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**Growth of** *Chlorella sorokiniana* **on a mixture of volatile fatty** 

## **acids: the effects of light and temperature**

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## **Abstract**

The feasibility of coupling of dark fermentation to produce biohydrogen, with

heterotrophic cultivation of microalgae, to produce biolipids from fermentation by-

products, is limited bybutyrate which inhibits the growth of themicroalgae. This study

investigated the influence of light and temperature on*Chlorella sorokiniana*grown on a

mixture of acetateand butyrate, two of the volatile fatty acids produced by dark

fermentation.Exposure to light caused autotrophicbiomass production (56% of the final

17 biomass) and reduced the time to reach butyrate exhaustion to 7 days at 25 °C from 10

18 days in the dark. For growth on acetate at the optimum temperature (35  $^{\circ}$ C), the

- presence of butyratereduced the growth rate (by 46%) and the carbon yield (by 36%).
- For successful microalgae growth on dark fermentation effluent, butyrate inhibition may

be reduced by setting the temperature to 30 °Cand providing light.

### **Keywords**

 *Chlorella sorokiniana*;Dark Fermentation; Heterotrophy; Mixotrophy; Butyrate inhibition

## **1 Introduction**

 Many studies and industrial projects have shown the value of heterotrophic cultivation of microalgae forproducing high added value compounds, such as docosahexaenoic acid (DHA), andcommodity compounds, such as lipids for biofuels (Lowrey et al., 2015). When microalgae are grown on organic carbon sources in the dark, they tend to grow faster with higher biomass and lipid yields than when they are grown usingconventional autotrophic cultivation (Liang, 2013). However, in order to reduce production costs, an alternative to glucose, the most common substrate, must be found, especially for producing biofuels (Liang, 2013). Acetate has been suggested as one of the best alternatives since it can be easily incorporated into lipids or carbohydratesby microalgaeandis widely available as a cheap source of carbon(Lowrey et al., 2015). Moreover, acetate is one of the main end-productsof microbial dark fermentation (DF)of various types of urban, agricultural and industrial waste (Ghimire et al., 2015). Recently, several studies have shown the benefits and feasibility of coupling DF, which produces hydrogen as the main product and volatile fatty acids (VFAs) as secondary metabolites, with microalgae cultivation, which produces bothmicroalgal biomass and lipids(Chandra et al., 2015; Liu et al., 2013, 2012; Ren et al., 2014; Turon et al., 2015; Venkata Mohan and Prathima Devi, 2012).During DF, complex organic compounds originating from waste are converted by anaerobic bacteria into simple VFAs, mainly acetate and butyrate, that can be further assimilated by microalgae(Ghimire et al., 2015).



 *Chlorella sorokiniana*is considered to be one of the most promising species forlipid and biomass production(Griffiths and Harrison, 2009; Lizzul et al., 2014; Zheng et al.,



89 at 25 °C (Turon et al., 2015). This study set out to determine the interaction between

these two VFAs while growing*C. sorokiniana* in presence of light and at different

91 temperatures. The effects of (i) light (with light and in the dark) (ii) temperature (25  $^{\circ}$ C,

92 30 °C, and 35 °C) and (iii) a combination of light and high temperature (35 °C) were

 testedon the growth rate and carbon yield of *C. sorokiniana* growing on a mixture of 94 acetate and butyrate at an inhibiting butyrate concentration (both at  $0.3 \text{ g}_\text{C} L^{-1}$ ). Control 95 experiments with either acetate or butyrate as single substrate  $(0.3 \text{ g}_C L^{-1})$  were also performed to give a better understanding of the interactions between acetate and butyrate uptake mechanisms.

- **2 Materials and methods**
- **2.1 Microalgae cultivation conditions**

#### *2.1.1 Chlorella sorokiniana stock cultivation conditions*

*C.sorokiniana* (CCAP 211/8K) was pre-cultivated axenically in 500 mL Erlenmeyer

flasks with a working volume of 200 mL. A modified BG11 medium was used as

described byTuron *et al*(2015). Sodium bicarbonate (10 mM) was used as an inorganic

carbon (C) source, ammonium chloride (5 mM) as a nitrogen (N) source and

dipotassium phosphate (0.31 mM) as a phosphorus (P) source. The flasks and

106 components of the medium were sterilized by autoclaving at 121<sup>o</sup>C for 20 min before

use. Before starting the experiment, the axenic culture was cultivated under autotrophic

108 conditions (light intensity of 100 µmol photons.m<sup>-2</sup>.s<sup>-1</sup>) at 25 °C for 7 days.

*2.1.2 General cultivation conditions*

110 The carbon concentration of each substrate was mainly set to 0.3  $g_C$ . L<sup>-1</sup> by adding

sodium bicarbonate, for autotrophic growth conditions, or acetic acid (glacial acetic

acid, 27221-Sigma-Aldrich®) and/or butyric acid (B103500-Sigma-Aldrich®) solutions

- at 500 mM, for heterotrophic and mixotrophic growth conditions. For somespecific
- experiments (Supplementary Information) the carbon concentration was set to 0.2  $g_C L$

115 <sup>1</sup>. As high acetate concentrations have been shown to increase the lag phase of *C*.

*sorokiniana*(Qiao et al., 2012), especially in heterotrophic conditions,relatively low

117 concentrations of acetate  $(0.3 \text{ g}_c \text{L}^{-1}$  equivalent to 0.75 g.L<sup>-1</sup> and 12.5 mM) and butyrate

118  $(0.3 \text{ g}_\text{C} \cdot \text{L}^{-1})$  equivalent to 0.55 g.L<sup>-1</sup> and 6.25 mM) were used.

