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Highly siderophile and chalcophile element behaviour in abyssal-type and supra-subduction zone mantle: New insights from the New Caledonia ophiolite

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A B S T R A C T

The New Caledonia Ophiolite hosts one of the largest obducted mantle sections worldwide, offering a unique opportunity to investigate key mantle processes. The ophiolite comprises refractory harzburgites, locally overlain by mafic-ultramafic cumulates, and minor lherzolites. Previous geochemical studies indicated that the lherzolites are akin to abyssal-type peridotites, while the harzburgites underwent multiple melting episodes in MOR and supra-subduction zone environments, followed by late stage metasomatism.

In this work, Os isotopes, highly siderophile (HSE) and chalcophile element data are reported for the New Caledonia peridotites, in order to constrain the behaviour of these elements in abyssal-type and fore-arc mantle.

The variably serpentinised lherzolites (LOI = 6.4–10.7%) yield slightly subchondritic to suprachondritic initial Os isotopic compositions ($^{187}\text{Os}/^{188}\text{Os}_i = 0.1273\text{--}0.1329$) and subchondritic to chondritic Re/Os ratios (0.04–0.11). The gently sloping HSE patterns with increasing depletion towards Au show concentrations in the range of other lherzolites from MOR or continental setting. Sulphur contents are high and variable (202–1268 ppm), and were likely increased during serpentinisation. By contrast, Se/Te ratios and concentrations are within the range of primitive mantle (PM) values, meaning that these elements were not significantly mobilised during serpentinisation.

Although displaying homogenous petrographic and geochemical features, the harzburgites are characterised by extremely heterogeneous Re—Os and HSE compositions.

Type-A harzburgites exhibit subchondritic $^{187}\text{Os}/^{188}\text{Os}_i$ (0.1203–0.1266) and low Re/Os ratios (0.01–0.04). The strong IPGE-PPGE fractionations ($\text{Pd}_N/\text{Ir}_N = 0.21\text{--}0.56$), coupled with positive Pt anomalies and S-Se-Te abundances often below the detection limit, suggest high melt extraction rates, resulting in sulphide consumption and Os—Ru metal alloy stabilisation.

Type-B harzburgites possess strongly fractionated, Os-Ir-Pt poor (Os = 0.003–0.072 ng/g, Ir = 0.0015–0.079 ng/g) and Pd—Re enriched patterns, associated with chondritic to suprachondritic measured $^{187}\text{Os}/^{188}\text{Os}$ (0.127–0.153). These characters are uncommon for highly depleted mantle residues. Interaction with an oxidised component does not appear as a viable mechanism to account for the IPGE-depleted patterns of type-B harzburgites, as calculated oxygen fugacities are close to the FMQ buffer ($\text{Log } \Delta\text{FMQ} = 0.35$ to 0.65).

The strikingly uniform mineralogical and geochemical features displayed by both harzburgite sub-types suggest that the different HSE patterns are not linked to their recent evolution, implying that subduction-related processes were superimposed on geochemical heterogeneous mantle domains, which exerted an important control on HSE behaviour during melt extraction and post melting metasomatism.

We propose that the HSE characters of the studied peridotites reflect the presence of a highly heterogeneous mantle source with a long term (>1 Ga) evolution, possibly linked to the Zealandia formation.

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1. Introduction

Highly siderophile elements (HSE: PGE+ Au—Re) are powerful geochemical tracers that can provide useful information for a variety of mantle processes, such as mantle melting, metasomatism and melt-fluid/mantle interaction (e.g. [Ackerman et al., 2009](#); [Lorand et al., 2008](#); [Luguet et al., 2001, 2003, 2007](#)). However, our knowledge concerning the behaviour of HSE in mantle source rocks of primitive arc magmas and the role of the subduction zone environment on HSE partitioning (i.e. hydrous melting, melt/fluid-mantle interaction) still remains quite fragmentary. Furthermore, although abundant HSE data are now available for different types of mantle peridotites, HSE data on fore-arc peridotites are remarkably scarce (e.g. [Becker and Dale, 2016](#)).

The New Caledonia ophiolite (Peridotite Nappe) hosts one of the largest and best preserved mantle sections worldwide, providing an excellent opportunity to investigate upper mantle processes. The rock exposures are dominated by harzburgite tectonites bearing a supra-subduction zone affinity ([Marchesi et al., 2009](#); [Pirard et al., 2013](#); [Secchiari et al., 2020](#); [Ulrich et al., 2010](#)). The main geochemical and isotopic features of these rock-types reflect a complex polyphase evolution, including several melting episodes in different geodynamic settings and subduction zone metasomatism ([Marchesi et al., 2009](#); [Secchiari et al., 2020](#); [Ulrich et al., 2010](#)). Minor spinel and plagioclase lherzolites, with compositions similar to abyssal peridotites, occur as discrete bodies in the north-western part of the island. The lherzolites record a different history compared to the extremely refractory harzburgites, as highlighted by their different geochemical signature ([Secchiari et al., 2016](#); [Ulrich et al., 2010](#)).

In this work, a set of fully characterised peridotites (i.e. whole rock and in situ major and trace element contents, Sr-Nd-Pb isotopes) from New Caledonia ([Secchiari et al., 2016, 2019, 2020](#)) has been used to investigate Re—Os, HSE and chalcophile element (S—Se—Te) systematics. The main aims of this work are: 1) to examine the behaviour of these elements in the lherzolites (i.e. presumed abyssal peridotites) and in the ultra-depleted harzburgites, which may represent rocks from a former supra-subduction zone mantle wedge; 2) to constrain the behaviour of HSE and chalcophile elements during subduction zone processes.

2. Geological setting and petrological background

New Caledonia is a NW–SE elongated island located in the SW Pacific region, between the eastern margin of Australia and the Vanuatu archipelago ([Figs. 1 and 2a](#)). The island represents the emerged portion of the submarine Norfolk Ridge and it is composed by a mosaic of volcanic, sedimentary and metamorphic terranes, ranging in age from Permian to Miocene ([Aitchison et al., 1995](#); [Cluzel et al., 2001, 2012](#); [Lagabrielle et al., 2013](#)). These terranes were amalgamated during two major tectonic events: 1) an Early Cretaceous tectonic convergence phase (Paris, 1981) and 2) a Paleocene to Late Eocene subduction culminated in the obduction of the ophiolite. Both events were characterised by high-pressure low-temperature (HP-LT) metamorphism in connection with plate convergence. New Caledonia can be sub-divided into four main geological domains ([Cluzel et al., 2001](#); see [Fig. 1](#)): (i) the Basement units (pre-Late Cretaceous basement and Late Coniacian-to-Late Eocene sedimentary cover), (ii) the

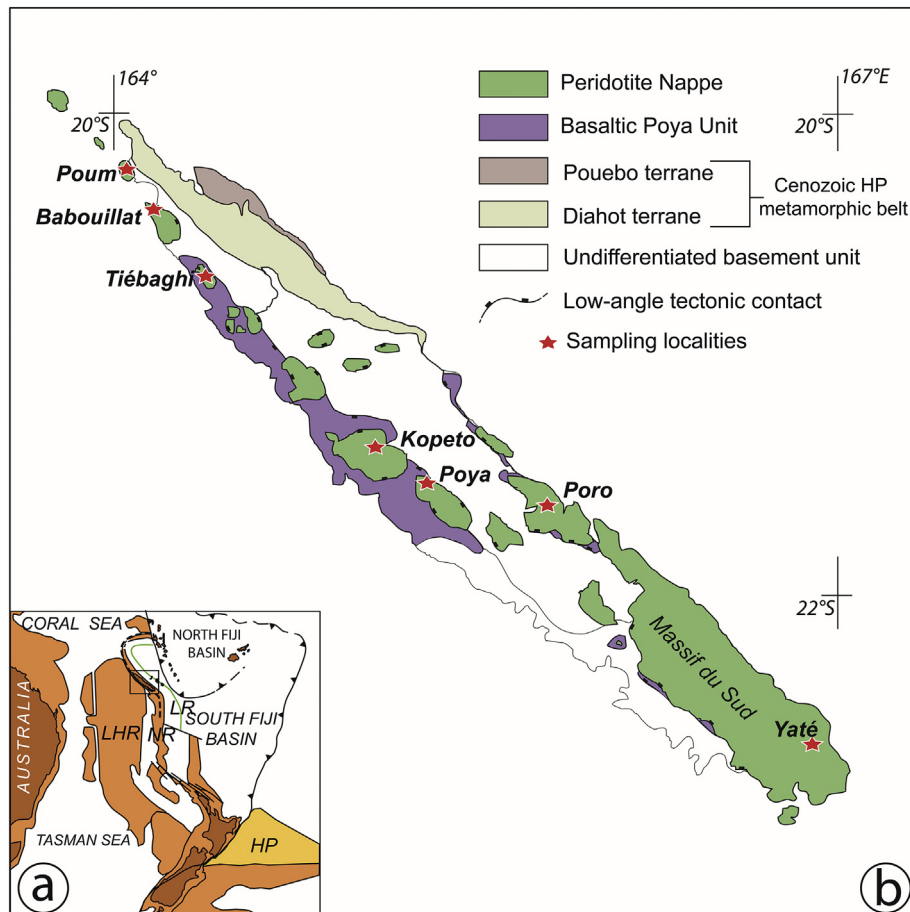


Fig. 1. a) Present-day structures of the Southwest Pacific region modified after [Cluzel et al. \(2012\)](#). Dark orange, land; light orange, continental plateau; white, oceanic basins (LHR: Lord Howe Rise, NR: Norfolk ridge, LR: Loyalty ridge, HP: Hikurangi Plateau); b) simplified geological map of New Caledonia showing distribution of the Peridotite massifs (modified after [Cluzel et al., 2012](#)). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

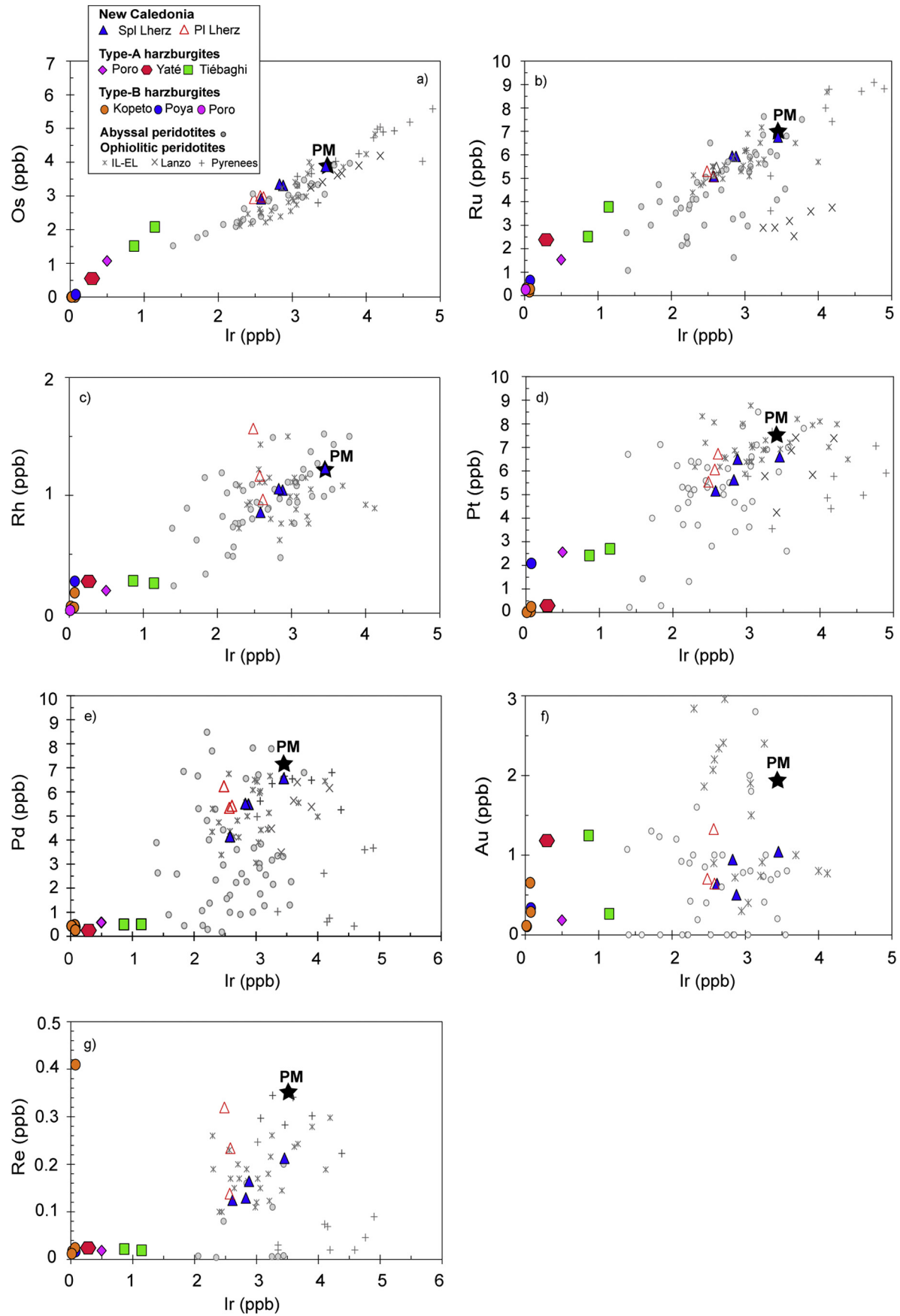


Fig. 2. Variations of Os, Ru, Rh, Pt, Pd, Au and Re vs. Ir for the New Caledonia peridotites. Abyssal peridotites (Kane fracture zone: Snow and Schmidt, 1998; Brandon et al., 2000; Luguet et al., 2001, 2003; Marchesi et al., 2013; MAR: Harvey et al., 2006; Lena trough: Lassiter et al., 2014) and ophiolitic peridotites (IL-EL: Internal and External Ligurides, Snow et al., 2000; Luguet et al., 2004; Fischer-Gödde et al., 2011; Lanzo: Becker et al., 2006; Pyrenees: Becker et al., 2006; Luguet et al., 2007) are shown for comparison.

