumulation of taranakite and gypsum, with the latter one resting on clay substratum that contains montmorillonite and kaolinite (Fig. 15). Variscite forms from guano-related phosphate leachates reacting with clays such as kaolinite that provide Al<sup>3+</sup>, and where K<sup>+</sup> and Fe<sup>2+</sup> are not available.

**Vashegyite** is present in Corona 'e Sa Craba Cave. It is a rare cave mineral resulting from the interaction of bat guano solutions with clays providing Al, in slightly acidic and permanent wet conditions (Forti et al., 2000; Onac et al., 2006b).

**Vivianite** is suspected to be present in Guano Cave. It is a rare cave mineral, mentioned first in Niah Great Cave, Malaysia (Bridge & Robinson, 1983), then in Megali Grava Cave, Greece (Theodorou et al., 2004) and Grotta Castellana, Italy (Hill & Forti, 1997). There, vivianite derives from the reaction of guano with "terra rossa" in a strong reducing environment, due to presence of coal and organic matter. We speculate similar conditions in Guano Cave.

Whitlockite is a rare mineral first described in a cave at Sebdou, Algeria (Bannister & Bennett, 1947) and in El Capote Cave, Mexico (Peréz Martínez & Wiggen, 1953). It constitutes various speleothems as isolated aggregates or earthy material (CAMIDA, 2018). We found whitlockite only in Guano Cave, on the weathered layer of a dolomitic wall. It is associated with hydroxylapatite and possibly vivianite (see above), together with muscovite, quartz, and birnessite (Mn

oxide of Ca-Na). The three last minerals are partly or entirely allogenic. Whitlockite derives from interaction of bat guano solutions with carbonate walls providing Ca and Mg. Alternatively, it could form from the dissolution of apatite and subsequent precipitation in the presence of low Ca concentration solutions (Peréz Martínez & Wiggen, 1953; Bridge & Robinson, 1983; Onac et al., 2009).

#### **Sulfates**

Gypsum is a common sulfate cave mineral, often related to bat guano deposits, the oxidation of pyrite from bedrock or sediments, or to sulfuric acid speleogenesis (De Waele et al., 2016). In the case of bat guano deposits, it occurs as crystalline crusts due to evaporation on the surface, or as interbedded lenses, pure or associated with other minerals. It forms from sulfuric acid solutions originating from the decaying guano that reacts with drippings providing carbonates in solution. As crystalline crust, it is present in yellow crystalline masses on old guano in Saint-Marcel Cave (Fig. 8C) and Raganeous Cave (Fig. 6C), in Colombo Cave associated with newberyite (Fig. 26B), in a black crust on a wall in Salnitro Cave associated to hydroxylapatite, in Eremita Cave as soft microcrystalline deposits on the surface of guano (Fig. 15), in Cocci Cave as crusts or transparent crystals. The thick white crust covering limestone blocks nearby old guano in *Rhinolophes* Chamber (Isturitz Cave; Fig. 5B), provided characteristic Raman spectra of gypsum. EDXRF shows that phosphorus concentration is here limited to 14 mg/g, whereas it reaches 40-100 mg/g in the uppermost dark phosphate crust directly exposed to bat drops (Prieto et al., 2015).

The second type of gypsum occurrence is that of layers of various thickness and purity. In Raganeous, Baradla, and Domica caves, as white paste, alone (Ba 4d), or associated with ardealite and brushite (Pr. B, RBB) (Fig. 2A). In Eremita (Eremita 3, 4, and ERE 14; Fig. 15) and Grosse Marguerite (Fig. 9B), pure gypsum constitutes a layer of soft yellow or compact white material, respectively.

