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Mikhael Bechelany

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Mechanical properties of boron nitride thin films prepared by atomic layer deposition

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Matthieu Weber^a, Emerson Coy^b, Igor Iatsunskyi^{b*}, Luis Yate^c, Philippe Miele^a and Mikhael Bechelany^{a*}

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Abstract. Due to their wide bandgap, boron nitride (BN) thin films are the focus of interest for their potential applications in microelectronic devices. The reliability of these devices is essential and is directly linked to the mechanical properties of the films used for their fabrication. Herein, an atomic layer deposition (ALD) process based on sequential pulses of BBr₃ and NH₃ at 750°C is used in order to prepare BN thin films. We report the main physicochemical properties of the films using various analytical methods, and we perform nanoindentation experiments in order to determine the elastic modulus and the hardness. Next, we annealed the films at 1000 and 1350°C in order to gain understanding on the relation between the annealing temperature, the microstructure obtained and the resulting mechanical properties. Although the hardness of the films presented similar values of 5±1 GPa for all temperatures, it has been found that the elastic modulus increases up to 150±9 GPa when applying an annealing treatment of 1350°C, which represents a 37% improvement compared to the initial film prepared at 750°C.

Boron nitride is a III-V material exhibiting exceptional physical and chemical properties that can be applied in various fields, from photonics to high temperature furnaces.¹⁻³ In the last decade, nanometer-thin films of boron nitride became the focus of interest for their potential application in semiconductor devices. For example, they could be used as dielectric layers for graphene based electronics.^{4,5} The thin films used in the fabrication of microelectronic integrated circuits must perform their electronic and magnetic functions, but they must present certain chemical and mechanical properties as well. The reliability of microelectronic devices is essential, they are expected to present structural integrity over their lifetime, and mechanical failures shall not occur. Thus, these materials, though not

selected exclusively for their mechanical properties, must provide adequate resistance to the mechanical forces that arise in these applications.⁶ Elastic modulus (E_r) is a material property related to atomic bonding, which characterizes its stiffness. It represents one of the most used material constants in microelectronic devices. Hardness (H) depicts the ability of a material to resist deformation. As it is related to the elasto-plastic response of a solid material under external loads, it represents an excellent indicator for material resistance to wear, scratch and deformation failure.⁷

Boron nitride films can be prepared using many manufacturing techniques, such as mechanical exfoliation,⁸ chemical vapor deposition,⁹⁻¹¹ ion-beam-enhanced deposition,¹² reactive ion plating,¹³ or magnetron sputtering.¹⁴ However, these technologies often present limitations in terms of conformality of the depositions, and they may require the use of energetic ions that can damage the substrates.

Atomic layer deposition (ALD) is a thin film deposition technique based on chemical surface reactions allowing for the preparation of high quality materials, with high uniformity, precise growth control, and excellent conformality.¹⁵⁻¹⁷ This route is currently used for the most challenging thin-film deposition applications such as microelectronics, as it is used for the preparation of high-k oxides gates.^{16, 18} Concerning the characterization of thin films prepared by ALD in terms of mechanical properties, some recent efforts have carried out on oxide materials. Ylivaara *et al.* reported elastic modulus of 170 GPa,¹⁹ and Tripp *et al.* measured similar elastic modulus of 168–182 GPa and Berkovitch hardness of 12 GPa, both teams focusing on 300 nm thick Al₂O₃ films.²⁰ Tapily *et al.* reported a hardness of 9.5±2 GPa and a modulus of 220±40 GPa for ALD HfO₂ films.²¹ Yuan *et al.* investigated the hardness of ~500 nm thick ZnO films, which was found to be comprised between 5 and 8 GPa, depending on the crystallinity.²² Jian and Lee measured H and E_r for ZnO thin films which thickness was about 200 nm, and the values found were respectively 10.3 GPa and 168.6 GPa.²³

Recently, several studies reported the synthesis of boron nitride thin films by ALD.²⁴⁻²⁸ Although the main physico-chemical properties, the electrical performances and the low leakage current density of BN ALD films have been reported,²⁶ these studies did not include the investigation of the mechanical properties of the films.