The C:N:P molar ratiowas set to 48:16:1.Ammonium chloride and dipotassium

phosphate were used as N and P sources, respectively. To encourage heterotrophic

metabolism, sodium bicarbonate was not added to the media for mixotrophicand

122 heterotrophic growth conditions. Only  $CO<sub>2</sub>$  from the air dissolved in the media was

available for mixotrophic growth. To maintain the same pH throughout the experiments,

the media were buffered with 100 mM of 2-(*N*-morpholino) ethanesulfonic acid(MES).

The initial pH was set to between 6 and 6.5.Prior to sterilization using a 0.2 µm pore

<span id="page-6-0"></span>filter, the working solutions of acetate and butyrate were adjusted to pH 6.5 with NaOH.

The flasks and all components of the medium were sterilized by autoclaving at 121°C

for 20 min before use.The flasks were inoculated with *C.sorokiniana*stock cultures at

10% V/V.

*C.sorokiniana*wascultivated in 125 mL black (heterotrophy) or transparent (autotrophy

and mixotrophy) Erlenmeyer flasks containing 40 mL of medium and sealed with cotton

plugs. The flasks were incubatedin the dark (heterotrophy) or under a non-saturating

l 133 light intensity of  $123 \pm 10$  µmol photons.m<sup>-2</sup>.s<sup>-1</sup>(autotrophy and mixotrophy) (Liu et al.,

2012; Van Wagenen et al., 2014a)at different temperatures as described in sections 2.1.4

and 2.1.5. The flasks wereshakenon a rotary shaker (150 rpm) for a maximum of 10

- days until the substrate was completely exhausted. All experiments and controls were
- performedin triplicate. During the experiment, axeny was checkeddaily by DAPI

- staining and phase contrast microscopy as well as byspreading the cultures on
- ATCC5solid media (ATCC, [http://www.lgcstandards-atcc.org/\)](http://www.lgcstandards-atcc.org/).
- *2.1.3 Cultivation at 25 °C*
- <span id="page-7-0"></span>*2.1.3.1 Using DCMU to inhibit autotrophic growth*
- A stock solution of 100 mMof 3-(3,4-Dichlorophenyl)-1,1-dimethylurea (DCMU),

143 diluted in ethanol, was used at a final nontoxic concentration of 10 µMfor cultivation

under mixotrophic, heterotrophic and autotrophic conditions (Zheng et al., 2014).The

145 temperature was set to  $25^{\circ}$ C and light to 123 µmol photons.m<sup>-2</sup>.s<sup>-1</sup> when required. For

- 146 the three growth conditions, a control with no DCMU was also carried out ina single
- flask.
- *2.1.3.2 Cultivation on a mixture of VFAs in the presence of light*
- The mixotrophic growth of *Chlorella sorokiniana*on a mixture of acetate and butyrate at
- 150 25 °C was compared to the mixotrophic growth on either acetateor butyrate, as single
- substrates (acetate-control and butyrate-control) andto the autotrophic growth
- (autotrophic control).The results obtained from a predictive model, as previously
- described by Turon *et al* (2015), on VFAs in the dark at 25°C were used to make
- assumptions aboutthe heterotrophic growth(Turon et al., 2015).A Monod equation was
- used to describe the heterotrophic growth on acetate and a Haldane equation was used
- for butyrate. The diauxic growth pattern on acetate and butyrate was also included in the
- model. The acetate and butyrate concentrations tested in this study were in the range of
- concentrations used to build and validate the model. This model was developed to
- 159 predict heterotrophic growth at 25 °C on acetate, butyrate or both acetate and butyrate.
- Since the lag phase was not consideredwhen building the model, the microalgae
- biomass and the acetate and butyrate concentrations,measured at the start ofthe
- microalgal growth curve, were used to initialize the Scilab simulations
- (http://www.scilab.org).

#### *2.1.4 Heterotrophic cultivation at 30 °C and 35 °C*

The microalgae growth on acetate and butyrate, as single substrates, and on a mixture of

acetate and butyrate in the dark at 30°C and at 35°C wascompared to the heterotrophic

growth simulatedat 25°C as described in sub-section [2.1.3.2.](#page-6-0)

### *2.1.5 Cultivation at 35 °C under light*

The microalgae growth on acetate and butyrate, as single substrates, and on a

- 170 combination of acetate and butyrate under light, set to  $123 \pm 10$  µmol photons.m<sup>-2</sup>.s<sup>-1</sup> at
- 171 35°C was compared to autotrophic (with bicarbonate as the sole carbon source) and
- heterotrophic growth at 35°C (sub-section [2.1.4\)](#page-7-0), to mixotrophicgrowth at 25°C (sub-
- 173 section [2.1.3.2\)](#page-6-0)and to predicted heterotrophic growth at 25 °C (sub-section 2.1.3.2).
- **2.2 Analytical methods**

### *2.2.1 Biomass measurement*

176 The biomass growth was quantified by measuring the optical densityat 800 nm  $OD_{800}$ )

177 to minimize pigment interference (Schmidt et al., 2005). Culture samples of  $300 \mu L$ 