#### Radiocarbon dating of guano piles

Thirty-two radiocarbon dates were performed on six guano piles, with 4 of them carried out along a vertical profile (Table. 2). Raganeous Cave shows a continuous deposit of about 1000 years. The oldest age at AD 1004, corresponds to the depth 1.72 m. The sample taken at 0.48 m deposited at AD 1843. The oldest cone was probably entirely mined for fertilizer around AD 1349 (according to the age of the wood torch found at a depth of 1.5 m), whereas the present cone has formed since. In Domica Cave, Prales guano pile is still actively accumulating, as compared to pictures taken about 65 years ago (Kettner, 1948). The two cores reaching the bottom date back to AD 1351 and AD 1102 respectively. Palmový Háj has a similar age at 1 m depth (AD 990; Krištůfek et al., 2008), whereas most of the accumulation between 60 cm and the subsurface is bracketed between AD 1654 and AD 1831. In Saint-Marcel, the date (1916 BC) corresponds to the surface of the guano accumulation, whereas in Baradla (281 BC) it corresponds to the bottom. For Grosse Marguerite, radiocarbon dates yielded calibrated ages comprised between 5040 BC and 2741 BC. The central white unit, made of pure gypsum, is framed between 4524 BC and 2889 BC.

Table 2. Radiocarbon dating of guano, including previous studies in Saint-Marcel (Dodelin, unpubl.) and Domica G1-5 samples (Krištůfek, 2008).

Cave	Site	Lab.	Lab. no.	Sample no.	Dist. from surface (cm)	Conventional radiocarbon age (AMS)	Median Probability Cal year AD/BC	cal AD/BC age ranges		Probability distribution
Raganeous	Guano hill (base of shaft)	Poznań Radiocarbon Laboratory	Poz-72000	RAG- C3-S4	48,5	85 ± 30 BP	1843	1809	1926	0.734
			Poz-59140	RAG- C3-S3	76,5	200 ± 30 BP	1771	1730	1809	0.567
			Poz-71999	RAG- C3-S3	79.5	175 ± 30 BP	1770	1725	1815	0.556
			Poz-59141	RAG- C3-S2	72,5	140 ± 30 BP	1807	1798	1891	0.383
			Poz-71998	RAG- C3-S2	99,5	180 ± 30 BP	1770	1726	1814	0.569
			Poz-71997	RAG- C3-S2	129,5	290 ± 30 BP	1566	1493	1601	0.675
			Poz-59142	RAG- C3-S1	125,5	260 ± 25 BP	1647	1630	1668	0.679
			Poz-72060	RAG- C3-S1	149,5	305 ± 30 BP	1562	1488	1603	0.745
			Poz-62932	Wood torch	158.5	620 ± 30 BP	1349	1292	1399	1.000
			Poz-59143	RAG- C3-S1	169,5	1015 ± 25 BP	1013	978	1042	0.981
			Poz-72061	RAG- C3-S1	171,5	1030 ± 30 BP	1004	962	1041	0.965
Saint-Marcel	Bat gallery	Centre de datation radiocarbone (CDRC), Lyon	Ly-16803	SM1	Surface	3565 ± 30 BP	1916 BC	1982 BC	1872 BC	0.834