^a Institut Européen des Membranes, IEM-UMR 5635 ENSCM, UM, CNRS, Université de Montpellier, Place Eugène Bataillon, 34095 Montpellier Cedex 5, France

^b NanoBioMedical Centre, Adam Mickiewicz University in Poznan, 61-614, Umultowska str. 85 Poznan, Poland

^c CIC biomaGUNE, Paseo Miramón 182, 20014, San Sebastián, Spain

^d *corresponding authors: Mikhael Bechelany: mikhael.bechelany@umontpellier.fr, phone: +33467149167, Fax: +33467149119 and Igor Iatsunskyi: yatsunskiy@gmail.com

§ Supporting Information: Experimental section, TEM image of an annealed sample
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Herein, we used an ALD process in order to prepare BN thin films and study their mechanical properties. We report the main physico-chemical properties of the films using various analytical methods, and we perform indentation testing at the nanometer scale in order to determine the elastic modulus and the hardness. Furthermore, we annealed the films at 1000 and 1350°C in order to gain understanding on the relation between the annealing temperature, the microstructure obtained and the resulting mechanical properties.

An ALD process based on boron tribromide (BBr_3) and ammoniac (NH_3) at 750°C has been developed in order to synthesize BN thin films (see Supporting information). The sequential exposures of BBr_3 and NH_3 were separated by argon purges. Briefly, the ALD cycle consisted of 0.1 s pulse of BBr_3 followed by 5s exposure, 10 s purge, 5s pulse of NH_3 and 10 s purge to finish the cycle. The linearity of the growth depicted in Figure 1 demonstrates the successful ALD process.

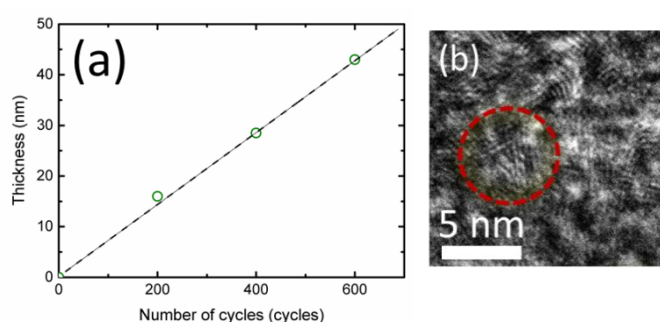


Figure 1. (a) Thickness of the BN film as a function of the number of ALD cycles, as measured by spectroscopic ellipsometry. The line serves as guide to the eyes. (b) High resolution TEM image of a ~ 25 nm thick BN film prepared by ALD.

As measured by spectroscopic ellipsometry (SE), the saturated growth rate of the ALD process was 0.08 nm/cycle, which is in the same order of magnitude that the ones measured by other authors.^{24, 26, 29} Our group has previously shown that this process based on self-limiting chemical reactions results in conformal and high quality BN films. More details about this ALD process can be found elsewhere.³⁰ Several analytical methods have been used to characterize the physico-chemical properties of the films (~ 25 nm thick) prepared at 750°C. X-Ray photoelectron spectroscopy (XPS) was used to determine the chemical composition. The films obtained showed near stoichiometric (1.1:0.9) contents of B and N with a clear BN structure, evidenced by the binding energy of B 1s (190.3eV) and N 1s (398.0eV). The atomic concentrations found were $4\pm 3\%$ and $7\pm 3\%$, for C and O, respectively, and no Br was found in the films (at the detection limit). X-Ray Diffraction (XRD) and Transmission Electron Microscopy (TEM) measurements have shown that the films present a turbostratic microstructure, and the presence of small areas of $\sim 5 - 7$ nm of BN planes of hexagons have been observed (see Figure 1b). A mass density value of 2.2 ± 0.3 g/cm³ has been

extracted from X-Ray reflectometry (XRR) measurements. Atomic force microscopy (AFM) has been performed in order to determine the RMS roughness of the deposited films, and a value of 0.9 ± 0.3 nm has been measured. The growth-per-cycle, C and O contents, mass density, and roughness values are given in Table S1 (Supporting Information).