Table 1

Concentrations of the HSE, S, Se and Te, Os isotopes, selected major elements and ratios for the New Caledonia peridotites.

Sample	Rock	Al ₂ O ₃ wt%	LOI %	Os (ng/g)	Ir (ng/g)	Ru (ng/g)	Rh (ng/g)	Pt (ng/g)	Pd (ng/g)	Au (ng/g)	Re (ng/g)	Os _N /Ir _N	Os _N /Ru _N	Pd _N /Ir _N	Ru _N /Ir _N
POU1A	Pl L	2.49	7.56	2.93	2.48	5.31	1.57	5.54	6.23	0.704	0.319	1.06	0.99	1.24	1.07
POU2	Sp L	1.46	8.63	3.88	3.45	6.76	1.23	6.60	6.56	1.04	0.213	1.01	0.18	0.94	5.71
POU2B	Sp L	2.57	10.69	3.31	2.88	5.93	1.04	6.49	5.49	0.503	0.165	1.03	1.09	0.94	0.95
POU3	Sp L	1.30	9.60	2.91	2.58	5.07	0.86	5.15	4.14	0.642	0.234	1.01	1.05	0.79	0.97
BA1	Sp L	1.74	6.39	3.34	2.85	5.95	1.05	5.62	5.51	0.943	0.129	1.06	0.27	0.96	3.88
BAB1B	Pl L	2.61	6.98	2.96	2.61	5.46	0.965	6.73	5.42	0.641	0.125	1.02	0.24	1.02	4.30
BAB2B	Pl L	2.83	8.43	3.00	2.57	5.08	1.16	6.06	5.33	1.33	0.138	1.05	1.67	1.02	0.63
TI1	H	0.78	9.01	1.51	0.864	2.52	0.275	2.42	0.488	1.25	0.022	1.57	1.08	0.28	1.46
TI2	H	1.03	6.04	2.07	1.14	3.78	0.254	2.70	0.499	0.264	0.019	1.63	0.98	0.21	1.65
PO4	H	0.43	0.18	1.07	0.499	1.53	0.191	2.56	0.570	0.160	0.018	1.93	1.26	0.56	1.53
YA1	H	0.46	6.83	0.554	0.297	2.35	0.326	0.268	0.263	1.17	0.022	1.67	0.42	0.44	3.95
PY1	H	0.78	0.00	0.049	0.077	0.643	0.269	2.08	0.378	0.336	0.016	0.58	0.14	2.43	4.19
Duplicate				0.023	0.133	0.693	0.171	2.44	0.588	0.311	0.062				
KPT2	H	0.70	3.03	0.004	0.022	0.323	0.057	0.078	0.431	0.107	0.018	0.18	0.02	9.50	7.22
Duplicate				0.036	0.017	0.211	0.040	0.210	0.620	0.144	0.010				
KPT5	H	0.74	0.12	0.004	0.066	0.165	0.048	0.038	0.464	0.654	0.024	0.05	0.04	3.46	1.25
Duplicate				0.007	0.062	0.165	0.054	0.046	0.512	0.603	0.056				
PO3	H	0.41	0.00	0.003	0.015	0.256	0.023	0.006	0.428	0.084	0.012	0.18	0.02	14.12	8.57
Duplicate				0.055	0.018	0.243	0.016	0.009	0.573	0.293	0.008				
KPT3	H	0.67	0.67	0.072	0.079	0.280	0.172	0.248	0.250	0.288	0.410	0.81	1.87	0.17	0.19

Cenozoic HP-LT metamorphic belt, (iii) the basaltic Poya Terrane and (iv) a large slab of peridotites, named the Peridotite Nappe.

The Peridotite Nappe represents an allochthonous sheet of oceanic lithosphere belonging to the former South Loyalty basin thrust on the continental basement of the Norfolk Ridge at the end of the Eocene subduction. The emplacement of the ophiolitic nappe resulted from the failed subduction of the Norfolk Ridge tip in a NE-dipping subduction zone, which culminated in the obduction of the Loyalty subarc lithosphere ~ 34 Ma ago (Cluzel et al., 2012).

The Peridotite Nappe has an extension of about 8000 km² and is mostly exposed in the Massif du Sud, where a thick harzburgite–dunite unit, locally overlain by kilometre-scale lenses of mafic and ultramafic intrusives, crops out. The sequence is believed to represent a crust–mantle boundary that records the onset of the Eocene subduction in a nascent arc setting (Marchesi et al., 2009; Pirard et al., 2013; Secchiari et al., 2018). Recent geochemical studies have shown that the ultramafic intrusives (i.e. dunites and wehrlites) crystallised from variably depleted melts with island arc basalt affinity, after massive interactions with the underlying harzburgite (Marchesi et al., 2009; Pirard et al., 2013). In contrast, the mafic rocks (i.e. gabbro-norites) have a cumulate origin (Marchesi et al., 2009; Pirard et al., 2013; Secchiari et al., 2018) and derive from crystallisation of primitive, non-aggregated, ultra-depleted melts showing involvement of a subduction-related component in their source (Secchiari et al., 2018).

The harzburgites are also exposed in the northern Tiébaghi massif (Ulrich et al., 2010) or as sparse tectonic klippen in the central part of the island (e.g. Kopeto, Poya, Koniambo), where exceptionally fresh peridotites display primary mineral assemblages similar to the more serpentinised rocks of the Massif du Sud.

The New Caledonia harzburgites bear an overall ultra-depleted composition, inherited from a complex multistage evolution linked to the development of the Eocene subduction system (Marchesi et al., 2009; Secchiari et al., 2020; Ulrich et al., 2010). Geochemical studies have proposed that the harzburgites formed by high degrees of fluid-assisted melting (up to 20–25% in a supra-subduction zone environment, see Marchesi et al., 2009; Ulrich et al., 2010). More recently, the work of Secchiari et al. (2020) provided further constraints on the evolution of these rock-types, tracking their history from melting to late stage metasomatism. Accordingly, the harzburgites underwent two partial melting episodes in the spinel stability field: a first melting phase in a MOR

setting (15% melting degrees), followed by hydrous melting in a supra-subduction zone setting (up to 18% fluid-assisted melting). Post-melting cooling and re-equilibration at lithospheric conditions was accompanied by interaction with ultra-depleted slab-derived hydrous melts (Secchiari et al., 2019, 2020). These metasomatic processes are testified in the harzburgites by the widespread occurrence of secondary metasomatic phases (i.e. thin films of Al₂O₃–, CaO– poor orthopyroxene, and low Al₂O₃ and Na₂O clinopyroxene), L-MREE and Zr–Hf bulk rock enrichments, and possibly by the poorly radiogenic Nd isotopic ratios shown by some samples (Secchiari et al., 2020).

Compared to the harzburgites from the central and the southern massifs, Tiébaghi samples display a more fertile nature, as revealed by the higher bulk trace element concentrations as well as by the occurrence of a small fraction (up to 4 vol%) of clinopyroxene (see Ulrich et al., 2010; Secchiari, 2016 PhD thesis).

The main geochemical and petrological features of the spinel and plagioclase lherzolites are thought to reflect moderate melt extraction degrees (8–9%) in a MOR environment, followed by refertilisation by depleted MORB-type melts, yielding plagioclase lherzolites. The main petrological and geochemical features of the lherzolites have been reported in detail by Secchiari et al. (2016).

2.1. Sample description

In this contribution, sixteen samples of peridotites fully characterised for lithophile element geochemistry (i.e. major, trace element and Sr–Nd–Pb isotope compositions) were analysed for mass fractions of all PGE, Re, Au, S, Se, Te and ¹⁸⁷Os/¹⁸⁸Os. Detailed descriptions of the lherzolites and harzburgites, including trace element chemistry and Sr–Nd–Pb isotopes, are provided in the works of Secchiari et al. (2016) and Secchiari et al. (2019, 2020), respectively.

Lherzolite samples are from the Poum and Babouillat areas, while the harzburgites were collected from several outcrops and mine zones along the island: Yaté, Kopeto, Poya, Poro and Tiébaghi (Fig. 1b and Table 1). The lherzolites include slightly serpentinised (LOI = 6.9–10.7%) spinel and plagioclase lherzolites, while the harzburgites are typically not or only weakly serpentinised (LOI = 0–3%), except for samples YA1, TI1 and TI2 (LOI = 6.0–9.0%).

Both lherzolites and harzburgites are low strain mantle tectonites, showing dominant porphyroclastic textures (Fig. S1a–b) and local protomylonite development. Spinel lherzolites have 7–8 vol%

Pt _N /Ir _N	Pt _N /Ru _N	¹⁸⁷ Re/ ¹⁸⁸ Os (2SE)	¹⁸⁷ Os/ ¹⁸⁸ Os <i>measured</i>	2SE	¹⁸⁷ Os/ ¹⁸⁸ Os _i	γOs (53Ma)	T _{MA} (PM) Ga	T _{RD} (PM) Ga	T _{RD2} (PM) Ga	S (μg/g)	Se (ng/g)	Te (ng/g)	S/Se	Se/Te
1.03	0.96	0.525(1)	0.130822	9.1E-06	0.130358	2.9	0.8	f	f	202	74.9	11.5	2703	6.5
0.88	0.15	0.265(1)	0.127574	6.8E-06	0.127340	0.5	0.7	f	0.3	582	69.6	11.6	8364	6.0
1.04	1.09	0.240(1)	0.128227	8.0E-06	0.128016	1.1	0.4	f	0.2	1268	77.8	10.8	16,289	7.2
0.92	0.95	0.387(1)	0.131356	9.1E-06	0.131015	3.4	f	f	f	362	54.0	7.3	6691	7.4
0.92	0.24	0.186(1)	0.131485	7.9E-06	0.131320	3.7	f	f	f	294	91.3	13.8	3218	6.6
1.19	0.28	0.202(1)	0.129664	8.9E-06	0.129485	2.2	f	f	0.0	327	66.2	9.5	4943	7.0
1.09	1.73	0.223(1)	0.133084	8.9E-06	0.132887	4.9	f	f	f	289	67.6	9.8	4278	6.9
1.29	0.89	0.070(25)	0.12479	1.7E-05	0.12473	-1.5	0.8	0.7	0.7	9	1.1	1.0	8188	1.0
1.09	0.66	0.045(18)	0.12662	1.3E-05	0.12658	-0.1	0.5	0.4	0.4	10	0.7	0.4	14,053	1.7
2.36	1.54	0.081(35)	0.12040	2.4E-05	0.12033	-5.0	1.5	1.3	1.3	6	3.1	0.8	2084	4.1
0.42	0.11	0.196(68)	0.12551	4.7E-05	0.12534	-1.0	1.0	0.6	0.6	40	4.9	0.4	8117	11.4
2.97	2.97	1.62(8)	0.1299	5.3E-04	0.1284	1.4	0.0	0.0	0.2	6	0.9	0.9	6103	1.1
0.22	0.22	13.0(1)	0.127	1.0E-03	0.115	-8.9	0.1	f	f	53	0.3	1.0	196,197	0.3
		19(1)	0.148	5.6E-03	0.131	3.6				5	1.3	0.4	4164	2.8
		1.4(1)	0.302	1.3E-03	0.301	137.8				7	1.3	8.2	4947	0.2
0.21	0.21	32(2)	0.147	6.2E-03	0.118	-6.6	0.0	f	f	3	3.0	0.7	1133	4.1
		37(1)	0.160	3.2E-03	0.127	0.5				3	0.7	1.3	4453	0.5
0.02	0.02	19(2)	0.153	8.4E-03	0.136	8.0	0.1	f	f	3	1.4	0.6	2129	2.2
		0.69(7)	0.1239	4.7E-04	0.1233					bdl	bdl	0.8	-	-
6.92	6.92	28(1)	0.1273	3.5E-04	0.1204	-5.1	0.0	0.3	1.3	6	bdl	bdl	-	-

Duplicate: replicate digestion of the same sample powder.