Baradla	Libanon- Hegy	ARTEMIS (LMC14, Saclay)	SAC_45654	BA4 3-4	≈ 50	2205 ± 40 BP	281 BC	379 BC	177 BC	1.000
Domica	Palmový Háj. Guano hill	Poznań RL	Poz-18867	G1	0-3	109.51 ± 0.35 pMC	Modern (Post to 1957 AD)	> 1957 AD		
			Poz-18868	G2	0-3	120 ± 30 BP	1831	1801	1939	0.659
			Poz-18869	G4	0-3	135 ± 30 BP	1813	1799	1892	0.414
			Poz-18870	G3	60-65	250 ± 30 BP	1654	1626	1679	0.596
			Poz-18871	G5	100-105	1055 ± 30 BP	990	945	1024	0.883
	Palmový Háj. Guano hill	ARTEMIS	SAC_45652	PH 56-57	18	185 ± 30 BP	AD 1770	1726	1813	0,578
			SAC_45653	PH 20-21	54	155 ± 30 BP	AD 1776	1718	1784	0,359
		Poznań RL	Poz#2- 79326	PH 1-2	73	250 ± 27 BP	AD 1654	1631	1675	0,658
	Prales Chamber.	ARTEMIS	SAC_45651	PR1 4-6	35	915 ± 30 BP	AD 1102	1030	1188	1.000
G	Guano	no Poznań RL	Poz-79327	PR2 4-6	59	625 ± 26 BP	AD 1351	1337	1398	0,606
	Guano hill (top of gallery)	ARTEMIS	SAC_46850	GM2 3-4	3,5	4145 ± 30 BP	2741 BC	2825 BC	2625 BC	0.805
Grosse Marguerite			SAC_46851	GM2 10- 11	10,5	4260 ± 40 BP	2889 BC	2937 BC	2854 BC	0.754
			SAC_45658	GM2 22- 23	22,5	5690 ± 40 BP	4524 BC	4619 BC	4450 BC	0.956
			SAC_45657	GM2 32- 33	32,5	5710 ± 45 BP	4551 BC	4625 BC	4458 BC	0.860
			SAC_45656	GM2 52- 53	52,5	6110 ± 40 BP	4986 BC	5068 BC	4892 BC	0.936
			SAC_45655	GM2 58- 59	58,5	6075 ± 35 BP	5040 BC	5208 BC	4942 BC	1.000
		ARTEMIS	SAC_46849	GM 1 3-4	3,5	4130 ± 30 BP	2725 BC	2780 BC	2617 BC	0.664
			SAC_45660	GM1 13- 14	13,5	4230 ± 60 BP	2795 BC	2932 BC	2620 BC	0.989
			SAC_45659	GM1 31- 32	31,5	5750 ± 40 BP	4601 BC	4701 BC	4500 BC	1.000

#### Acidity of the decaying guano

We measured two profiles along guano accumulations in Domica Cave in Prales (PR2) and  $Palmov\acute{y}$   $H\acute{a}j$  (PH) (Fig. 33). In Prales the guano is still actively accumulating; its basal layer was dated at about AD 1351. The pH profile shows a continuous decrease, from 4.3 to 3.7, with a slight increase at the very bottom.  $Palmov\acute{y}$   $H\acute{a}j$ , on the contrary, is not active anymore, showing an oscillation of pH in a narrow range between 3.9 and 3.5. No trend associated to depth-age is visible.

#### **DISCUSSION**

#### Guano-related mineral genesis

The presence of seventeen different phosphate minerals and gypsum in the twenty-two investigated cave systems is related to the presence of different physico-chemical conditions and the availability of different elements in the host rock (Fig. 34). Guano can slowly transform by itself, but its leachates also react with the bedrock and minerals hosted therein (e.g., sulfides such as pyrite), clays, and airborne material. The processes leading to the formation and precipitation of these minerals are: acid digestion, microbially-driven organic decay, self-combustion (since guano decay is an exothermic reaction), and evaporation.

In limestone caves and in acidic, wet and fresh guano deposits, the typical mineral association is composed of brushite, ardealite, and gypsum. These

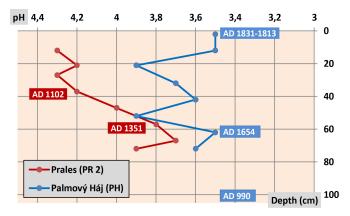


Fig. 33. pH decreases along the PR2 and PH cores in the guano hill of Prales Chamber and  $Palmov\acute{y}$   $H\acute{a}j$ , Domica Cave, with indication of calibrated  $^{14}$ C ages.

are rather soluble minerals which often do not persist over long periods of times (as guano decomposes), and as such they are found only in rather fresh deposits. Such conditions exist in Salnitro and Toirano caves, and in Saint-Marcel and Raganeous caves. In this last cave, the low maturation of guano is interpreted to be the result of permanent airflow between the entrances that maintain dry conditions, which slows down mineralization. Often, hydroxylapatite also forms under similar conditions, with fluorapatite occurring where guano is in contact with skeleton remains, such as in the three Macedonian caves. These two minerals, which are typical for more neutral condition, due to the buffering effect of the limestone, are the most abundant phosphates found