Next, in order to study the mechanical properties of the films, we performed nanoindentation measurements. The nanoindentation method enables the extraction of two main pieces of information: the universal hardness, H , and the reduced modulus of elasticity, E_r , which are determined from the nanoindentation load–displacement curves by the Oliver–Pharr method (see Experimental section). The method involves making a small indentation in the film (usually with a Berkovich indenter presenting a triangular pyramid geometry), while continuously recording the indentation load and the displacement during one complete cycle of loading and unloading.³¹ Figure 2 depicts a typical nanoindentation loading curve for a ~ 80 nm thick BN film prepared at 750°C by ALD.

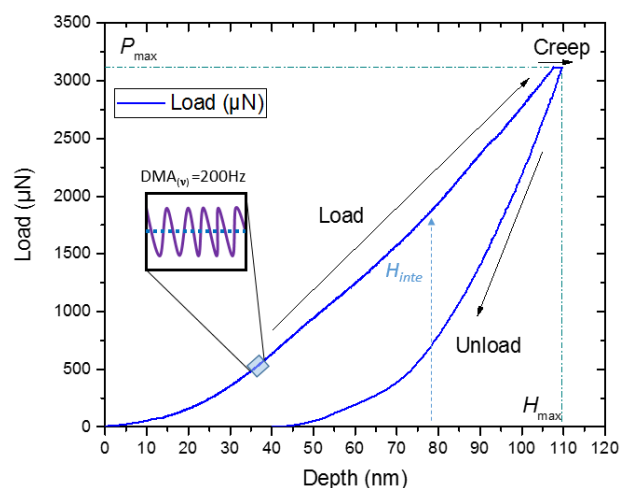


Figure 2. Nanoindentation loading curve for a BN film prepared at 750°C by ALD. The film was ~ 80 nm thick and the substrate used was a Si(100) wafer.

The modulus of elasticity E_r and hardness H obtained were 109 ± 10 GPa and 5.3 ± 0.6 GPa, respectively. Although no results have been reported on the mechanical properties of turbostratic BN thin films to our knowledge, our results can be compared to the ones reported for hexagonal BN thin films.

Previous works have reported that hexagonal BN monolayer has a bulk modulus around 160 GPa,³²⁻³⁴ which is in the same order of magnitude than the value we measured for turbostratic BN in our study. Concerning crystalline BN films, E_r values between 420 and 505 GPa have been obtained on nanocrystalline BN films (400 nm thick) prepared by pulsed laser deposition.³⁵ Nanoindentation measurements yielded hardness values for cubic BN films up to 60–70 GPa (700 nm thick).³⁶ Jiang *et al.* studied the mechanical properties of cubic BN films (2 μm thick) prepared by CVD using the same Oliver–Pharr method.³⁷ Hardness values from 25 to 53 GPa and

effective Young modulus of 600 ± 100 GPa have been measured.

These values are substantially higher than the ones measured for our ALD films, but they were performed on crystalline BN film. It is indeed logical that the cubic phase present better mechanical properties than the turbostratic phase. In addition, one has to precise that the E value approaches the property of the substrate when the penetration depth increases, and that the 600 GPa value measured by Jiang *et al.* was obtained by extending the curve averaged from the experimental data to zero penetration depth.³⁷

The results we obtained for BN ALD films are in the same order of magnitude as the mechanical properties measured for ZnO, Al₂O₃ and HfO₂ films prepared by ALD.²⁰⁻²² These data provide input for the design of semiconductor devices such as microelectronic chips and micro electro mechanical systems (MEMS) with ALD of BN, and can be of use for other applications as well.