- were dispensed into a 96 well BD Falcon® microplate and analyzed using an
- Infinite®M200 NanoQuant spectrophotometer (Tecan). DW was determined after
- 180 filtering 15 mL of algal samples onto a pre-weighed GF/F Whatman® filter that was
- 181 then dried overnight at 105 $^{\circ}$ C. The calibration curve between DW and OD<sub>800</sub>was
- determined usingvarious dilutions of algal biomass for a wide range of dry weight
- 183 values  $(0 1.4 \text{ g.L}^{-1})$ . Three calibration curves were determined to allow for the

184 difference in microalgae cell shapes during heterotrophic and mixotrophic/autotrophic 185 cultivation(Kumar et al., 2014).The equations were:



191 The apparent growth rates,  $\mu_{app}$  (d<sup>-1</sup>), during exponential growth were calculated as 192 follows (Eq 1):

$$
\mu_{app} = \frac{\ln(B_f) - \ln(B_0)}{t_f - t_0}
$$
 Equation 1

193

194 where  $t_0$  and *t* fare the start and end of the exponential growth phase and  $B_0$  and  $B_f$  are the 195 DWs  $(g.L^{-1})$  at  $t_0$  and  $t_f$ , respectively.

196 The apparent linear production rates of biomass,  $r_{app\_lin}$  (g.L<sup>-1</sup>.d<sup>-1</sup>), during linear growth 197 were calculated as follows (Eq 2):

$$
r_{app\_lin} = \frac{B_f - B_0}{t_f - t_0}
$$
 Equation 2

198 where  $t_0$  and *t*fare the start and end of the exponential growth phase and  $B_0$  and  $B_f$  are the 199 DWs  $(g.L^{-1})$  at  $t_0$  and  $t_f$ , respectively.

200 Under mixotrophic conditions, the mixotrophic carbon yields,  $Y_{MixO}^{MixO}$  (g<sub>C</sub> of biomass per

201 g<sub>C</sub>of substrate), on acetate and butyrate separately were calculated as follows (Eq 3):

$$
Y_{Mixo}^{Mixo} = \frac{(X_f - X_0) * \alpha}{S_i}
$$
 Equation 3

202 where *Xfand X*<sub>0</sub>are the DWs (g.L<sup>-1</sup>) at the start and the end of substrate exhaustion, *a*is 203 the estimated content, 50%, of carbon in microalgae DW(Chen and Johns, 1996), 204  $S_i(g_C \cdot L^{-1})$  is the initial concentration of substrate.

205 Under mixotrophic conditions, the heterotrophic carbon yields,  $Y_{Het}^{Mixo}$  (g<sub>C</sub> of estimated 206 heterotrophic biomass per  $g_C$  of substrate), on acetate and butyrate separately were 207 calculated as follows (Eq 4):

$$
Y_{Het}^{Mixo} = \frac{(X_f - X_0 - X_{ctrl\_auto}) * \alpha}{S_i}
$$
 Equation 4

208 where *Xfand X*<sub>0</sub>are the DWs (g.L<sup>-1</sup>) at the start and the end of substrate exhaustion, 209 *Xctrl\_auto*is the DW in the strict autotrophic control at the same time as substrate 210 exhaustion, *α*is the estimated content, 50%, of carbon in microalgae DW (Chen and 211 Johns, 1996),  $S_i(g_C, L^{-1})$  is the initial concentration of substrate.

212 Under mixotrophic conditions, the fraction of mixotrophic biomass due to heterotrophic 213 growth on acetate and/or butyrate,  $X_{Het}^{Mixo}$  (%), was calculated as follows (Eq 5):

$$
X_{Het}^{Mixo} = \frac{Y_{Het}^{Mixo}}{Y_{Mixo}^{Mixo}} * 100
$$
   Equation 5

214 Under mixotrophic conditions, the fraction of mixotrophic biomass due to autotrophic 215 growth on CO<sub>2</sub>,  $X_{Auto}^{MixO}$  (%), was calculated as follows (Eq 6):

$$
X_{Auto}^{Mixo} = 100 - X_{Het}^{Mixo}
$$
 Equation 6

#### *2.2.2 Measuring organic acids*

- Volatile fatty acids (VFAs), e.g. acetate and butyrate, were quantified using a gas
- chromatograph (GC 3900 Varian) equipped with a flame ionization detectoras
- previously described by Rafrafi *et al* ( 2013).
- The errors associated with OD, DW and organic acid measurements were 2%, 6% and
- 5%, respectively.

#### **2.3 Statistical analysis**

- Pairwise comparisons of all results were performed by a one-way ANOVA and Tukey's
- post-hoc analysis. All statistical analyses were carried out using the Rcmdr package 1.9-
- 6, R version 2.15.2 (R Development Core Team, 2012).

#### **3 Results and discussion**

**3.1 Effect of light on***C. sorokiniana* **growth** 

# *3.1.1 Mixotrophic conditions: a combination of autotrophic and heterotrophic conditions*

DCMU is a specific inhibitor of electron transport between Photosystem I (PSI) and

Photosystem II (PSII). DCMU was used to estimate the growth due to heterotrophic

- metabolism only,by organic carbon fixationfrom acetate,during mixotrophic growth by
- inhibiting autotrophic inorganic carbon fixation (Li et al., 2015). DCMU inhibits the
- transport of electrons from PSII to plastoquinone which further blocks the generation of
- 236 NADPH and ATP in the chloroplast (Li et al., 2014).  $CO<sub>2</sub>$  fixation is subsequently
- hampered by the lack of both NADPH and ATP. The production of ATP via the cyclic
- electron flow in photosystem I is not affected(Li et al., 2014).



 biomass, a strict autotrophic experiment (autotrophic control) was always run in parallel to the mixotrophic experiments. This control was used to assessthe heterotrophic carbon 262 yield,  $Y_{Het}^{MixO}$ , associated with butyrate or acetate uptake during mixotrophic growth. The

 biomass reached under autotrophic conditions can be subtracted from the observed mixotrophic biomassto assess the fraction of microalgae growth due to organic carbon assimilation, as described inVan Wagenen et al. (2014a). The excess biomass due to the positive interaction between the two metabolisms was considered as a boost to the biomass generated by heterotrophic growth.