in guano pots, or as crusts on the limestone walls and roofs. In some caves, these are the only phosphates (e.g., Isturitz, Acqua Fitusa, Monello, Scrivilleri, Personaggi). Their lower solubility makes them the most stable mineral phases during the initial guano digestion. In warmer and drier cave conditions, when bedrock is dolomitic, such as Toirano, newberyite has been found. The presence of gypsum on top of guano deposits is related to the evaporation of water, and these gypsum layers can become rather thick if the evaporative conditions prevail over long-time periods. This is the case for the Grosse Marguerite Cave (where the 15 cm-thick gypsum layer is about 5,000 years old), and in Eremita Cave in Sicily.

If the guano deposits and its leachates are in contact with clays (as interlayers in the limestone or as residual sedimentary deposits (both allochthonous and autochthonous), taranakite, variscite, and the rare montgomeryite and crandallite (e.g., Cocci Cave in Sicily) can typically form (Fig. 35). This often occurs in slightly more acidic conditions, because of the lower buffering effect of the carbonate rock. Both taranakite and variscite are among the most stable phosphates in the cave environment, and once formed they are often preserved over long periods of time, even in rather wet conditions (e.g., Pertosa Cave, Palombara

and Eremita caves). Instead, the presence of sulfides, mainly pyrite, in the bedrock will trigger the formation of (Al-rich) strengite and leucophosphite. In a fresh guano environment, spheniscidite can also form in the presence of pyrite or other sulfides such as galena (e.g., Corona 'e sa Craba, Baradla, and Julio caves). Corona 'e Sa Craba is a uniquely rich mineralogical environment, as the cave develops in a quartz vein in close contact with dolomitic and sulfide-rich limestones, thus causing the formation of phosphates such as robertsite, spheniscidite, leucophosphite, vashegyite, strengite, and possibly the rare berlinite. Robertsite has also been found in Palombara Cave, and if combined with some geomorphological indicators, this mineral might point to a hypogene origin of this cave.

A mature occurrence of mineralized guano normally leads to the formation of apatite-group minerals, occurring in well-crystallized forms, even layered stalagmites and stalactites in Eremita Cave, Cocci Abyss, Sicily (D'Angeli et al., 2018).

#### Survival of the very soluble gypsum

As most sulfates, gypsum is rather soluble and therefore, in very wet cave environments it can be dissolved. However, gypsum can be present where it

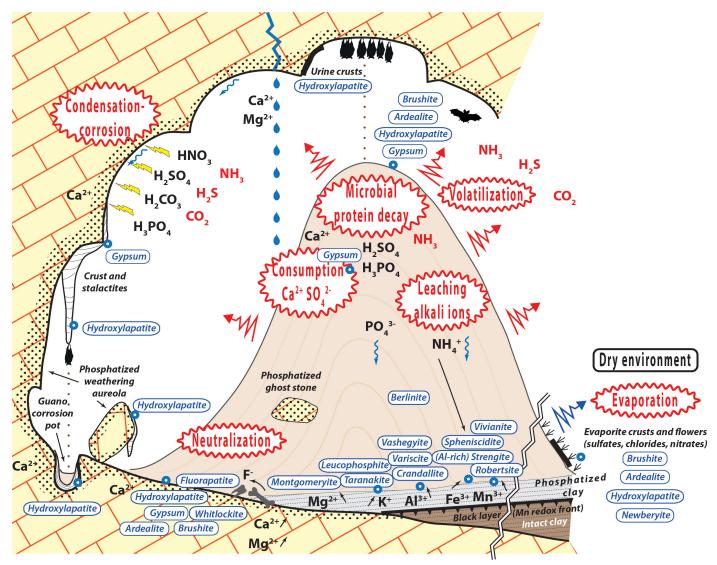


Fig. 34. Idealized diagram showing the occurrence of the eighteen guano-derived minerals in the studied cave environments and the processes involved in their formation.

forms from fresh guano decaying, in the upper part of guano piles which are drier. A typical occurrence is on the surface of a guano heap, especially at a distance from the dripping point location, where evaporation dries up the guano material, attracts the dissolved sulfates by capillary flow, and allows the crystallization and the conservation of sulfates at the interface between wet and dry areas. Providing no leaching occurs, gypsum can even be present at the wet base of guano deposits such as in Domica-Baradla. Gypsum can also persist in areas protected from dripping and leaching (Vanara, 2015).