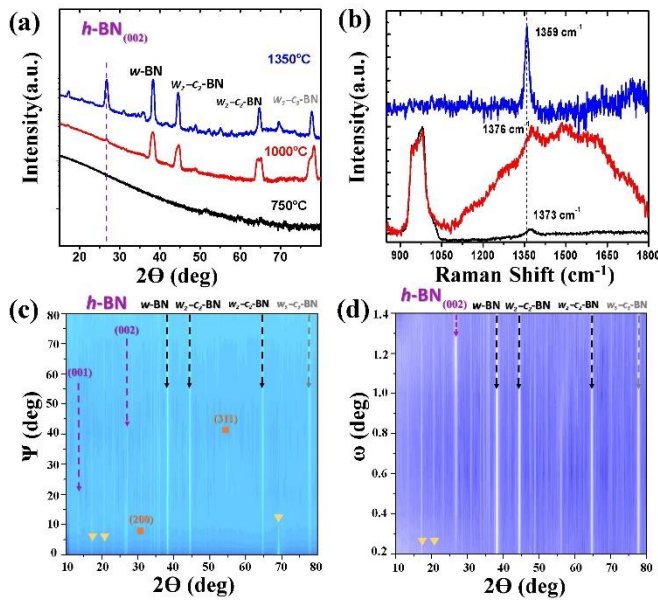


Figure 3. (a) XRD patterns of BN films prepared at 750°C, as-prepared and annealed at 1000 and 1350°C. The peaks originated from h-BN (002) are shown with the purple line. (b) Raman spectroscopy of BN films prepared at 750°C, as-prepared and annealed at 1000 and 1350°C. The BN films were 80 nm thick and the substrates were silicon with a native oxide layer. (c) 2θ scans at different Ψ angle (0-80 degrees). (d) 2θ scans performed at different penetrations of the incident angle (ω). The \blacksquare and \blacktriangledown symbols represent the Zinc blende and spurious phases, respectively.

It is known in the literature that the mechanical properties are generally linked to the microstructure of the films,³⁸ and that this latter can be modified by high temperature treatments. In

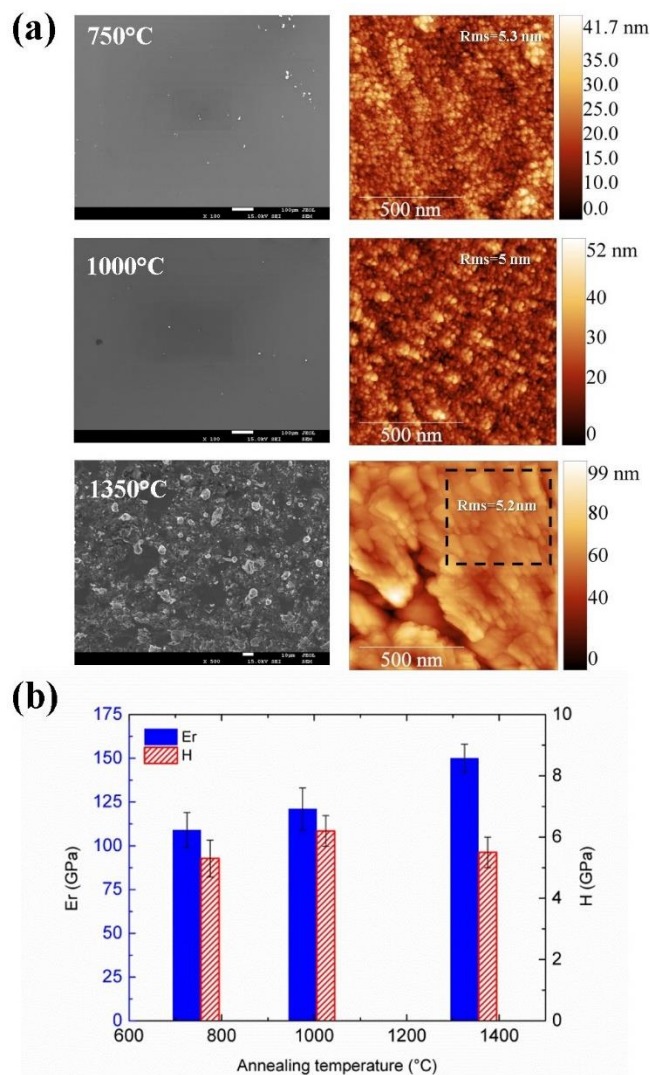
fact, several studies have shown the crystallographic changes of boron nitride films. Starting from imperfect turbostratic microstructures, the heat treatments typically promoted the ordering of BN structures leading to crystalline hexagonal BN.^{11, 39, 40} Thus, in order to gain understanding on the relation between the annealing temperature and the resulting mechanical properties, we annealed the films at 1000°C and 1350°C (for 3 hours). Aiming to reveal the phase transformations after annealing, we performed XRD and Raman spectroscopy measurements. First, XRD measurements have been carried out for the sample prepared at 750°C and for the annealed samples at 1000 and 1350°C, in order to confirm the crystallographic changes. The as prepared sample showed no features, as depicted in Figure 3a. This indicates that the films present nanocrystal size smaller than the XRD detection limit or even amorphous phase. In fact, as can be seen in Figure 1b, ALD of BN typically results in films presenting a turbostratic microstructure with small nanocrystals.^{24, 30, 41, 42} However, the high temperature annealing treatments resulted in XRD patterns showing different slidings in the structures of BN, and peaks characteristic from hexagonal BN (h-BN), wurtzite (w-BN), w₂-c₂-BN and w₈c₃-BN were found for the samples annealed at 1000 and 1350°C, confirming the phase transformation.⁴³ These new visible microstructures are attributed to the evolution of the original turbostratic microstructure after the high temperature treatments, that allowed the sliding and buckling of BN sheets.⁴³⁻⁴⁵