# *3.1.2 Increase in the butyrate uptake rate in the presence ofacetateunder mixotrophic conditions*

The effect of light on *C. sorokiniana*cultivatedon a mixture of acetate and butyratewas

studied. The strict autotrophic control (without organic substrate) was used to give a

better explanation forthe mixotrophic growth observed in Figure 1. During the

273 exponential phase (first two days), the apparent autotrophic growth rate was  $1.04 \pm 0.05$ 

274  $d^{-1}$  During the linear phase (from day 2 to day 8), the biomass production rate was 0.11

275  $\pm$  0.01 g.L<sup>-1</sup>.d<sup>-1</sup>. With limited light availability (low light intensities and cell self-

276 shading) or  $CO_2$  limitation (no air or additional  $CO_2$ ), the exponential growth phase in

autotrophic batch cultivation will be short and rapidly followed by linear growth

(Ogbonna et al., 1995; Smith et al., 2015). The growth rates during autotrophic growth

were consistent with previously reported results obtained under similar conditions with

*C. sorokiniana*(Kim et al., 2013; Li et al., 2013; Rosenberg et al., 2014).

During mixotrophic growth on a mixture of acetate and butyrate (Figure 1-B),

assimilation of acetate and butyrate was diauxic under mixotrophic conditions since

butyrate uptake started only afterthe acetate had been completely exhausted, as

previously observed in heterotrophic conditions, (Turon et al., 2015). Thegrowth rates

on acetate andbutyrate were, therefore, analyzed separately.

286 The growth rate on acetate was slightly higher  $(2.7 \pm 0.1 \text{ d}^{-1})$  under mixotrophic 287 conditions than estimated by modeling under heterotrophic conditions  $(2.21 d^{-1}$  - see 288 Table 1)(Turon et al., 2015). The total biomass accumulated just after acetate 289 exhaustion in mixotrophic conditionswas higher than the biomass predicted by the 290 model in heterotrophic conditions (Figure 1-B). Furthermore, the mixotrophic carbon 291 yieldon acetate,  $Y_{MixO}^{MixO}$ , (Eq 3), was almost twice as high (0.79  $\pm$  0.04 d<sup>-1</sup>) under 292 mixotrophic conditions than predicted under heterotrophic conditions  $(0.42 \text{ g}_C \cdot \text{g}_C^{-1})$ 293 (Table 1). These results confirmed that the presence of light increased both the apparent 294 growth rate and the mixotrophic carbon yield on acetate compared to those 295 underheterotrophic conditions at 25°C. Under mixotrophic conditions,the heterotrophic 296  $\cdot$  carbon yield,  $Y_{Het}^{Mixo}$  - see Eq. 4, was calculated by subtracting the carbon yield 297 forautotrophic growth (autotrophic control) from the mixotrophic carbon yield ( $Y_{Het}^{MixO}$  = 298 0.48  $\pm$  0.05 g<sub>C</sub> biomass per g<sub>C</sub> acetate, see Table 1). Wherethere was uptake of both 299 organic and inorganic carbon, only 39% of the microalgal biomass obtained after 300 acetate exhaustion was due to  $CO_2$  assimilation ( $X_{Aut_0}^{Mix_0}$ , see Eq 6 and Table 1). In the 301 acetatecontrol (with no butyrate), the fraction of biomass due to  $CO<sub>2</sub>$  assimilation 302 ( $X_{Auto}^{MixO}$ , 30%) was statistically similar ( $p$ >0.05) (see Table 1 and Supplementary 303 material Figure S1) but the mixotrophic growth rate on acetate reached 4.1  $\pm$  0.4 d<sup>-1</sup>. 304 When using mixtures of VFAs, there may be a high ATP demand to deal with the 305 inhibitory effects of butyrate, such as cytosolic pH acidification, resulting in lower ATP 306 availability for fast growth on acetate (Tromballa, 1978). In conclusion, the growth rate 307 and carbon yield on acetate were higherin the presence of light than under heterotrophic 308 conditions, suggesting that the mixotrophic growth on acetate probably relied on a 309 synergy between heterotrophicand autotrophic conditions.