Pl L = plagioclase lherzolite, Sp L = spinel lherzolite, H = harzburgite.

Values of PM ¹⁸⁷Os/¹⁸⁸Os = 0.1296 and ¹⁸⁷Re/¹⁸⁸Os = 0.434 used for calculation of T_{MA} and T_{RD} ages (Meisel et al., 2001); f = future model ages. T_{RD2} (PM) indicates depletion ages calculated taking into account Re addition that may have occurred during peridotite evolution.

clinopyroxene and display a typical abyssal-type REE signature. The plagioclase lherzolites show melt impregnation microstructures (Fig. S1b) and are slightly enriched in incompatible trace element (REE, Ti, Y, and Zr) with respect to the spinel lherzolites. Harzburgites are extremely depleted rocks, as highlighted by the general absence of clinopyroxene (with the exception of sample TI2, where clinopyroxene is ~ 4 vol%, Fig. S1c) and remarkably low incompatible trace element contents (Secchiari et al., 2020). The primary mantle assemblage is composed of olivine, orthopyroxene and spinel. The occurrence of thin films of metasomatic ortho- and clinopyroxene (Fig. S1d) was interpreted as resulting from percolation of small fractions of subduction-related magmas (Secchiari et al., 2019, 2020).

All the lherzolites studied here contain trace amounts of sulphides with variable size, shape and position. Sulphides in five selected samples (i.e. BA1, POU2, POU3, POU1A, BAB1B, see Table S1) were analysed for their major element chemistry composition.

Frequent interstitial sulphide grains occur as polyhedral blebs, rounded or irregular-shaped grains (Fig. S2a-d-e), generally located at olivine-orthopyroxene or at olivine-olivine grain boundaries. They range from tiny crystals (~ 60 × 30 μm) up to 300 × 100/150 μm in maximum dimensions. Enclosed sulphides hosted within large olivine porphyroclasts have been also observed. Sulphide inclusions (Fig. S2a-e) have variable size (30 to 80 μm × 60 to 150 μm) and shape, from polyhedral to spherical. Very tiny enclosed sulphide grains (~ 5–6 μm × 10 μm) have been recognised in sample POU1A.

Major element composition of the sulphide phases is relatively homogeneous (Table S1 and Fig. S2, S3), showing no significant difference among enclosed and interstitial sulphides. Chemical composition of the studied sulphides mainly fall in the field of Ni-poor (Fe/Ni = 1.3–2.1) monosulphide solid solution + Liquid (Fig. S3), but Ni-rich (Fe/Ni = 0.7–0.8) pentlandite, resulting from cooling and re-equilibration (Guo et al., 1999), has been also identified in four of the investigated samples (POU2, POU3, POU1A, BAB1B).

In the harzburgites, sulphide phases have not been identified, nor through petrographic observation or microprobe analyses. This is consistent with the refractory nature and the high melt extraction rates experienced by these rock-types, which prevented sulphide retention in the residual mantle source.

3. Analytical methods

3.1. HSE and chalcophile elements

Seven lherzolites and nine harzburgites (including four duplicates) have been analysed in the geochemistry laboratory at Freie Universität for HSE, S, Se, Te mass fractions in whole rocks and ¹⁸⁷Os/¹⁸⁸Os.

Detailed procedure descriptions have been given in previous works from this laboratory (e.g. Fischer-Gödde et al., 2011; Wang et al., 2013; Wang and Becker, 2013). The methods will only be briefly summarised below.

About 2.5 g of sample powder was weighed into 90 mL quartz glass digestion vessels and spiked with mixed ¹⁹¹Ir–⁹⁹Ru–¹⁹⁴Pt–¹⁰⁵Pd, ⁷⁷Se–¹²⁵Te, ¹⁸⁵Re–¹⁹⁰Os and ³⁴S solutions. Then, 5 mL 14 mol/L, N₂-bubbled HNO₃ and 2.5 mL 9 mol/L HCl were added. The vessels were immediately sealed with Teflon tape and samples were digested for 16 h at 320 °C and 100 bar. After digestion, osmium was extracted from the reverse aqua regia into chloroform, back extracted into HBr (Cohen and Waters, 1996), and further purified by micro distillation from a H₂SO₄-dichromate solution into 15 μL of HBr (Birck et al., 1997).

Osmium isotopes were determined as OsO₃⁻ in negative mode using the Thermo Finnigan Triton TIMS, using a secondary electron multiplier. Signal intensities of the spike isotope ¹⁹⁰Os of samples were ~150,000–500,000 cps. Standard runs with different amounts of Os on the filament (10 pg and 100 pg) were also run in between the studied samples, yielding an average value of 0.1139 ± 0.0002 (2 s. d., n = 24) for 100 pg loads. Two hundreds scans were collected in each measurement for high-Os samples, while at least 120–140 scans were obtained for the low-Os samples. Raw data were corrected for isobaric OsO₃⁻ interferences, mass fractionation using the ¹⁹²Os/¹⁸⁸Os ratio of 3.08271, contributions from the Os spike solution and blank contributions. ¹⁸⁷Os/¹⁸⁸Os were finally adjusted relative to the mean of the Os standard. The oxygen isotope compositions used for the oxide correction of Os oxide molecules were ¹⁸O/¹⁶O of 0.00204 and ¹⁷O/¹⁶O of 0.00037 (Nier, 1950).

About 50% of the digestion solution was used for separation of the HSE fraction and about 30% for S–Se–Te separation. Chemical separation of the HSE fraction from the matrix was performed on columns filled

with 10 mL of pre-cleaned Eichrom 50 W-X8 (100–200 mesh) cation exchange resin (Fischer-Gödde et al., 2011). During separation, the HSE fraction was collected in 14 mL 0.5 mol/L HCl–40 vol% acetone mixture. After the volume of the solution has been reduced to about 2 mL it was analysed for Au, Re, Ir and Pt. In order to remove interfering Cd, the remaining solution was further purified in 0.2 mol/L HCl on 3 mL Eichrom 50 W-X8 (100–200 mesh) resin. The collected solution was evaporated to near dryness and the residue was taken up in 0.28 M HNO₃ for ICP-MS analysis. The analyses were carried out using a single collector Element XR instrument. We used either a Scott-type spray chamber (Re, Ir, Pt, Au) or an Aridus-I desolvator (Ir, Ru, Pt, Rh, Pd) at an oxide formation rate of $\text{CeO}^+/\text{Ce}^+ < 0.004$.

A two-step ion exchange chromatography method was used for separation of S, Se and Te (see Wang et al., 2013). Sulphur measurements were performed on the S–Se fraction at medium mass resolution mode on the Element XR. Selenium and Tellurium were measured using a double pass Scott type glass spray chamber at low mass resolution mode on the Element XR, combined with a hydride generation sample introduction system by reacting the sample solution with 1 g/100 g NaBH₄ in 0.05 mol/L NaOH (see Wang et al., 2013 for details).

For each batch of analysis, one procedural blank has been used. Procedural blanks yielded the following mean values (± 1 s.d., $n = 4$ –5): Re = 2.5 ± 2.0 pg; Os = 0.5 ± 0.3 pg with $^{187}\text{Os}/^{188}\text{Os}$ ratios of 0.14 ± 0.03 ; Ir = 15 ± 5 pg; Ru = 45 ± 14 pg; Rh = 24 ± 22 pg; Pt = 23 ± 29 pg; Pd = 640 ± 330 pg; Au = 4 ± 2 pg; Te = 1.1 ± 0.8 ng; Se = 2.3 ± 0.8 ng; S = 2.8 ± 0.7 μg . Samples were corrected for total procedural blanks using the mean values. Blank corrections for Re are negligible for most of the analysed samples (≤ 0.3 – 0.8%), but more significant for the harzburgites (~ 4 – 8%). Blank corrections for Pt and Pd are again negligible for the lherzolites (~ 0.2 – 0.3%), a few percents for the harzburgites (~ 0.4 – 4% , with the exception of KPT2, KPT5 and PO3 for which the correction for Pt is ~ 11 – 36%). Blanks of Os, Ir, Ru and Rh are insignificant for most of the samples ($\leq 0.4\%$) but higher for the most depleted harzburgites, i.e. KPT2, KPT5 (~ 2 – 7% for Os, Ir, Rh) and PO3 ($\sim 9\%$ for Os and Ir, 13% for Rh). Blank corrections for S, Se and Te in lherzolites range between 1 and 1.7% (S – Se) and 3–6% (Te), while for the harzburgites blank corrections for these elements strongly affected the obtained results (corrections ~ 10 – 26% for S and up to 40–80% for Se and Te), given the very low measured abundances.

3.2. Oxygen fugacity

Iron oxidation state in studied spinels was measured using the “flank method” developed for the JEOL JXA-8200 electron microprobe at Freie Universität (Goncharov, 2018). The position shift and intensity variation of FeL α and FeL β x-ray lines were investigated using as standards a collection of nineteen natural mantle spinels previously characterised by Mössbauer spectroscopy for their $\text{Fe}^{3+}/\Sigma\text{Fe}$ at IPGG RAS (St. Petersburg, see Goncharov and Ionov, 2012; Goncharov et al., 2015).

The analytical procedure was similar to the experiments performed over the last decades to study $\text{Fe}^{3+}/\Sigma\text{Fe}$ in mantle garnets after the “flank” approach developed by Höfer and Brey (2007).

The flank positions for the spectrometer were obtained from the difference spectrum of almandine and andradite with known iron oxidation state in the wavelength range related to FeL α and FeL β lines. The intensities at flank positions near FeL α and FeL β lines were collected as two fake elements, setting counting time at 300 s on both measurements, with 3 repetitions in the core and in the rim respectively, and considering 5 separate spinel grains within one thin section. Measurement conditions were 15 kV and 60 nA using TAP crystal for intensities at flank positions and with the remaining four spectrometers measuring chemical composition at the same spot simultaneously. The studied spinels show no significant core to rime zoning in terms of FeO content, Cr# and FeL β /FeL α ratio. Averaged FeL β /FeL α ratios for each sample were used to calculate iron oxidation state of spinel using the equation

with coefficients obtained after investigation of the standard collection (see Table 2).

4. Results

4.1. HSE and chalcophile elements in spinel and plagioclase lherzolites

HSE and chalcophile element compositions of the New Caledonia lherzolites are reported in Table 1 and displayed in Figs. 2, 3, 4. Spinel and plagioclase lherzolites are relatively homogeneous in terms of HSE, Se, Te concentrations, abundance patterns and Os isotopic compositions, with plagioclase-bearing samples showing indistinguishable patterns from those of spinel lherzolites. The concentrations of the highly siderophile and chalcophile elements are in the range of those observed for modern abyssal and ophiolitic peridotites, displaying good correlation for Ir group PGE (IPGE, e.g., Os vs. Ir and Ir vs. Ru) and more dispersed variations for the Pt group PGE (PPGE, Fig. S4). In primitive mantle (PM) normalised concentration diagrams (Fig. 3), the lherzolites exhibit flat or gently sloping negative patterns with similar PM-normalised PGE concentrations and depletions in Au (except for sample BAB2B) and Re compared to the PGE (with the exception of POU1A and POU3 for Re). Overall, absolute contents of the PGE are similar or slightly lower than primitive mantle (PM) values (Becker et al., 2006; Fischer-Gödde et al., 2011), overlapping the field of the abyssal peridotites and peridotite tectonites from continental settings (e.g., Fig. 2 and Becker and Dale, 2016). Ru/Ir and Pd/Ir ratios are suprachondritic, as observed for other mantle lherzolites (e.g. Becker et al., 2006; Lorand et al., 1999; Luguet et al., 2003; Rehkämper et al., 1999).

Initial $^{187}\text{Os}/^{188}\text{Os}$ ratios calculated at 53 Ma (i.e. the inferred age of initial magmatism in the subduction system, e.g. Cluzel et al., 2006) vary from chondritic to slightly suprachondritic (0.1273 – 0.1329 , Fig. 4a), corresponding to $\gamma_{\text{Os}(53\text{Ma})}$ of 0.5 to 4.9 . These values overlap with data of abyssal peridotites and orogenic peridotites, but tend to be somewhat higher than for other mantle lherzolites bearing comparable depletion degrees (Fig. 4). $^{187}\text{Re}/^{188}\text{Os}$ ranges from subchondritic to slightly suprachondritic values (0.186 – 0.525 , see Fig. 4c).