#### Age and research potential of guano deposits

Some caves host long records covering the last millennium, such as in *Raganeous*, and possibly the two accumulations in Domica (*Palmový Háj* and *Prales*).

Older records are available in Grosse Marguerite (2741 to 5040 BC), and possibly also for Saint-Marcel, older than 1916 BC, and Baradla, younger than 281 BC. These deposits kept their original stratigraphy and were only affected by strong compaction due to their own weight. High-resolution dating would help to show their continuity or their gaps resulting from temporary abandonment of the shelter by bats, generally in relationship with climatic and/or environmental changes. However, the regular laminated guano accumulation in Raganeous was demonstrated to be continuous over the last millennium (Bentaleb et al., in prep.). Since guano deposits can be used as paleoenvironmental proxy, their potential for reconstruction based on the layered records, stable isotopes, pollen, and possibility of dating is very important (Onac et al., 2015; Cleary et al., 2017).

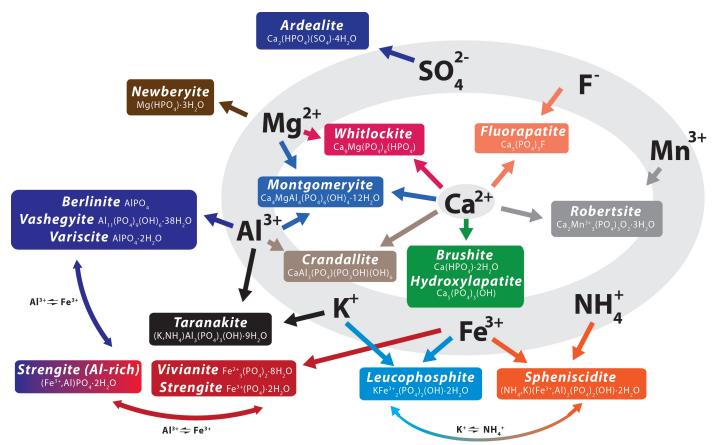


Fig. 35. Composition and inter-relationships of the 17 phosphate minerals according to the contribution of secondary components (Ca-Mg from host rock; Sulfate from organic decay or from pyrites; NH<sub>4</sub> from organic decay; F from bones or teeth; Al-K-Fe-Mn from detrital clay).

# Strong acidity of the decaying guano and environmental conditions for mineral genesis

Although fresh guano is slightly alkaline (Audra et al., 2016), its organic decay progressively delivers acids (nitric, sulfuric, phosphoric, and carbonic) that contribute to the increase of acidity with pH as low as 2-4 (Martini, 1993, 2000; Hill & Forti, 1997, 2001). This acidic environment contributes to the genesis of minerals, mainly phosphates but also some sulfates and sometimes nitrates, through the interaction with organic matter (e.g., NH<sub>4</sub><sup>+</sup>), with solutes from dripping (e.g., Ca, Mg), with the carbonate host rock (calcite or dolomite), and with detrital material providing Al, Fe, K, Mg, Mn, etc. In Domica Cave, we observed that the pH in the active guano pile decreases almost continuously with depth, where the 600-years old

mineralized guano shows pH of 3.7. The slight increase to 3.9 at the bottom could be interpreted as the beginning of acid solutions neutralization the proximity with the limestone pavement. On the contrary, in the inactive and already partly mineralized guano pile, the pH is very acidic all along the profile (3.5 to 3.9), with fluctuations that do not depend on depth. Such evolution of the acidity throughout the guano deposit profile leads to various environmental conditions that constrain the stability of minerals (Fig. 36). The stable end-members of phosphatization are the minerals of the apatite group formed in alkaline environment buffered by the carbonate substratum, and strengite and variscite in acidic conditions. Gypsum has a quite large stability field.