 After a one-day delay after the acetate had been completely exhausted, there was linear butyrate uptake during the linear growth phase (Figure 1-B).Butyrate exhaustion in mixotrophic conditions was 3 days shorter than predicted for heterotrophic conditions 313 (Figure 1-B). Based on the difference between the mixotrophic ( $Y_{MixO}^{MixO}$ , Eq 3), and 314 heterotrophic ( $Y_{Het}^{MixO}$ , Eq 4) carbon yields on butyrate, 62% of the biomass reached after 315 butyrate exhaustion was probably due to  $CO_2$  assimilation ( $X_{Auto}^{MixO}$ , see Eq 6 and Table 1). Similarly, in the butyratecontrol (without acetate - see Figure 1-C), 74% of the biomass obtained after butyrate exhaustion was probably due to  $CO<sub>2</sub>$  assimilation  $(X_{Auto}^{MixO} - \text{seeTable 1}).$  The model predicted thatat 25 °C no heterotrophic growth would have been observed at such initial butyrate concentration (with no acetate - see Figure 1- C). Furthermore,the linear butyrate uptake rate measuredafter acetate exhaustion was 1.5 times higher than measured for the butyratecontrol. It can, therefore, be concluded that mixotrophic conditionscan substantially accelerate the apparent butyrate uptake 323 through the production of algal biomass by  $CO<sub>2</sub>$  fixation.

## **3.2 Effect of temperature on heterotrophic growth on VFAs**

# *3.2.1 Inhibition by butyrate on heterotrophic growth on acetate at high temperature (35°C)*

- 
- *C.sorokiniana* was grown heterotrophically on acetate as a single substrate (acetate control), on butyrate as single substrate (butyrate control) and on a mixture of acetate and butyrate, at 35°Cknown to be the optimum temperature (Janssen et al., 1999; Li et al., 2014; Van Wagenen et al., 2014b). On acetate(Supplementary material, Figure S2), 331 the heterotrophic growth rate reached 5.88  $d^{-1}$  which was consistent with previously reported values at 35-37°C (Van Wagenen et al., 2014b).

 Forheterotrophic growth on a mixture of acetate and butyrate (Figure 2-A), the apparent 334 growth rate on acetate, at  $35^{\circ}$ C (3.17  $\pm$  0.45 d<sup>-1</sup>) was higher than at  $25^{\circ}$ C (2.23 d<sup>-1</sup>- see Table 2). However, microalgae biomass concentrationsafter acetate exhaustionwere 336 similar at 25 $^{\circ}$ C and 35 $^{\circ}$ C (Figure 2). The carbon yields on acetate at 25 $^{\circ}$ C and at 35  $^{\circ}$ C were also similar (Table 2).However, the growth rate and carbon yield on acetate in the acetatecontrol (Supplementary material, Figure S2) were almost 2 and 1.6 times higher than on the mixture of acetate and butyrate(Table 2). Even though the growth rate on 340 acetate was highest at 35  $\degree$ C in the acetatecontrol, the presence of butyrate inhibited the 341 increase growth rate on acetate at the higher temperature. At  $25 \degree C$ , the presence of butyrate did not reduce the growth rate on acetate for butyrate concentrationsup to 0.5  $\text{g}_c \text{L}^{-1}$ (Turon et al., 2015).Ugwu et al (2000) reported that when one abiotic parameter (irradiance) was set to the optimum, the negative effects of another parameter (such as high dissolved oxygen concentration or temperature) wereaggravated(Ugwu et al., 2007). Thus, when one growth factor is set at its optimum, the fast metabolism will, in particular, reduce energy storage and the microalgae might be less able to protect themselves from any adverse conditions. The negative effect of butyrate on heterotrophic growth on acetate at 35°C was reduced when the butyrate concentration 350 was lowered to 0.2  $g_C L^{-1}$  (Supplementary material Figure S4). At this concentration, 351 the growth rate  $(4.71 \pm 0.24 \text{ d}^{\text{-1}})$  and carbon yield  $(0.65 \pm 0.02 \text{ g}_C \text{g}_C^{\text{-1}})$  on acetate were 352 higher than with  $0.3 \text{ g}_\text{C} L^{-1}$  of butyrate. As a consequence, these results confirmed that butyrate inhibition ofheterotrophic growth depended on theconcentration,as previously suggested (Liu et al., 2012; Turon et al., 2015).

355 The apparent growth rate on butyrate was lower at 35 °C (0.11 d<sup>-1</sup>) than the maximum 356 growth rate at 25 °C (0.16 d<sup>-1</sup>) (Table 2). However, when acetate was completely

 exhausted, the butyrate was taken up and was exhausted after 9 days at 35 °Cwhereas acetate was not predicted to be completely exhausted after 10 days at 25°C (Figure 2). The growth rate associated with butyrate uptake,  $\mu_b(S_b)(d^{-1})$ , at 25°C, was described by Turon *et al* (2015) as following a modified Haldane equation (Eq 7).

$$
\mu_b(S_b) = \mu_{b\text{-max}} * \frac{K_D}{K_D + S_a} * \frac{Sb}{S_b + \frac{\mu_{b\text{-max}}}{\alpha} * \left(\frac{Sb}{S_{b\text{-opt}}} - 1\right)^2}
$$
 Equation 7

362 where  $S_b$  is the concentration of butyrate  $(g_C, L^{-1})$ ,  $S_{b\_opt}(0.05 g_C, L^{-1})$  is the concentration 363 of butyrate when  $\mu_b$  (S<sub>b</sub>) is maximum and equivalent to  $\mu_{b_{\text{max}}}(0.16 \text{ d}^{-1})$ , the maximum 364 growth rate associated with butyrate assimilation,  $\alpha$  (15.1 L.d.g<sub>C</sub><sup>-1</sup>) is the initial slope 365 and $K_D$  ( 2.10<sup>-10</sup>g<sub>C</sub>.L<sup>-1</sup>) is the half inhibitory constant associated with the diauxic growth.