Se and Te are positively correlated in the lherzolites (Fig. 5a) and range between 54 and 91.3 ng/g and 7.3 – 13.8 ng/g, respectively. Se/Te ratios (5.9 – 7.1) are slightly lower than the PM value and similar to the data previously obtained on depleted lherzolites (Wang and Becker, 2013). Se and Te do not display any correlation with PGE abundances, with the exception of Te, which shows a weak correlation with Pd (Fig. 5b). S contents are high and variable (202 – 1268 $\mu\text{g/g}$) compared to unserpentinised peridotites, leading to high S/Se ratios (2703 – $16,289$, Fig. 5c).

4.2. HSE and chalcophile elements in harzburgites

On the basis of HSE behaviour and Os isotopic compositions (Figs. 2, 3 and 6, Table 1), the studied harzburgites can be grouped into two different sub-types: type-A and type-B.

Type-A harzburgites (TI1, TI2, PO4 and YA1) are characterised by notably lower PGE and chalcophile elements mass fractions (Figs. 2, 4) and more fractionated patterns (Fig. 6a) compared to the lherzolites. Mass fractions of the PGE are 1.07 – 2.07 ng/g for Os, 0.50 – 1.14 ng/g for Ir, 1.53 – 2.52 ng/g for Ru, 0.19 – 0.27 ng/g for Rh, 2.42 – 2.70 ng/g for Pt and 0.49 – 0.57 ng/g for Pd. Among the sub-group A, sample YA1 displays distinct PGE abundances, showing much lower Os, Ir, Pt, Pd contents (0.55 ppb for Os, 0.30 ppb for Ir, 0.27 ppb for Pt and 0.26 ppb for Pd), with exceptions for Ru, Rh, Au and S.

HSE and chalcophile element diagrams of type-A harzburgites display fractionated patterns, with concentrations decreasing from Os towards Re. Os and Ru are enriched compared to Ir, leading to correlated suprachondritic Os/Ir and Ru/Ir ratios (Os/Ir = 2.9 – 7.9 ; Ru/Ir = 1.8 – 2.1 , Fig. 4). Pt and Au generally show positive spikes, more

Table 2

Equilibration temperature, pressure, and oxygen fugacity calculated for selected peridotite samples. Average spinel compositions are also reported.

Sample	Type	$\text{Fe}^{3+}/\Sigma\text{Fe}^a$	$\text{Fe}^{3+}/\Sigma\text{Fe}$ (rel. err.%) ^b	T Ol-Spl (°C) ^c	P (GPa)	Wood ^d (1990)	
						log <i>f</i> (O ₂)	ΔFMQ
TI2	Type-A harz	0.21 (0.03)	15.0	870	1.0	−11.37	0.92
YA1	Type-A harz	0.18 (0.03)	14.5	815	1.0	−12.66	0.67
KPT5	Type-B harz	0.21 (0.03)	16.0	840	1.0	−12.20	0.65
PO3A	Type-B harz	0.20 (0.03)	14.8	940	1.0	−10.76	0.33
PY1B	Type-B harz	0.21 (0.03)	16.6	930	1.0	−10.90	0.35

Sample	Type	Spinel											L ₃ /L _α ^e
		SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	FeO	Fe ₂ O ₃	MnO	MgO	Total	Cr#	Mg#	
TI2	Type-A harz	0.04	0.03	43.45	25.35	10.20	2.99	0.15	16.93	99.13	0.281	0.747	0.804
YA1	Type-A harz	0.02	0.01	21.56	45.88	16.76	4.03	0.29	10.81	99.35	0.588	0.535	0.896
KPT5	Type-B harz	0.01	0.01	29.66	41.41	11.31	3.31	0.21	14.41	100.33	0.484	0.695	0.825
PO3A	Type-B harz	0.02	0.00	17.27	53.48	12.58	3.54	0.25	12.32	99.47	0.675	0.636	0.833
PY1B	Type-B harz	0.01	0.01	29.35	41.87	11.20	3.32	0.20	14.42	100.38	0.492	0.696	0.826

^a $\text{Fe}^{3+}/(\text{Fe}^{2+} + \text{Fe}^{3+})$ ratio has been calculated using the equation: $L\alpha/L\beta \times 0.42 + \text{Cr\#} \times 0.03 - (\text{FeO} + (\text{Fe}_2\text{O}_3/1.113)) \times 0.01$ obtained after standardisation.^b $\text{Fe}^{3+}/(\text{Fe}^{2+} + \text{Fe}^{3+})$ in spinel determined by “flank”-method.^c Relative error in % calculated as a total of measurement errors for Fe, Cr, Al and X-ray intensities at flank positions.^d T Ol-Spl = Equilibration temperature calculated using Li et al. (1995) formulation.^e Oxygen fugacity estimates are from Wood et al. (1990) and are reported as Δlog *f*O₂ from the quartz-fayalite-magnetite (FMQ) buffer using the $\text{Fe}^{3+}/\Sigma\text{Fe}$ values of the analysed spinels. See text for further details.^e Current normalised ratio of the intensities at flank position for L_α and L_β Fe lines.

pronounced for Au, with the exception of YA1, which displays a negative Pt anomaly. Pd contents are low (< 0.1 PM values) and constant for PO4, TI1, TI2, with Pd/Ir showing subchondritic ratios for all the studied samples. Positive correlations are observed between IPGE (Fig. 2a-b-c) and Pt—Ir (not shown), and, somewhat surprisingly, between IPGE and the fertility indicators (i.e. Al₂O₃ and CaO, not shown). Mass fractions of Te, Se and S are low, often close to or below the detection limit, again with the exception of the harzburgite YA1.

For all type-A harzburgites, Re concentrations are very low (about 0.02 ng/g), leading to subchondritic ¹⁸⁷Re/¹⁸⁸Os ratios (0.045 to 0.196, the latter value also reflecting the low Os concentrations in YA1). Os isotopic compositions are subchondritic to chondritic (0.1203–0.1266, corresponding $-5 \leq \gamma_{\text{Os}(53 \text{ Ma})} \leq -0.1$) and do not define any correlation with ¹⁸⁷Re/¹⁸⁸Os or incompatible element depletion indices (i.e. Al₂O₃, see Fig. 4).

Type-B harzburgites comprise very fresh samples from Kopeto (KPT2, KPT3, KPT5), Poro (PO3) and Poya (PY1) massifs. Compared

to type-A harzburgites, these samples have much lower HSE abundances and display variable and strong fractionations Fig. 3 among PGE and more incompatible chalcophile elements (Fig. 2 and 6b). In detail, Os, Ir and Pt show positive correlations (Fig. 2) and are strongly depleted compared to Ru, Rh and Pd (Os = 0.003–0.071 ppb, Ir = 0.015–0.079 ppb, with Os/Ru = 0.01–0.26 and Ru/Ir = 2.5–20). For sample PY1, Pt is enriched relative to IPGE, Rh and Pd (Pt/Rh = 7.7; Pt/Pd = 5.5). Pd, Re and S-Se-Te have similarly low normalised abundances, with chalcophile elements often close to or below the detection limit. Au exhibits positive spikes for PY1, KPT3 and PO3 samples and tends to be more enriched than similar incompatible chalcophile elements (i.e. Pd and Re).

Measured ¹⁸⁷Os/¹⁸⁸Os ratios vary from chondritic to suprachondritic (0.1273–0.1534) and are coupled with high and variable ¹⁸⁷Re/¹⁸⁸Os (1.62–32). Initial Os isotopic compositions calculated at 53 Ma range from depleted to slightly suprachondritic values (0.1181–0.1365, $-7 \leq \gamma_{\text{Os}(53 \text{ Ma})} \leq 3$).

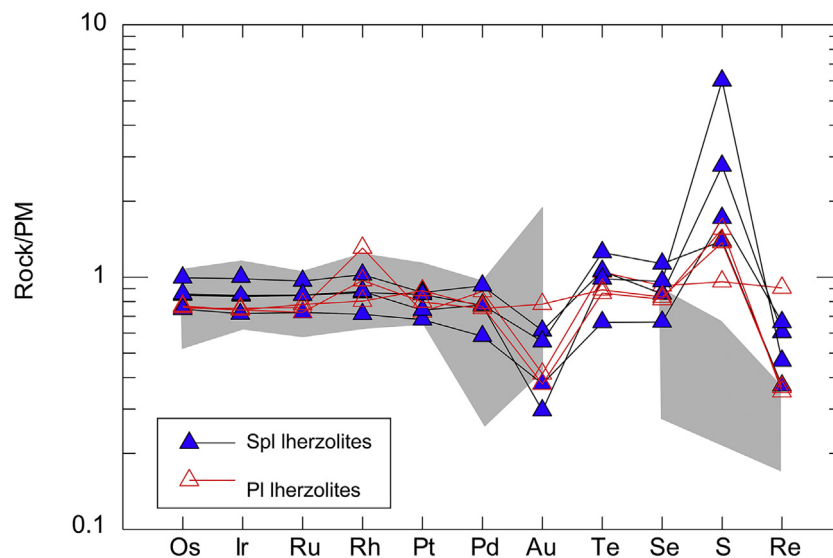


Fig. 3. Primitive mantle normalised HSE and chalcogen patterns for the New Caledonia spinel and plagioclase ilherzolites. Grey shaded area includes oceanic ilherzolites from Mid-Atlantic and South West Indian ridges (Snow and Schmidt, 1998; Luguët et al., 2001; Luguët et al., 2003) and ophiolitic ilherzolites from the Ligurian Units (Fischer-Gödde et al., 2011; Luguët et al., 2004; Snow et al., 2000). Normalising values after Becker et al. (2006), Fischer-Gödde et al. (2011) and Wang and Becker (2013).

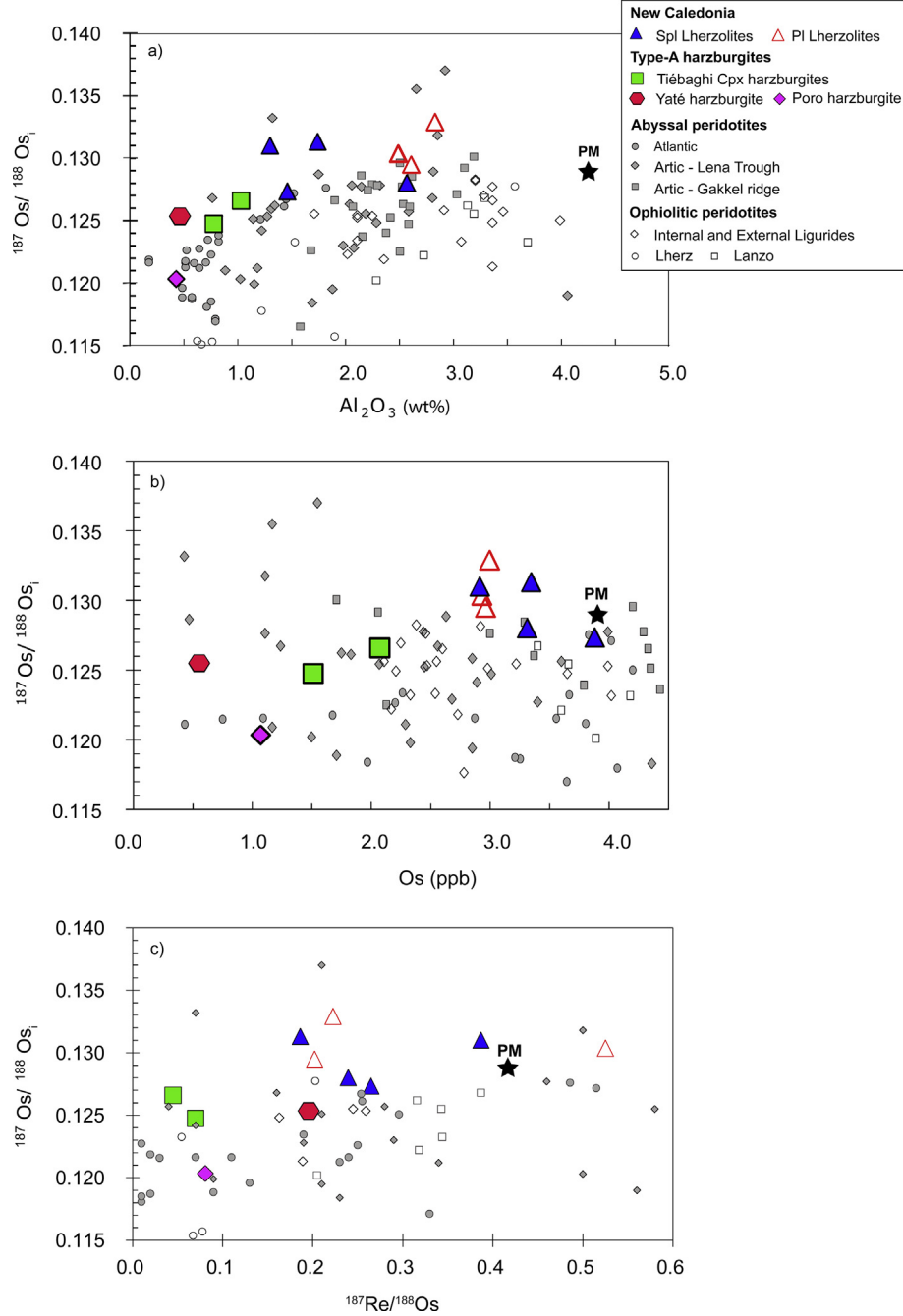


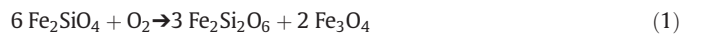
Fig. 4. a) Al_2O_3 (wt%)– $^{187}\text{Os}/^{188}\text{Os}_i$ and b) Os – $^{187}\text{Os}/^{188}\text{Os}_i$ and c) $^{187}\text{Re}/^{188}\text{Os}$ – $^{187}\text{Os}/^{188}\text{Os}_i$ diagrams showing data from New Caledonia lherzolites and type-A harzburgites in comparison to PM compositions. PM data from Meisel et al., 1996. Data for abyssal peridotites are from Harvey et al., (2006) for Atlantic peridotites, Lassiter et al., (2014) for Lena trough and Liu et al. (2015) for Gakkal ridge. See Fig. 2 for ophiolitic peridotites references.