Next, we carried out Raman spectroscopy measurements in order to provide information about the vibrational modes in the BN systems obtained. Figure 3b shows the Raman spectroscopy of BN samples prepared at 750°C, and annealed at 1000 and 1350°C. It is known that the E_{2g} vibrational mode for bulk h-BN appears at approximately 1366 cm⁻¹.⁴⁶ The Raman spectrum of the as prepared sample at 750°C shows a weak peak of BN at 1373 cm⁻¹, indicating that the prepared BN layer was amorphous with a small ratio of hexagonal crystalline phase. For the annealed sample at 1000°C, the intensity of the BN peak increased, indicating the amorphous-to-crystalline phase transition. A broad Raman peak between 900 and 1100 cm⁻¹ corresponds to the superposition of three transverse optical phonons (~2TO (X), 2TO (W) and 2TO (L)).⁴⁷ However, a broad peak around 1400 cm⁻¹ consisting of at least 4 components (1376, 1500, 1620 and 1720 cm⁻¹) also appears. This might be explained by the presence of various BN phases in the material. For the sample annealed at 1350°C, only one main peak at 1359 cm⁻¹ is visible, which corresponds to the h-BN phase, suggesting that the prepared layer is fully crystalline.

Prior to nanomechanical analysis, it is important to understand and investigate the high temperature mixture of phases observed in the 1350°C sample. Therefore, we performed long XRD diffraction experiments in non-geometrical configurations. These experiments enable us to investigate the possible stress induced in the films and the ordering of the phases. From Figure 3(c) where the Ψ angle is analyzed, it is clear that the *h*-BN phase, although rapidly disappearing with the increment of Ψ , is still visible at $40^\circ\Psi$, showing that even though the turbostratic configuration should promote randomly oriented *h*-BN crystals, this phase is somehow aligned perpendicularly to the substrate interface. Additionally, as expected by the sliding and buckling of the BN, the w2-c2-BN and w8c3-BN phases are randomly distributed, as demonstrated by their strong presence even at $90^\circ\Psi$. In addition, the presence of a Zinc blend BN phase is found. Interestingly, all phases are stress free, as evidenced by the constant *d* spacing and 2θ angle. These results suggest that the phase composition is not homogenous in the high temperature sample. Therefore, in order to investigate different penetration depths, Grazing-Incidence XRD experiments were performed at different ω angle. Figure 3(d) shows high intensity peaks for the w2-c2-BN and w8c3-BN at angles as low as $0.2^\circ\omega$. On the other hand, the *h*-BN phase intensity increases regularly with the ω angle, suggesting that the highest concentration of this phase is present in the deepest part of the film, whereas the secondary phases are present everywhere in the sample.