361

366 The predicted growth rate on butyrate at 25°C varied with the butyrate concentration

367 and reached its maximum,  $\mu_b$ <sub>max</sub>, after 9.5 days of cultivation when the butyrate

368 concentration reached  $S_{b\_opt}(0.05 \text{ g}_C \cdot L^{-1})(\text{Supplementary Material Figure S3}).$  At 35°C,

369 the apparent growth rate was calculated for a butyrate concentration of 0.23  $g_C$ . L<sup>-1</sup> which

370 was reached after 5.7 days of cultivation (Figure 2-B). Consequently, the time to reach

371 butyrate exhaustion was shorter at 35  $\degree$ C than at 25  $\degree$ C despite a higher

372 maximumgrowth rate at 25 °C than the apparent growth rate at 35 °C (Figure 2). The

373 carbon yield on butyrate at 35°C was half that at 25°C. Contrary to the hypothesis

374 suggesting that the butyrate inhibitionmight be reduced at 35°C, butyrate inhibition was

375 stronger at 35°C than at 25°C. Furthermore, no microalgae growth was observed at

376 either 25 °C or 35 °C in the butyrate control (no acetate). As for growth on acetate in

377 mixture, butyrate inhibition at 35 °C depended on the concentration since the butyrate

 uptake rate was faster at 35 °C than 25 °C when butyrate concentration was reduced to 379  $0.2 \text{ g}_\text{C} L^{-1}$  (Supplementary material Figure S4).

#### *3.2.2 Reduced butyrate inhibition at 30 °C*

As shown in Figure 2-A and Table 2, the growth rate and carbon yield on acetatein

<span id="page-18-0"></span>382 mixturewere both higher at 30 °C than at 25 °C or 35 °C. However, there was no

significant difference (p>0.05)between these growth rates and carbon yields and thosein

the acetatecontrol(Table 2, Supplementary material Figure S2). The presence of butyrate

did not appear to inhibit microalgae growth on acetate at 30°C.

Similarly, when butyrate was taken up (in mixture), the apparent growth rate and the

387 microalgae biomass yield were higher at 30 °C (0.16 d<sup>-1</sup> and 0.56  $g_C.g_C$ <sup>-1</sup> respectively)

388 than at 35 °C (0.11 d<sup>-1</sup> and 0.28  $g_C.g_C^{-1}$  respectively) (Table 2). The apparent growth rate

389 at 30°C was calculated for a butyrate concentration of 0.29  $g_C$ . L<sup>-1</sup> which was reached

after 2 days of cultivation (Figure 2 and Table 2).As explained in the previous

391 paragraph (3.2.1), the maximum growth rate at  $25^{\circ}$ C (0.16 d<sup>-1</sup>) could only be reached at

392 a low butyrate concentration (0.05  $g_C L^{-1}$ ). These results suggest that there was less

butyrate inhibition at 30 °C thanat 25 °C. Furthermore, microalgae growth was observed

394 in the butyratecontrol whereas no growth was observed at  $25 \degree C$  or  $35 \degree C$ . A cultivation

temperature of 30 °C thus successfully reduced butyrate inhibition and consequently

396 butyrate exhaustion occurredmore than 3 days earlier than at  $25 \text{ °C}$  (Figure 2-A). At

30°C,enzymatic reactions countering butyrate inhibition mayhave beenencouraged.

Temperatures higher than 25 °C increased heterotrophic growth on both acetate and

butyrate. However, thenear-optimum temperature for acetate was 35 °C while for

butyrate it was 30°C. Cultivation on a mixture of acetate and butyrate ata

 suboptimumtemperature for growth on acetate alonemay have reduced butyrate inhibition.

 **3.3 Combined effects of temperature and light on growth of** *C. sorokiniana* **on a mixture of acetate and butyrate**

 *3.3.1 At 35 °C in the presence of light, microalgae growth on acetate or on butyrate reliedmoreon heterotrophic growth than at 25°C*

A strict autotrophic control (bicarbonate as the sole carbon source) was carried out at

408 35 °C to assess the effect of temperature in autotrophic conditions. In the autotrophic

409 control, the autotrophic production rate of biomass (0.09 g.L<sup>-1</sup>.d<sup>-1</sup>) at 35 °C (Figure 3-A)

410 was similar to that observed at 25 °C (0.11 g.L<sup>-1</sup>.d<sup>-1</sup> - see Figure 1-B). Temperature

appeared to have no significant effect on autotrophic growth.

Under mixotrophic conditions for the acetate control (no butyrate), the growth rate was

- 413 significantly higher( $p$ < 0.05) at 35 °C (5.65 d<sup>-1</sup>)than at 25 °C (4.14 d<sup>-1</sup>) in the presence
- 414 of light but was not significantly different from the growth rate observed at  $35^{\circ}$ C with

415 no light (5.88 d<sup>-1</sup>) (p > 0.05- Tables 1 and 3). About 85% of the biomass content ( $X_{Het}^{MixO}$ ,

Eq 5) at the time of acetate exhaustion was due to acetate uptake(Table 3). These results

suggest that *C. sorokiniana* followed a heterotrophic type of metabolism at 35 °C

- despite the presence of light.
- The combined effects of temperature and light on microalgae growth for the

butyratecontrol (no acetate) was also studied (Figure 3-A). During the first six days, the

- biomass in the butyratecontrol was lower than the biomass in the autotrophic control.
- The presence of butyrate seemed to inhibit autotrophic growth under mixotrophic
- conditions at 35 °C. This inhibition depended on the concentrationsince autotrophic