Replicate analyses of samples PY1, KPT2, KPT5, PO3 yield quite similar results for Ru (< 5% relative deviation, except for sample KPT2) and Au (~ 6% for PY1 and KPT5) and acceptable results for Pt for PY1-KPT5 (11.5–14.0% relative deviation). Values appear much more scattered for Os and Re (RSD > 30%) and less dispersed for Rh and Pd (7 ≤ RSD% ≤ 32). The relative deviation of chalcophile elements is more limited, mostly <15%.

The large variations of mass fractions of HSE and chalcophile elements in duplicate samples reflects the very low mass fractions of these elements combined with the inhomogeneity in the distribution of HSE carrier phases in gram-size quantities of rock powder, an issue that has already been recognised in peridotitic rocks (e.g. Becker et al., 2006; Luguet et al., 2007).

4.3. Oxybarometry

Results for calculations of f_{O_2} using coexisting olivine, orthopyroxene and spinel (e.g., Wood, 1990, see Table 2 and Table S2) with the reaction are given in Table 2.



The calculation of f_{O_2} depends on the pressure and temperature of equilibrium. For this study, we used olivine-spinel geothermometry (Li et al., 1995) as this thermometer is based on the same elements and minerals (Fe and Mg content in olivine and spinel) that are used to calculate f_{O_2} . As spinel peridotites lack a good barometer, in order

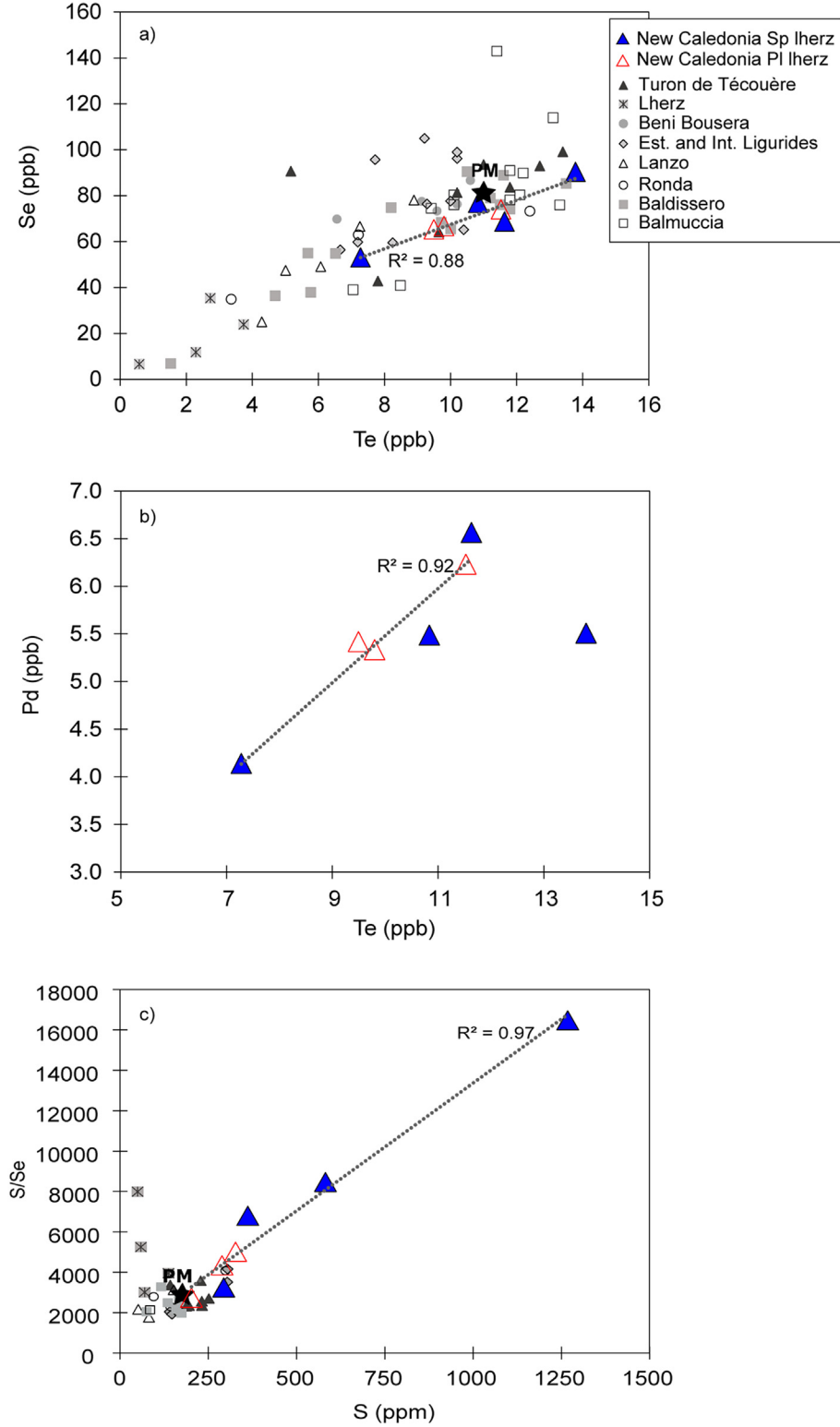


Fig. 5. a) Te vs. Se, b) Te vs. Pd and c) S vs. S/Se correlation diagrams for New Caledonia spinel and plagioclase ilherzolites. Data for orogenic ilherzolites from Wang and Becker (2013) are also plotted for comparison.

to maximise consistency with temperatures determined with Li et al. (1995) thermometer, we have assumed a pressure of 1.0 GPa.

Ferric to total iron ratios ($\text{Fe}^{3+}/\Sigma\text{Fe}$) in spinel are similar within the harzburgites, covering a limited range of values ($0.18 \leq \text{Fe}^{3+}/\text{Fe}^{3+} + \text{Fe}^{2+} \leq 0.21$, see Table 2) indicating similar f_{O_2} values for the two harzburgite sub-groups ($\Delta\log\text{FMQ} = 0.35\text{--}0.92$).

In Fig. 7a-b spinel Cr# is reported against $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratio and $\Delta\log\text{FMQ}$ for the analysed samples. Literature data for mantle peridotites from different geodynamic settings (i.e. MOR and supra-subduction zone) and arc xenoliths have also been plotted for comparison. Overall, the harzburgites share remarkable similarities with supra-subduction zone peridotites from the Izu-Bonin area (Parkinson and

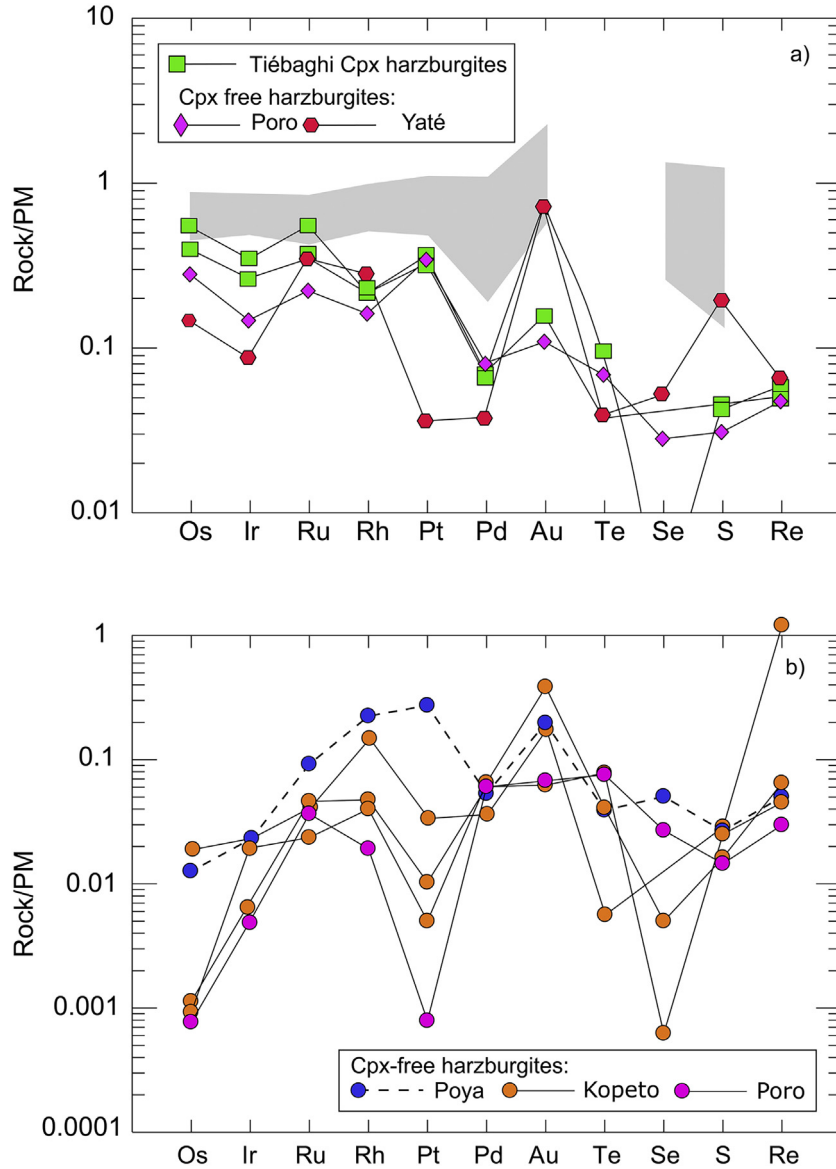


Fig. 6. a) Primitive mantle normalised HSE and chalcogen abundances in type-A harzburgites. Light grey shaded field encompasses the area of modern MOR harzburgites (Snow and Schmidt, 1998; Luguet et al., 2001, 2003; Harvey et al., 2006; Marchesi et al., 2013); b) Primitive mantle normalised diagram showing HSE and chalcogen patterns of type-B harzburgites. Normalising values are after Becker et al. (2006), Fischer-Gödde et al. (2011) and Wang and Becker (2013).

Pearce, 1998). Slightly lower Cr# values compared to the other harzburgites are displayed by T12 sample, which in turn falls between abyssal peridotites and arc xenoliths array (Fig. 7a).

5. Discussion

5.1. HSE and Re—Os systematics of the lherzolites

Major element composition and lithophile trace element chemistry of spinel lherzolites indicate moderately depleted compositions, inherited from moderate partial melting degrees (8–9%) of a DMM source, whereas plagioclase lherzolites originated through reactive melt percolation of spinel lherzolites by highly depleted, incremental melt fractions of a DMM source in the shallow lithosphere (Secchiari et al., 2016). In the following sections, the processes that may have affected HSE and Os isotopic signature of the New Caledonia lherzolites will be discussed: low temperature alteration, in particular serpentinisation, partial melting and the role of melt infiltration and chemical disequilibrium of the HSE in mantle rocks.

5.1.1. Effects of serpentinisation on HSE and $^{187}\text{Os}/^{188}\text{Os}$

Serpentinisation is a widespread process of hydrothermal alteration in ultramafic lithologies. However, its influence on HSE behaviour has not been investigated with much detail, despite some authors have proposed it as a possible cause for ^{187}Os ingrowth and Re variations in the upper mantle (Snow and Reisberg, 1995; Standish et al., 2002; Walker et al., 1996). Recent experimental studies have shown that during serpentinisation the formation of secondary sulphides, Fe—Ni alloys and native metals (Au—Cu) is promoted by reducing f_{O_2} conditions (Klein and Bach, 2009; Foustoukos et al., 2015) and thus, with the exception of Au, HSE may be retained in the host rock. Comparison of partially serpentinised and unserpentinised peridotites displaying similar major element features supports the notion that at least PGE ratios are not significantly affected by moderate to strong serpentinisation (Becker and Dale, 2016).