From these analyses, it is clear that the sample annealed at 1350°C has a very complex phase diagram, and its morphology should reflect such changes. SEM and AFM analysis have been performed to confirm this assumption. In Figure 4(a), both SEM and AFM micrographs are shown for the whole set of samples. The roughness of the samples remains close to 5 nm for all the samples, whereas a large change in morphology is observed in the 1350°C sample. Large areas of the film have reconstructed in complex microstructures. Even if no stoichiometric changes were observed from XPS studies, the clear differences in contrast seen in the SEM images suggest several changes in electronic structures, which points to different phases of BN.

Next, in order to depict the relation between the mechanical properties and the microstructure, we performed the nanoindentation experiments on the annealed films. The values obtained after the annealing treatment at 1000°C were 121 ± 12 GPa and 6.2 ± 0.5 GPa for the modulus of elasticity *E_r* and hardness *H* obtained, respectively. Considering the sample on which a heat treatment at 1350°C has been performed, values of 150 ± 8 GPa (for *E_r*) and 5.5 ± 0.5 GPa (for *H*) have been measured. The elastic modulus and hardness obtained for BN films prepared at 750°C, with an annealing at 1000°C and 1350°C are summarized in the Figure 4.



As can be depicted from these results, whereas the hardness is not substantially affected, a strong increase of the modulus of elasticity is obtained when applying the temperature treatments, which can be attributed to the crystalline phase transition of the BN films. The increase in elastic modulus and hardness in the films after annealing might be correlated with the microstructural evolution we observed, but also to possible shifts in the phase composition and grain size as a consequence of the annealing process.

Figure 4. (a) SEM and AFM surface micrographs of the samples prepared at 750, 1000 and 1350°C. The insets show the roughness of the samples (dashed area in 1350°C shows the area where the RMS was taken). (b) Elastic modulus (*E_r*) and hardness (*H*) obtained for BN films (80 nm thick) prepared by ALD at 750°C without heat treatment, and with heat treatments (annealing) at 1000°C and 1350°C. The substrates were Si(100) wafers.

In particular, the elastic modulus increases up to 150 ± 9 GPa when applying an annealing treatment of 1350°C, which is a

significant 37% enhancement compared to the initial film prepared at 750°C. The mechanical properties obtained are therefore very promising and can be used as a useful comparison point for future studies and for the design of micro devices.

Conclusions

In this work, we reported the mechanical properties of BN thin films prepared by ALD. The process was based on BBr₃ and NH₃ at 750°C. The main physicochemical properties of the films have been determined by various analytical methods, and we performed nanoindentation loading experiments in order to determine the elastic modulus and the hardness of the films. Films grown at 750°C presented an elastic modulus of 109±10 GPa and a Berkovich hardness of 5.3±0.6 GPa. These results are in line with other E and H measured for other ALD materials reported in the literature. Furthermore, in order to gain understanding on the relation between heat treatments and the resulting mechanical properties, we annealed the films at 1000°C and 1350°C and performed again the nanoindentation experiments. It has been found that the elastic modulus increased up to 150±9 GPa when applying a heat treatment of 1350°C, denoting a considerable enhancement of the mechanical properties compared to the initial film prepared at 750°C. XRD and Raman spectroscopy measurements were used to confirm the phase transition. The results presented in this work can be used as a useful comparison point for future studies on the mechanical properties of ALD films, and open prospects for the use of BN ALD films in electronic devices and other applications where mechanical properties are crucial.

Notes and references

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