 62% (Table 3). The time taken to exhaustbutyrate completely was 3 daysless than under 448 heterotrophic conditions at 25  $\degree$ C and 35 $\degree$ C, probably because of the high biomass reached after acetate exhaustion and because of the autotrophic biomass growth at 35°C. Light increased butyrate uptake at 35°C for cultivation on a mixture of acetate and 451 butyrate. At 35 °C, the presence of butyrate reduced the apparent growth rate on acetate under both heterotrophic and mixotrophic conditions and also inhibited autotrophic growth in the butyratecontrol under mixotrophic conditions. Further investigation on the effect of butyrate on the respiration rate and/or photosynthetic activity mayprovide further information on the negative effect of butyrate on mixotrophic and heterotrophic growth observed in this study at high temperature.

### **4 Conclusions**

 The previously accepted optimum cultivation temperature (35°C) did not provide the best conditions forheterotrophic or mixotrophic growth of *C. sorokiniana*on a mixture of acetate and butyrate. The apparent heterotrophic growthrate on acetate was highest at 30 °C(4.1d<sup>-1</sup>). At 25 °C light improved the apparent butyrate uptake (71 mg<sub>C</sub>.L<sup>-1</sup>.d<sup>-</sup> <sup>1</sup>) because simultaneous heterotrophic and autotrophic growth increased the biomass 463 (reaching  $1.14$  g.L<sup>-1</sup>). In conclusion, *C. sorokiniana* may be cultivated successfullyon DF effluents, at a temperature lower than that previously considered to be optimum 465 (30 $^{\circ}$ C) and with exposure to light.

### **5 Acknowledgements**

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# **6 References**





# **7 Figure captions**

# **Figure 1. Effect of DCMU and light on growth of** *C. sorokiniana* **cultivated on**

**butyrate and acetate at 25 °C.** 

571 **(A)** Dry weight of *C. sorokiniana* cultivatedwithout DCMU under autotrophic

- 572 conditions (with 0.3  $g_C L^{-1}$  of NaHCO<sub>3</sub> and under  $123 \pm 10$  µmol photons.m<sup>-2</sup>s<sup>-1</sup>)( $\bullet$ ),
- mixotrophicconditions (with 0.3  $g_C$ , L<sup>-1</sup> of acetate and under 123  $\pm$  10 µmol photons.m<sup>-</sup> 573
- 574  $^2$  s<sup>-1</sup>) ( $\blacklozenge$ ) and heterotrophic conditions (with 0.3 g<sub>C</sub>.L<sup>-1</sup> of acetate in darkness) ( $\blacksquare$ ). Dry
- 575 weight of *C. sorokiniana* cultivatedwith 10µM DCMU under autotrophicconditions
- 576 (with 0.3 g<sub>C</sub>.L<sup>-1</sup> of NaHCO<sub>3</sub> and under  $123 \pm 10$  µmol photons.m<sup>-2</sup>.s<sup>-1</sup>) (<sup>o</sup>),
- mixotrophicconditions (with 0.3  $g_C$ , L<sup>-1</sup> of acetate and under 123  $\pm$  10 µmol photons.m<sup>-</sup> 577
- 578  $\left(2\right)$   $\left(3\right)$  and heterotrophic conditions (with 0.3 g<sub>C</sub>.L<sup>-1</sup> of acetate in darkness) (**0). (B)**
- 579 **and (C)***C. sorokiniana* cultivated under mixotrophic conditions at 25 °C. Dry weight (
- 580 ), butyrate concentration  $(\bullet)$  and acetate concentration  $(\blacksquare)$  during cultivation **(B**) on a
- 581 mixture of butyrate and acetate,  $0.3 \text{ g}_\text{C} L^{-1}$  of each and **(C)** on 0. 3  $\text{g}_\text{C} L^{-1}$  of butyrate as
- 582 single substrate (butyratecontrol). The dry weight for autotrophic cultivation  $\odot$ ) and the
- 583 predicted values for heterotrophic cultivation at  $25 \text{ °C}$  dry weight (green dashed lines),
- 584 acetate concentration(red dashed lines) and butyrate concentration(blue dashed lines)-
- 585 are shown for comparison.
- 586 **Figure 2. Effect of increasing temperature, from 25 °C to 35 °C, on heterotrophic**
- 587 **growth of** *Chlorella sorokiniana* **cultivated on a mixture of acetate and butyrate.**
- 588 **(A)**Dry weight of *C. sorokiniana* cultivated under heterotrophic conditions on a mixture
- 589 of acetate and butyrate at 30 °C ( $\bullet$ ) and 35 °C ( $\bullet$ ). **(B)**Acetate and butyrate
- 590 concentrations for cultivation at 30 °C ( $\Box$ and  $\Diamond$ ) and 35 °C ( $\Box$ and  $\Diamond$ ). The
- 591 predictedvalues for heterotrophic cultivation at  $25 \degree C$  dry weight (green dashed lines),
- 592 acetate concentration(red dashed lines) and butyrate concentration(blue dashed lines) at
- 593 25 °C are shown for comparison.
- 594 **Figure 3. Effect of butyrate on growth of** *C. sorokiniana* **cultivated on acetate and**  595 **butyrate at 35 °C under mixotrophic conditions.**
- 596 Dry weight of *C. sorokiniana*  $\left( \bullet \right)$ , butyrate concentration  $\left( \bullet \right)$  and acetate concentration
- 597 **(D)** during cultivation on  $(A)0.3$  g<sub>C</sub>.L<sup>-1</sup> of butyrate (butyratecontrol) and **(B)**on a
- 598 mixture of 0.3  $g_C$ . L<sup>-1</sup>butyrate and 0.3  $g_C$ . L<sup>-1</sup> acetate. The dry weight for autotrophic
- 599 cultivation ( $\bullet$ ) and the predicted values for heterotrophic cultivation at 25 °C dry
- 600 weight (green dashed lines), acetate concentration (red dashed lines) and butyrate
- 601 concentration (blue dashed lines) are shown for comparison.