The New Caledonia lherzolites underwent intermediate serpentinisation degrees (LOI = 6.4 to 10.7%, see paragraph 2.1 and Table 1), which had limited effects on the budget of fluid immobile moderately incompatible lithophile trace elements in these rocks

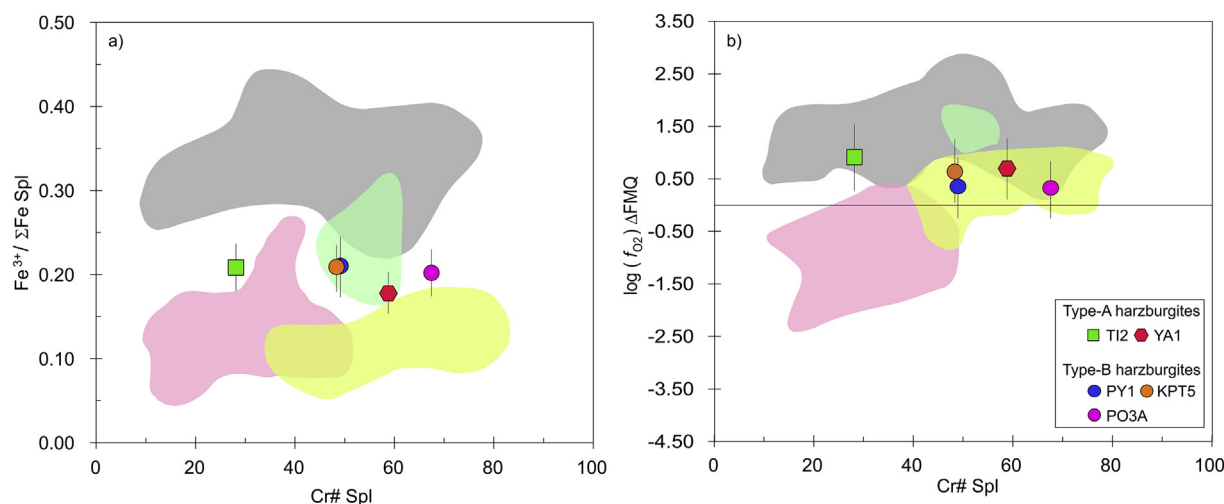


Fig. 7. a) $\text{Fe}^{3+}/\Sigma\text{Fe}$ ratios in spinel and b) $\log(f_{\text{O}_2}) \Delta\text{FMQ}$ vs Cr# in spinel for samples from this study. Error bars in a = relative error expressed as % (see Table 2). Error for $\log(f_{\text{O}_2}) \Delta\text{FMQ}$ includes error related to f_{O_2} sensor (0.5 log units) plus uncertainties derived from $\text{Fe}^{3+}/\text{Fe}^{\text{total}}$ ratio measurements (see Table 2). Background fields represent literature data for mantle peridotites: pink = abyssal peridotites (Bryndzia and Wood, 1990); grey = arc xenoliths (Brandon and Draper, 1996; Parkinson et al., 2003; Wood and Virgo, 1989); light green and light yellow = Toroshima and Conical seamount respectively (Parkinson and Pearce, 1998). Modified after Birner et al. (2017). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

(Secchiari et al., 2016). Notably, PGE contents and ratios in the lherzolites are similar to other unaltered and serpentinised lherzolites from the modern oceans and ophiolitic complexes (see Figs. 2 and 3 e.g.; Snow et al., 2000; Luguët et al., 2001, 2004; Pearson et al., 2004; Alard et al., 2005; Becker et al., 2006; Fischer-Gödde et al., 2011; Becker and Dale, 2016). This observation supports the hypothesis that PGE abundances are comparable in fresh and variably serpentinised ultramafic rocks (Becker et al., 2006; Becker and Dale, 2016; Fischer-Gödde et al., 2011; Liu et al., 2009; Marchesi et al., 2013), implying that serpentinisation results in minor changes in PGE ratios.

By contrast, the possible influence of serpentinisation on Au and Re is more difficult to evaluate, as no study has systematically investigated its effect on the behaviour of the aforementioned elements. In the lherzolites from New Caledonia, Au displays similar normalised concentrations as Re and, with the exception of a few samples, both elements are depleted relative to Pd, Te and Se. Au abundances tend to be somewhat lower than abundances in other lherzolites with similar major element composition (Fig. 3). Although the compositions can be entirely explained by magmatic fractionation processes (see subsequent chapters), minor losses of Au due to hydrothermal alteration cannot be ruled out (e.g. Lorand et al., 1999). The lack of correlation with Al_2O_3 and Yb_N (not shown) could be a hint that Au abundances may have been affected by a combination of magmatic processes and serpentinisation (i.e. Fischer-Gödde et al., 2011). Rhenium is slightly depleted compared to the PGE for most of the studied lherzolites, but displays higher concentrations than other mantle lherzolites (Fig. 3). In addition, Re contents do not correlate with LOI and Re/Os ratios cover the range generally reported for moderately depleted mantle rocks. The samples with the lowest Re contents display the highest LOI values, suggesting that no significant quantities of Re were added during the interaction with seawater during serpentinisation, as also supported by the chondritic to subchondritic $^{187}\text{Re}/^{188}\text{Os}$ ratios (see Table 1).

Likewise, the chondritic to slightly suprachondritic $^{187}\text{Os}/^{188}\text{Os}$ cannot be ascribed to serpentinisation, as unrealistically high water-rock ratios ($\sim 10^3$ – 10^4) would be required in order to perturb the whole rock $^{187}\text{Os}/^{188}\text{Os}$ at the % level or higher (e.g. Becker and Dale, 2016). The lherzolite data also show mass fractions of Se and Te and Se/Te that are similar to values in unserpentinised lherzolites (e.g., Wang and Becker, 2013). In contrast, sulphur in most lherzolites from New Caledonia shows much higher concentrations than typical for peridotites, which is readily explained by contamination with seawater-derived sulphur during serpentinisation.

We thus conclude that the HSE (perhaps with the exception of Au), Se, Te and Re—Os signature of the lherzolites offer no evidence to support that serpentinisation and associated reactions affected these elements in a noticeable way.

5.1.2. Partial melting and chemical disequilibrium of the HSE in the mantle

Spinel and plagioclase lherzolites exhibit comparable HSE contents and patterns, similar to other lherzolites from oceanic or continental settings that underwent low to moderate degrees of melt extraction (Figs. 2 and 3).

Partial melting has often been invoked as a possible cause for HSE and $^{187}\text{Os}/^{188}\text{Os}$ variations in mantle rocks (e.g. Meisel et al., 2001; Reisberg and Lorand, 1995). Studies of the behaviour of HSE during mantle melting and their abundances in mantle rocks have supported the hypothesis that HSE concentrations in residual peridotites result from sulphide-silicate partitioning during magmatic processes (i.e. Becker and Dale, 2016; Brenan et al., 2016 and references therein). Experimental studies have also highlighted that at temperatures relevant for mantle processes sulphide liquid and, in special cases, sulphide solid solutions, coexist in equilibrium with silicate melt, olivine, pyroxenes and an Al-rich phase (e.g. Brenan et al., 2016; Mungall and Brenan, 2014; Rehkämpfer et al., 1999). Experimentally determined sulphide melt-silicate melt partition coefficients ($D^{\text{sulph/sil}}$) for PGE have been shown to be high and constant (10^5 to 10^6 , e.g. Mungall and Brenan, 2014; Brenan et al., 2016), while Au shows slightly lower $D^{\text{sulph/sil}}$ ($\sim 10^4$). Therefore, up to moderately high degrees of melting, PGE behave as compatible elements and their inter-elemental ratios remain similar as long as sulphide is present in the mantle residue (i.e. until 20–25% of partial melting, depending on initial S content of the source, see Fonseca et al., 2011; Mungall and Brenan, 2014). By contrast, Re is much less chalcophile ($D^{\text{sulph/sil}} \sim 300$ –800, e.g. Fonseca et al., 2007) and is expected to become more quickly depleted in the residual mantle.

Major element compositions of the sulphide phase observed in the spinel lherzolites (Table S1) are consistent with a residual origin after incongruent melting processes (e.g. Bockrath et al., 2004; Ballhaus et al., 2006). The occurrence of homogeneous monosulphides also suggests relatively high cooling rate after the melting event.

Overall, the studied lherzolites are characterised by flat to gently sloping PGE patterns, with similar PM-normalised abundances, no PPGE fractionation and nearly constant ratios for IPGE (i.e. Os/Ir, Ru/Ir). By contrast, Au and Re display the strongest depletion. These

features imply that HSE, with the exception of Au and Re, exhibit a similar compatible behaviour during mantle melting, as expected for low to moderate melting degrees in presence of residual sulphide melt. This observation is consistent with the previous estimates obtained through geochemical modelling (Secchiari et al., 2016) and with the occurrence of a residual subsolidus sulphide assemblage in spinel lherzolites.

Although the PGE patterns are nearly flat, with only slight depletion of Pd in a few samples, the depletion of Au and Re, the range of chondritic to slightly suprachondritic $\gamma^{187}\text{Os}_i$ and the higher mass fractions of Se and Te compared to Re and the other HSE suggest a multi-stage history of the lherzolites. Notably, $\gamma^{187}\text{Os}_i$ do not correlate with mass fractions of incompatible HSE such as Re, Re/Os nor with fertility indicators (Fig. 4), as was observed in some other suites of lherzolites (Becker and Dale, 2016).

The Os isotopic signature may be a pre-existing feature of the mantle source, i.e. already present before the recent melt extraction event (Secchiari et al., 2016), as supported by the dispersed Os isotopes-fertility indicators trends. In addition, the remarkable absence of magmatic Cu-Fe-rich sulphides (e.g. see Lorand et al., 2003) argue against a recent, post-melting sulphide addition. We thus speculate that the bulk HSE, Se, Te and Os isotope compositions of the lherzolites are the result of partial melting event which affected a mantle source previously characterised by an heterogeneous sulphide population including both residual and magmatic sulfides precipitated along grain boundaries by infiltrating melts (Burton et al., 1999; Lorand et al., 1999; Alard et al., 2000; Alard et al., 2002).

5.2. Sulphur, Se and Te behaviour in the New Caledonia lherzolites

Sulphur mass fractions are variable in the lherzolites from New Caledonia and typically much higher compared to estimates of the depleted MORB mantle source (DMM ~ 150–200 ppm, Mathez, 1976; Salters and Stracke, 2004) and unaltered lherzolites (e.g., Wang and Becker, 2013). In addition, sulphur does not correlate with fertility indicators (i.e. Al_2O_3), as commonly observed in unserpentinised mantle tectonites (e.g., Lorand and Alard, 2010; Wang and Becker, 2013). The high S concentrations and the lack of correlation with melting indicators suggest that S experienced a late addition during the evolution of these rocks. Sulphur budget of mantle peridotites can be strongly affected by seawater-rock interaction, because of the high sulphate content of seawater, leading to precipitation of hydrothermal sulphides and sulphate (Alt and Shanks, 1998). We note that the major element chemistry indicates a residual origin for the sulphide phases of the lherzolites (see paragraph 4.1 and 5.1.2). Hydrothermal sulphides or sulphates were not identified during microprobe analyses.

Recent geochemical works have demonstrated the role of serpentine as a sink of S under various oxidation states (S^{2-} , S^- , S^0 and S^{6+} , Debret et al., 2017). These studies have shown that S concentrations can be anomalously high in serpentinised peridotites (up to 1 wt%, see Alt and Shanks, 2003), as S can be accommodated in serpentine minerals, accounting from 60 to 100% of the sulphur budget of the peridotite (Debret et al., 2017). In situ XANES analyses have also revealed that S can be hosted in nano-phases associated with serpentine or trapped either via Si substitutions in the tetrahedra, or as a sulphate ion in the network of the tetrahedral sheet of serpentine minerals (Debret et al., 2017).

The addition of S during serpentinisation is also reflected in the high suprachondritic S/Se ratios (up to 16,500) and the excellent correlation observed between S concentrations and S/Se ratios (see Fig. 5c). Despite the strong S enrichments, Se and Te display 'normal' concentrations and Se/Te ratios are in the range of other lherzolites (see Wang and Becker, 2013). These data confirm that Se—Te contents and ratios were not significantly impacted by serpentinisation, as previously observed for other peridotites that experienced low to moderate serpentinisation degrees (e.g. Marchesi et al., 2013; Wang and Becker, 2013).