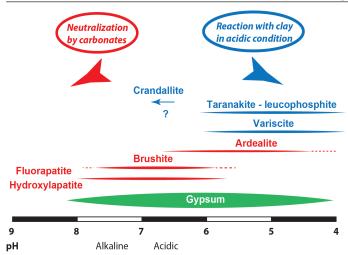


Fig. 36. Synthesis of approximate stability field of studied phosphate minerals and gypsum (after Onac & Vereş, 2003; Shahack-Gross et al., 2003; Onac et al., 2005b; Giurgiu & Tămaş, 2013).

#### CONCLUSIONS

We investigated guano-related minerals in 22 European caves located in the Slovak Republic, Hungary, France, Macedonia, and Italy. Using XRD, micro-Raman spectroscopy, and SEM, we identified 17 phosphate minerals (not considering Al-rich Strengite that is not recognized by IMA), some of them being very rare in caves, and one sulfate (gypsum). Combining pH and micro-climatic measurements with <sup>14</sup>C radiometric dates, we characterized some environmental parameters that control mineral genesis and the relationships between some minerals and their environment.

- On fresh decaying guano, the typical precursor mineral association is composed of brushite, ardealite, and gypsum. Such minerals are soluble and do not persist long in older guano accumulations.
- The apatite minerals (hydroxylapatite, fluorapatite) precipitate onto carbonates limestone or calcite speleothems by buffering of acid phosphate leachates. They generally occur as light or brown crusts on the walls and floors that can cover extensive areas. Their low solubility triggers high stability, hence the dominant abundance of this mineral association.
- The contact of acidic guano leachates with cave sediments, mainly clay and fluvial deposits and sometimes sulfides such as pyrite, allows the hydrolysis of the detrital minerals and the combination of their cations (Al, K, Fe, and Mn) with the phosphatic leachates. Taranakite, strengite, and variscite are the end-members of the mineral suites, hence the most stable and frequent minerals. However, the large amount of combinations and environmental characteristics allows the formation of rare phosphates, most of them being known only from a limited number of caves on Earth (montgomeryite, crandallite, (Al-rich) strengite, leucophosphite, vashegyite, vivianite, whitlockite, berlinite). Spheniscidite was found for the first time in caves in three of the studied sites, and robertsite is known only in

- three caves in the world, two of them presented in this study.
- Caves are generally wet, hence the abundance of hydrated phosphates. Less frequently, strong dehydration may produce rare minerals: guano self-combustion is suspected to be at the origin of a possible occurrence of berlinite (requiring temperatures above 570°C). Gypsum may persist in especially dry environment and even at the bottom of guano accumulation with poor drainage (damp conditions).
- Acidity of a decaying guano heap is well known. We measured pH as low as 3.5, which can persist even in old guano deposits. The distribution of acidity controls the mineral assemblage and the stability of end-member phosphates, in acid environment or alkaline at the contact with the limestone bedrock.
- We found guano more than 6000 years old, with an accumulation covering more than 2300 years in Grosse Marguerite Cave, France. Although the continuity of this deposit is still questioned, in Raganeous Cave it appears to be as old as 1000 years.
- Most of the previous studies on phosphate minerals, including this one, have been carried out by visually-oriented sampling strategy. Future investigations, which would also benefit from the fast development of portable analytical tools, should progressively turn to more continuous analyses. First, this will allow new mineral discoveries, and second it would lead to a better understanding of the environmental parameters that control mineral genesis. Finally, the use of different biological (pollen) and chemical (stable isotope) proxies available in guano offer a great potential for paleoenvironmental and paleoclimatic reconstruction.
- Some caves, characterized by specific environmental conditions, host rock composition, sediments, sulfides, or micro-climatic conditions, may represent outstanding sites for mineral studies, and particularly for phosphates and secondary sulfates related to bat guano deposits. Corona 'e sa Craba (Sardinia) and Domica-Baradla (Slovakia-Hungary), are two such examples of unique caves for their phosphate mineralogy.

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