Figure 1 **Figure**









## **Supplementary Material**

**Figure S1. Effect of light on** *C. sorokiniana***'s growth on acetate (0.3**  $g_C$ **,L<sup>-1</sup>) at 25**  $\text{°C.Microalgae concentration (g.L<sup>-1</sup>) (•})$  and acetate concentration ( $\blacksquare$ ) are presented. Microalgae concentration (g.L<sup>-1</sup>) ( $\bullet$ ) during autotrophic growth is presented. The simulated heterotrophic microalgae concentration (green dashed lines) and acetate concentration (red dashed lines) (blue dashed lines) at 25 °C are represented.

## **Figure S2. Effect of temperature on microalgae heterotrophic growth on acetate**

 $(0.3 \text{ g}_C \cdot \text{L}^{-1})$ . Microalgae concentration, in g. L<sup>-1</sup>, during heterotrophic growth on acetate at 30 °C ( $\bullet$ ) and 35 °C ( $\bullet$ ) are represented in subfigure A. Acetate concentrations, in  $g_C$ . L<sup>-1</sup>, during growth at 30 °C ( $\blacksquare$ ) and 35 °C ( $\blacksquare$ ) are represented in subfigure B. The simulated heterotrophic microalgae concentration (green dashed lines) and acetate concentration (red dashed lines) (blue dashed lines) at 25 °C are represented.

**Figure S3. Variation of the growth rate on butyrate**  $(\mu b(Sb))$  **according to the simulations of the model representing heterotrophic growth at 25°C.**

**Figure S4. Heterotrophic growth of** *Chlorella sorokiniana* **on mixtures of acetate and butyrate (0.2 gC.L-1 each) at 25 °C, 30 °C and 35 °C.**Microalgae concentration, in g.L<sup>-1</sup>, during heterotrophic growth on mixtures of acetate and butyrate at 30 °C ( $\bullet$ ) and 35 °C ( $\bullet$ ) are represented in subfigure A. Acetate and butyrate removals, in  $g_C L^{-1}$ , during growth at 30 °C ( $\blacksquare$ and  $\blacklozenge$ ) and 35 °C ( $\blacksquare$  and  $\blacklozenge$ ) are represented in subfigure B. The simulated heterotrophic microalgae concentration (green dashed lines), acetate concentration (red dashed lines) and butyrate concentration (blue dashed lines) at 25 °C are represented.

**Figure S5. Comparison of autotrophic and mixotrophic growth of C. sorokiniana on 0.3**  $g_c$ **.L**<sup>-1</sup> of acetate (A) and 0.2  $g_c$ **.L**<sup>-1</sup> of butyrate (B) at 35 °C.Microalgae concentration (g.L<sup>-1</sup>) ( $\bullet$ ), butyrate uptake ( $\bullet$ ) and acetate uptake ( $\bullet$ ) are presented. Microalgae concentration (g.L<sup>-1</sup>) ( $\bullet$ ) during autotrophic growth is presented. The simulated heterotrophic microalgae concentration (green dashed lines), acetate concentration (red dashed lines) and butyrate concentration (blue dashed lines) at 25 °C are represented.

# **Figure S 1 Figure**







2

**B**









Figure S5.



## **Table 1**

**Effect of light on growth and productionrates (µappandrlin)and yields of** *C. sorokiniana***for cultivation at 25 °C on acetate (A), butyrate (B) and a mixture of butyrate and acetate (A + B).**Mean values and standard deviations calculated from triplicates are given.



a: The heterotrophic carbon yield ( $Y_{Het}^{Mixo}$  ) was calculated by subtracting the carbon yield associated with autotrophic growth from the mixotrophic carbon yield ( $Y_{\text{MixO}}^{\text{MixO}}$  ).

b: The fraction of mixotrophic biomass due to autotrophic growth on CO<sub>2</sub> (X $_{\rm Auto}^{\rm Mixo}$  ) was calculated as follows:

$$
X^{Mixo}_{Auto} = \frac{Y^{Mixo}_{Mixo} - Y^{Mixo}_{Het}}{Y^{Mixo}_{Mixo}} * 100
$$

## **Table 2**

**Effect of temperature on apparent growth rate (µapp) and heterotrophic carbon yield of**  *Chlorella sorokiniana* **under heterotrophic conditions on acetate (A), butyrate (B) and a mixture of butyrate and acetate**  $(B + A)$ **.** The figures at 25 °C are taken from a previous study for heterotrophic growth of *C. sorokiniana*. For 30 °C and 35 °C, the mean values and standard deviations calculated from triplicates are given. Valueswith different letters are statistically different ( $p \le 0.05$ , one-way ANOVA and Tukey's post-hoc analysis). The carbon yield was estimated for a microalgae cell composition of 50% of carbon [12].