Moreover, Se—Te show a good correlation between each other (Fig. 5a), implying that they are controlled by the same mineral phases. In mantle peridotites, Se can replace S as a chalcogen anion within the crystalline structure of sulphides (e.g. Bulanova et al., 1996; Hattori et al., 2002; Helmy et al., 2010) or can form Se-rich micro phases, while Tellurium, owing to its semi metal nature, tends to partition between sulphides and late exsolved micrometric tellurides (Pt, Pd, Te, As, Bi phases). The latter are thought to crystallise at low temperatures during cooling, once sulphide melt becomes saturated with respect to Te (Lorand et al., 2008; Lorand and Alard, 2010; Luguet et al., 2004). In the lherzolites, Se and Te do not correlate with melting indicators, but Te displays a rough positive correlation with Pd (Fig. 5b), which suggests that the sulphide melt-bulk silicate partition coefficient of Te should be between Se and Pd (e.g., Figs. 3, 4, 5).

5.3. Type-A harzburgites: highly siderophile element systematics of a residual sub-arc mantle section

The distinct HSE patterns and Os isotopic signature testify that the New Caledonia harzburgites recorded a different evolution compared to the northern lherzolites.

Three of the four samples belonging to the sub-group A (TI1, TI2, PO4) show similar HSE patterns and chalcophile elements depletion, suggesting that the same processes contributed to the HSE and chalcophile element budget of these rocks. The low chalcophile element concentrations, close to or below the detection limit, coupled with low Pd/Ir ratios and subchondritic $^{187}\text{Os}/^{188}\text{Os}_i$, point out that type-A harzburgites are residues after high degree of melt extraction, where sulphides melts must have been nearly completely dissolved in the coexisting silicate melt.

Experimental studies have predicted a compatible behaviour for all the PGE during mantle melting as long as the sulphide phase is retained in the peridotite (Mungall and Brenan, 2014). Depending on the initial S content of the mantle rocks, ~17–20% of melting is required for sulphide exhaustion (Fonseca et al., 2011; Mungall and Brenan, 2014). As melting proceeds, sulphides are progressively dissolved into the melt and the PGE concentrate in the residual sulphide melt (Mungall and Brenan, 2014). Assuming that chemical equilibrium is reached, subsequent melt increments should lead to a slightly increase of whole rock PGE contents, leaving elemental ratios almost constant. At the point when sulphide is completely removed, IPGE and Pt are accommodated in metallic alloys, while Re, Au and Pd mass fraction should become extremely low, as these elements are not hosted in any residual mantle phase (Mungall and Brenan, 2014). Hence, for high melting degrees, HSE abundances in the residue should reflect mineral-melt partitioning and the P-T and f_{O_2} -dependent solubility of Pt and IPGE alloys in silicate melt (Brenan et al., 2016; Fonseca et al., 2011, 2012; Mungall and Brenan, 2014).

The IPGE-PPGE fractionation and the resolvable fractionations between specific PGE displayed by the type-A harzburgites bear witness of high melt extraction degrees, which resulted in the formation of a S-free mantle residue. The fractionated Os-Ir-Ru-Rh segments in the HSE patterns and the positive Pt anomalies are likely carried by tiny residual sulphides (i.e. laurite) and metallic alloys (Os—Ir and Pt—Ir, see Lorand et al., 1999; Luguet et al., 2001, 2007). These latter are thought to precipitate from sulphide melt shortly before the complete exhaustion of sulphide (Mungall and Brenan, 2014) or immediately after sulphide consumption, due to f_{S_2} lowering and diminished metal-sulphide complexation in the silicate melt (Fonseca et al., 2012). The variable but broadly systematic IPGE inter-elemental fractionation (high Os/Ir, Ru/Ir and Ru/Rh) and the occurrence of positive Pt anomalies possibly suggest the presence of different residual Ir—Pt alloy proportions and preferred Os—Ru retention compared to Ir in the residual PGE alloys (e.g. Brenan and Andrews, 2001; Fonseca et al., 2012).

The HSE fractionations observed for type-A harzburgites are different from HSE patterns of modern harzburgites in MOR environments (Fig. 6a), as the latter are characterised by flat or weakly-fractionated Os-Ir-Ru triplet, rarely displaying positive Pt spikes (Snow and Schmidt, 1998; Luguët et al., 2001, 2003). By contrast, HSE elemental fractionations of type-A harzburgites resemble those observed for some arc xenoliths (Liu et al., 2015; Saha et al., 2005; Scott et al., 2019) or ophiolitic peridotites bearing a supra-subduction zone affinity (see Büchl et al., 2004, 2002; O'Driscoll et al., 2012). Notably, similar HSE fractionations have also been reported for mantle xenoliths from the Chatam Islands (New Zealand, see Pearson et al., 2004).

Accordingly, trace element geochemical modelling has shown that the extreme depletion in trace element contents displayed by the New Caledonia harzburgites was achieved through a polyphase evolution, including a first melting event in a mid-ocean ridge setting, followed by fluid-assisted melting reaching clinopyroxene exhaustion in a subduction system (see Secchiari et al., 2020). Such high melting degrees are permissible in supra-subduction zone environments, where hydrous conditions at relatively low pressures can produce melt fractions substantially exceeding 20% without invoking extremely high temperature (e.g. see Ulmer, 2001). In addition, LREE and FME (Sr, Ba, Pb) enrichments coupled with variable Pb isotope compositions of the type-A harzburgites may be explained by *syn*- and post-melting interactions with different subduction-related components (i.e. aqueous fluids and melts originated in the forearc setting, see Secchiari et al., 2019, 2020).

We thus conclude that the HSE and chalcophile element signature displayed by TI1, TI2 and PO4 predominantly reflect high degrees of melt extraction in a supra-subduction zone environment. The positive Pt spikes suggest that Pt-rich alloys were stable in the mantle residue and were only in part dissolved in the melt during melt extraction.

The enrichments of Au are modest (0.2–1.3 ng/g Au) and may be related to fluid overprint, either from slab-derived fluids (Kepezhinskas et al., 2002; McInnes et al., 1999) or from low-T alteration, e.g. during obduction (e.g. Snow et al., 2000).

The harzburgite YA1 shows higher S and Se concentrations (Fig. 5a), which, considering the significant LOI value of 6.83 wt%, could be related to serpentinisation and precipitation of secondary sulphides. The strongly fractionated HSE pattern and the low concentrations of the incompatible HSE (i.e. Pd, Re) indicate that the HSE budget of YA1 is also governed by melting, as for TI1, TI2 and PO4 harzburgites. The low Os, Ir and Pt concentrations, are much closer to the values reported for type-B harzburgites (see Table 1 and Fig. 5b). Sample YA1 can thus be seen as transitional between type-A and type-B sub-group.

5.4. Origin of type-B harzburgites – strong depletion followed by subduction zone metasomatism?

Type-B harzburgites mostly occur in the central massifs, however one sample (PO3) has also been identified in the eastern zone, close to the area where one type-A harzburgite (PO4) was sampled. Type-B harzburgites display remarkably different HSE and Re/Os behaviour compared to type-A sub-group (Figs. 2 and 6).

Type-B harzburgites show low abundances of incompatible chalcophile elements, i.e. Pd, S, Se and Re, with values in the range of type-A harzburgites. In principle such low concentrations could be reconciled with high melting degrees and sulphide exhaustion in the mantle source. The strong depletion of Os, Ir and Pt, (Fig. 6b) coupled with slightly subchondritic to suprachondritic $^{187}\text{Re}/^{188}\text{Os}$ (0.23 to 37) and $^{187}\text{Os}/^{188}\text{Os}$ (0.1239 to 0.302) are remarkable and do not occur in residual harzburgites from convecting mantle, as represented by abyssal peridotites. By contrast, type-B harzburgites share some similarities with mantle xenoliths from arc settings, such as low Os contents associated with chondritic to suprachondritic $^{187}\text{Os}/^{188}\text{Os}$ (Brandon et al., 1996, 1999; Saha et al., 2005; Widom, 2011). These features have been ascribed to interactions with subduction zone fluids (i.e. fluids from

subducted altered oceanic crust and/or its sedimentary cover), which may have induced sulphide breakdown, locally overprinting the Os isotopic signature of the mantle (Widom et al., 2003). Such qualitative observations have also been supported by experimental works, that highlighted the critical influence of oxygen fugacity (f_{O_2}) on sulphide and alloy stability (e.g. Andrews and Brenan, 2002; Fonseca et al., 2011, 2012; Mungall and Brenan, 2014). In addition, the strongly fractionated, IPGE-depleted, HSE patterns of the type-B harzburgites closely resemble those observed in some refractory harzburgites and replacive dunites from ophiolitic complexes that underwent interaction with S-undersaturated melts (Büchl et al., 2002; Lorand et al., 2004).

As metasomatism by subduction fluids and hydrous melts has been proposed for the New Caledonia harzburgites based on isotopic and incompatible element studies, we have determined oxygen fugacities on a set of five harzburgites, in order to test if the HSE signature of the two sub-groups could reflect different oxygen fugacity conditions.

As a whole, the harzburgites in our dataset bear witness of similar oxygen fugacity conditions, showing no significant difference among sub-types A and B. Both groups register oxygen fugacity values close to or only slightly higher than the FMQ buffer, displaying a nearly horizontal trend in the Cr# vs $\log(f_{\text{O}_2})$ ΔFMQ variation diagram (Fig. 7b).

Partial melting of a spinel peridotite source is expected to induce progressive $\text{Fe}^{3+}/\Sigma\text{Fe}$ lowering in the residual mantle, due to the preferential removal of Fe^{3+} in Al-rich spinel and clinopyroxene during melt extraction (Woodland et al., 2006). Thus, as melting proceeds and the aforementioned phases are removed from the mantle assemblage, melt extraction in un-buffered conditions should result in f_{O_2} lowering in the residual mantle, generating negative Cr# vs. $\log(f_{\text{O}_2})$ ΔFMQ trends (e.g. Brandon and Draper, 1996). However, these correlations are rarely preserved in mantle peridotites, as subsequent geochemical modifications, e.g. metasomatism, tend to decouple f_{O_2} from melting depletion indexes. In particular, in the sub-arc region, interaction of the refractory peridotite with subduction components is expected to shift the oxygen fugacity state towards higher values, due to the high oxidation capacity of slab-derived fluids and melts (e.g. Parkinson and Arculus, 1999; Brandon and Draper, 1996; Parkinson et al., 2003).

The subhorizontal trends observed in Fig. 7a–b, as well as the slightly oxidised f_{O_2} values, thus bear evidence that oxygen fugacity was modified during melting and/or post-melting evolution of the New Caledonia harzburgites. The measured f_{O_2} are similar to those observed for the less oxidised peridotites from the Izu-Bonin region, plotting at the lower end of the arc xenoliths domain (see Fig. 7b; Wood and Virgo, 1989; Brandon and Draper, 1996; Parkinson et al., 2003). Such oxygen fugacity values most likely reflect limited interaction with slab-derived fluids and melts, in agreement with the geodynamic scenario proposed for the New Caledonia archipelago and the previous geochemical models (see Secchiari et al., 2020). In fact, Eocene subduction is believed to have started close to or in correspondence of an active oceanic spreading center, where hot and young (~6–9 My old, Cluzel et al., 2016) lithosphere was forced to subduct. In such a context, fluid fluxes from the downgoing slab must have been limited, due to the young age of the subducted material and the intra-oceanic nature of the subduction (Cluzel et al., 2016). In addition, post melting metasomatism involved small fractions (0.5–1%) of depleted (boninitic) melts, which may have not been able to shift mantle oxidation state towards more oxidised conditions. In fact, while large oxygen fugacity variations in mantle peridotites are permissible for low Fe^{3+} contents in spinel, more and more Fe^{3+} has to be added to cause even a small increase in f_{O_2} when f_{O_2} approaches FMQ buffer (Woodland et al., 2006).

The similar f_{O_2} values registered by the two harzburgite sub-types attest that HSE behaviour was not critically influenced by different oxygen fugacity conditions. This observation supports the hypothesis that the two types of patterns are not directly linked to a specific process (i.e. higher melt extraction or enrichment degree or interaction with an oxidised component). Rather, the occurrence of two distinct patterns in the harzburgites may reflect an effect of source-control on the HSE

behaviour during the recent evolution, indicating the presence of geochemically heterogeneous mantle domains below the New Caledonia archipelago before Eocene subduction.

5.5. Type-B harzburgites: a broader perspective

Despite being similar in terms of chemistry or mineralogy, type-A and B harzburgites possess distinct HSE signatures. In addition, the HSE signatures of type-B harzburgites have not yet been identified in other mantle tectonites, either from modern oceanic lithosphere or ophiolitic complex.

However, similar compositions have been recently reported for some moderately depleted to highly refractory peridotites and mantle xenoliths from New Zealand (Scott et al., 2019). The New Zealand mantle is composed of isotopically heterogeneous mantle fragments with evolutionary histories extending over 2.75 Ga (Os model ages = 0.1–2.75 Ga, with a broad peak at 1.2 Ga), and PGE systematics decoupled from major element compositions (see Liu et al., 2015; Scott et al., 2019). These features have been explained by accretion of Zealandia lithospheric mantle from amalgamation of genetically

unrelated convecting mantle fragments which were swept together beneath the Gondwana subduction margin, variably re-melted and laterally accreted (Scott et al., 2019).

Among the New Zealand peridotite suites, mantle xenoliths from Lake Moana and Chatam Island show HSE patterns that are similar to our dataset (Fig. 8). Lake Moana Cretaceous xenoliths include cpx-free harzburgites that experienced up to 30% melting, while Eocene-aged Chatam Island harzburgites exhibit a less refractory nature, as attested by the presence of primary clinopyroxene (up to 1.8% modal, Scott et al., 2016). Such depletion degrees were achieved either by plume melting or hydrous melting in an arc setting for the Lake Moana xenoliths, whereas Chatam Island samples are thought to represent fragments of fore-arc lithosphere (Scott et al., 2016, 2019).

Overall, HSE diagrams highlight that New Zealand mantle xenoliths reproduce with good approximation both patterns observed in our harzburgites, namely type-A and type-B (Fig. 8). IPGE patterns are broadly sub-parallel, with the New Caledonia harzburgites falling within or at the lower range of values displayed by the New Zealand samples, whereas Pt, Pd and Re exhibit greater variability.

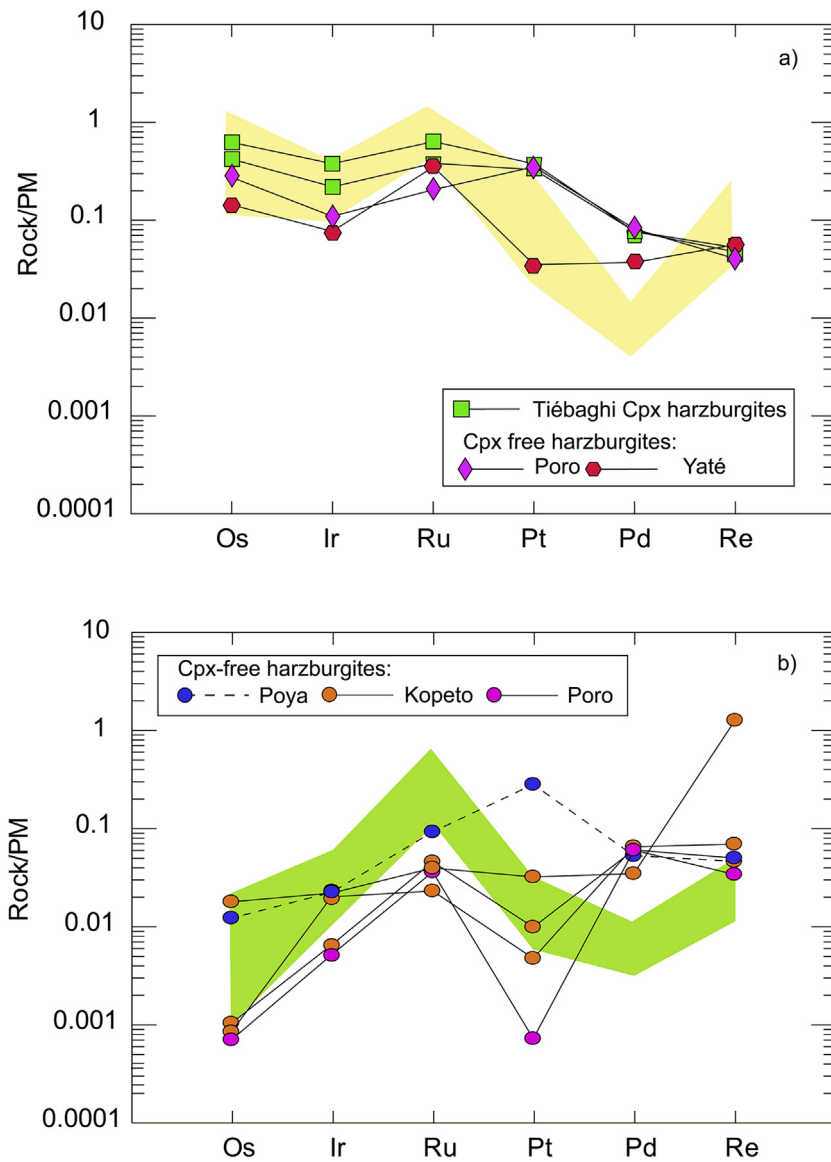


Fig. 8. Primitive mantle normalised HSE abundances of a) type-A and b) type-B harzburgites compared to the HSE composition displayed by New Zealand mantle xenoliths (Scott et al., 2019). See text for further detail.

For most of the New Zealand xenoliths, Ru concentrations are higher compared to the contents in the New Caledonia harzburgites. Higher Ru contents cannot be ascribed to different degrees of melting, as fertility indicators (i.e. Mg#(Ol), Cr#(Spl)) indicate comparable depletion degrees for both peridotite suites (see Scott et al., 2016, 2019; Secchiari et al., 2020). The increased Ru retention in the New Zealand samples may be reconciled to the higher Cr-spinel content of these lithologies (up to 2.8%, see Scott et al., 2016, vs. up to 0.8% for our harzburgites). Numerous studies have in fact demonstrated that spinel can be a significant host for Ru ($D^{\text{spinel/melt}} \sim 20$, Capobianco and Drake, 1990). In addition, increase in oxygen fugacity markedly enhances Ru compatibility in Cr-rich spinel ($D^{\text{spinel/melt}}$ up to 500 for f_{O_2} of -0.5 FMQ, Park et al., 2012), which can accommodate Ru within its crystal lattice (Pagé and Barnes, 2016) or as laurite and/or Ru-rich alloy inclusions (Brenan and Andrews, 2001). Recent geochemical works have also illustrated the importance of Ru retention in the sub-arc mantle for the HSE signature of arc lavas (Dale et al., 2012; Park et al., 2013). These studies explain the low Ru concentrations and the high Pt/Ru ratios of the volcanic products as related to Ru retention in the mantle source due to the presence of Cr-rich spinel or PGM (see Dale et al., 2012; Park et al., 2013). Likewise, positive Ru anomalies in our type-B harzburgites may reflect the presence of small laurite or Ru-rich phase inclusions, which could have escaped the high melting degrees due to the shield effect of spinel.

Other HSE (e.g. Pd and Re) in the New Caledonia harzburgites display a wider range of values compared to the New Zealand samples. Enrichments in Pd have also been recognised in other sub-arc mantle sections, where they have been attributed to slab-derived fluids metasomatism, due to the high solubility of Pd in aqueous fluids (McInnes et al., 1999; Park et al., 2013). Likewise, Re addition in mantle wedge peridotites may be due to Re release in slab derived fluids during dehydration of the mafic portion of the subducting slab (see Dale et al., 2009). Hence, we conclude that Pd—Re (as well as Te—Se—S) re-enrichments in type-B harzburgites may have been facilitated by minor sulphide precipitation from slab-derived fluids or melts.

By contrast, the widespread negative Pt anomalies may reflect destabilisation of a pre-existing Pt-alloy phase, possibly related to melting after alloy saturation (see Mungall and Brenan, 2016), and Pt release into the melt.

5.6. Inferences from Re—Os systematics of the New Caledonia harzburgites

All type-A harzburgites, including YA1, show unradiogenic $^{187}\text{Os}/^{188}\text{Os}_i$ and very low Re/Os relative to the range of chondritic values (Walker et al., 2002) or primitive mantle estimates (Meisel et al., 1996). The γ_{Os_i} overlap with data from depleted abyssal peridotites and mantle sections from some ophiolites (e.g., Becker and Dale, 2016).

In order to obtain an estimate of the time of melt depletion, we calculated Re—Os model ages (T_{MA}) and Re—depletion model ages (T_{RD}). T_{RD} is generally used to estimate model ages of mantle peridotites that experienced high degrees of melt extraction, leading to low Re contents and low Re/Os ratios. For mantle peridotites which are pure residues of high partial melting degrees sufficient to remove all Re, then T_{MA} and T_{RD} ages should be identical. If a minor amount of Re remains, the two values should be relatively similar, with T_{RD} providing a minimum depletion age.

The Re-depletion ages for type-A harzburgites are quite homogeneous for three of our samples (T11, T12, YA1) ranging between 0.4 and 0.7 Ga (see Table 1), while sample PO4 yields an older Re-depletion age of 1.3 Ga. Slightly older ages are provided by T_{MA} (0.5–1.0 Ga for T11, T12, YA1 and 1.5 Ga for PO4), indicating that Re was not effectively removed after partial melting.

The younger model ages recorded by the harzburgites are likely linked to the evolution of the New Caledonia mantle in relationship to the eastern Australian margin, from which the New Caledonia archipelago was separated via marginal rifting about 90 Ma ago (Cluzel et al.,

2001, 2012; Whattam, 2009). The ancient Re depletion age of the harzburgite PO4, on the other hand, reflects a mantle domain characterised by long-term low Re/Os ratio. This age is also mirrored by Nd isotopic signature, which shows a highly radiogenic value ($\varepsilon_{\text{Nd}_i} = +13.32$, Secchiari et al., 2020), indicative of a mantle reservoir that underwent long-term depletion of Nd.

In addition, similar depletion ages (i.e. 1.2 Ga) occur in the New Zealand mantle rocks (see Scott et al., 2019).

These results are consistent with recent Re—Os studies on abyssal peridotites and mantle tectonites from ophiolitic sequences showing that the convecting mantle contains harzburgite domains that underwent depletion events much older than the age of peridotite processing under the ridge (e.g., Harvey et al., 2006).

In summary, the New Caledonia harzburgites show Re—Os systematics typical of depleted upper mantle, showing Early Paleozoic Os isotopic equilibration and evidence for ancient depletion events (> 1.0 Ga). This multi-stage history led to the depleted nature of the harzburgites and possibly resulted in their complex HSE geochemical signatures.

6. Summary and conclusions

A Re—Os isotopes, highly siderophile and chalcophile element investigation of the New Caledonia peridotites was carried out to unravel the behaviour of the aforementioned elements in lherzolites and harzburgites from the New Caledonia ophiolite.

The lherzolites exhibit subchondritic to slightly suprachondritic $^{187}\text{Os}/^{188}\text{Os}_i$ (0.1273–0.1329). PM-normalised HSE abundance diagrams are characterised by gently sloping patterns showing increasing depletion towards Re—Au, similar to lherzolites that experienced low to moderate melt extraction. However, the lack of correlation between HSE and fertility indicators, as well as the slightly suprachondritic Os isotopic ratios, argue against a simple partial melting history. Rather, such features and the presence of included and interstitial residual monosulphides possibly indicate that melting involved a mantle domain that has previously experienced a melt percolation event. The high S concentrations of the lherzolites (202–1268 ppm) most likely resulted from late-stage seawater-rock reactions.

By contrast, the New Caledonia harzburgites record higher degrees of melt extraction, as attested by the strikingly low, often below the detection limit, concentrations of incompatible chalcophile elements. Despite their homogeneity in terms of mineralogical and major element compositions, HSE patterns and Os isotopic compositions indicate the occurrence of two distinct harzburgite sub-groups.

Type-A harzburgites are characterised by steeply plunging HSE patterns, showing IPGE-PPGE and Os—Ir—Ru fractionation, coupled with low Re/Os ratios and subchondritic $^{187}\text{Os}/^{188}\text{Os}_i$. The strongly fractionated HSE patterns and the positive Pt anomalies, together with the high modelled melting degrees, indicate that melting occurred under hydrous conditions in sub-arc mantle.

Type-B harzburgites display notably different HSE patterns, showing depleted Os—Ir compared to Ru, positive anomalies and Pd—Re re-enrichments (relative to IPGE), coupled with chondritic to strongly suprachondritic measured Os isotopic ratios ($^{187}\text{Os}/^{188}\text{Os} = 0.127\text{--}0.153$). These features have not been yet identified in mantle tectonites.

The similar and slightly oxidising oxygen fugacity values recorded by the two sub-types, named type-A and B, do not convincingly support the hypothesis that type-B patterns could be ascribed to massive interactions with an oxidising agent. By contrast, this observation, coupled with the uniform mineralogical and geochemical characters of the New Caledonia harzburgites, suggest that type-B patterns were not the result of a specific depletion or enrichment process related to their recent tectonic evolution in a subduction setting. Rather, these patterns could reflect the occurrence of a geochemically heterogeneous mantle source below the New Caledonia archipelago, similar to what was speculated for the New Zealand mantle. The HSE

signature carried by the studied peridotites, as well as the puzzling similarity observed between the New Caledonia harzburgites and the New Zealand mantle xenoliths, might attest the presence of a mantle source bearing a long lasting evolution (> 1 Ga), possibly linked to the Zealandia formation.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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